FIGURE CAPTIONS

- Fig.1-1. Structural view of a C₆₀ molecule drawn by Burgi et al [2].
- Fig.1-2. Structure view of a FCC unit cell of a C_{60} single crystal drawn by Burgi *et al* [2].
- Fig.1-3. A schematic view of tetrahedral and octahedral regular interstitial sites in a FCC unit cell of a C₆₀ single crystal. Solid circles represent C₆₀ molecules.
- Fig.1-4. (a) A schematic energy level diagram for the C₆₀ molecule. S and T indicate the singlet and triplet states, respectively. M and D indicate molecule and dimer states, respectively. (b) The bimolecular chemical reaction model of the [2+2] cycloaddition mechanism [19].
- Fig.1-5. A NaCl structure of an A₁C₆₀ compound. Intermolecular bonds are created in the temperature lower than *ca.* 400K in [110] directions. Large and small balls represent C₆₀ molecules and alkali ions, respectively.
- Fig.1-6. Structural model of C₆₀ polymers fabricated by high-pressure and high-temperature treatments.
- Fig.1-7. Structural model of three-dimensional C_{60} polymer reported by Okada *et al.* [40]. Red spheres denote the four-coordinated sp^3 carbon atoms.
- Fig. 2-1. A schematic picture of electric furnace and a quartz tube. T_1 and T_2 denote the temperatures at the cold and hot ends of the quartz tube, respectively.
- Fig. 2-2. (a) A schematic illustration for weighing alkali metals. (b) A schematic illustration for the *in situ* Raman measurements during potassium doping process to C₆₀ thin film.
- Fig. 2-3. Optical arrangement for optical transmittance measurements.
- Fig. 2-4. Optical arrangement for photoluminescence measurements.
- Fig. 2-5. Optical arrangement for Raman scattering measurements.
- Fig. 2-6. A set of diamond anvil cell of the Mao-Bell type used in the present study.
- Fig. 2-7. An illustration of pressure generation with a diamond anvil cell.
- Fig. 2-8. The pressure dependence of ruby florescence lines when paraffin oil is used for the pressure transmitting medium. The FWHMs of the R1 line are plotted as a function of pressure in the upper part of the figure.
- Fig. 3-1. X-ray diffraction (XRD) patterns of pure-C₆₀ and I_xC₆₀ compounds. XRD

- patterns show that the structure of the I_xC_{60} varies from FCC to hexagonal lattice.
- Fig. 3-2. A structural model of an I₄C₆₀ compound with hexagonal lattice [30]. The small closed circles denote iodine atoms. Large open circles denote C₆₀ molecules.
- Fig. 3-3. (a) Raman scattering spectra of the C₆₀ single crystal, fcc-I_xC₆₀, and hex-I_xC₆₀ in the frequency region from 120 to 800 cm⁻¹ at 80 K and room temperature. (b) Raman scattering spectra of the C₆₀ single crystal, fcc-I_xC₆₀, and hex-I_xC₆₀ in the frequency region from 1380 to 1600 cm⁻¹ at room temperature.
- Fig. 3-4. (a) Raman scattering spectra of the hex-I_xC₆₀ in the temperature range from 4 K to 250 K. (b) The result of the peak fitting to data using some Gaussian components. A broken line denotes the Raman spectrum which was obtained at 18 K. Asterisks indicate the plasma lines of an Ar⁺ laser. (c) Temperature dependence of the frequencies of Raman bands obtained by the peak fittings in the same way as illustrated in (b).
- Fig. 3-5. Structural models of alkali metal doped C₆₀ compounds. Large and small balls denote C₆₀ molecules and alkali ions, respectively. A FCC C₆₀ single crystal is drawn in an equivalent body centered tetragonal representation (bct) in order to describe structural changes to body centered cubic (bcc) structure of A₆C₆₀ compound. Body centered cubic representation of a FCC unit cell has been shown in Fig. 1-5.
- Fig. 3-6. In situ Raman scattering spectra of A_xC₆₀ film during potassium doping at 140 °C.
- Fig. 3-7. A schematic picture of the A_g(2) intramolecular vibrational mode and changes in the Raman spectrum of a C₆₀ single crystal caused by photo-polymerization. Dotted and solid lines represent the Raman data and Gaussian components fitted to data.
- Fig. 3-8. Changes in photoluminescence spectrum of a C₆₀ single crystal caused by photo-polymerization. (a) and (b) correspond to the PL spectra of the pristine and photo-polymerized C₆₀, respectively.
- Fig. 3-9. Photoluminescence spectra of PIHP materials. All PL spectra were recorded

- at ambient pressure. Dots denote PL data. Solid and dotted lines denote Gaussian components fitted to data.
- Fig. 3-10. Applied pressure dependence of PL peak energy and band gap energy of C₆₀ samples. and denote the band gap energy of the PIHP materials at ambient pressure obtained from PL (Fig. 3-9) and OT data (Fig. 3-11), respectively. A broken line denotes the energy shifts of the C₆₀ single crystals under pressure obtained by *in situ* PL measurements (Fig. 3-12). □ indicates the PL peak energy of C₆₀ samples obtained after the pressure treatments without photo-irradiation (Fig. 3-13).
- Fig. 3-11. Optical transmittance spectra of PIHP materials: (a) original, (b) 7.7 GPa, (c) 10.6 GPa, (d) 18.9 GPa, (e) 30.0 GPa. All spectra were recorded at ambient pressure. The straight lines attached to spectrum (a) illustrate an estimation of the energy gap, E_g .
- Fig. 3-12. PL spectra of C₆₀ single crystal under high pressure up to 5 GPa. The left and right hand side graphs correspond to pressure increase and decrease runs, respectively.
- Fig. 3-13. PL spectra of C₆₀ single crystal at ambient pressure after subjecting to the pressure treatment up to 26 GPa without photo-irradiation.
- Fig. 3-14. Raman scattering spectra of PIHP materials at ambient pressure.
- Fig. 3-15. Results of peak fitting with Gaussian components to data which are presented in Fig. 3-14. Dots and solid lines denote Raman data and the Gaussian components, respectively.
- Fig. 3-16. Peak frequencies and FWHM (full width at half maximum) of the A_g(2) mode and the p-A_g mode of the PIHP materials at ambient pressure as functions of the applied pressure at PIHP treatments. The A_g(2) and p-A_g modes are denoted by circles and squares, respectively.
- Fig. 3-17. Photoluminescence spectra of HPHT polymers in the temperature range from 15 to 300 K [72].
- Fig. 3-18. Raman spectra of the pristine C₆₀ (a), C₆₀ dimer (b), 1D orthorhombic (c) and (d), 2D tetragonal (e), and 2D rhombohedral (f) structures [33]. Spectra (a)~(c) were excited with the 1064 nm line, and spectra (d)~(f) were excited with the 568.2 nm line.

- Fig. 3-19. Pressure dependence of the frequency of the p-A_g mode (•) and the energy of the PL band (■) of the C₆₀ photopolymer, and the applied pressure dependence of the frequency of the p-A_g mode (o) and the energy of the PL band (□) of PIHP materials. The solid and broken lines are the least squares fits to the data.
- Fig. 3-20. The fictive pressures which were obtained from Raman and PL data. Circles and squares denote the P_f^R and P_f^{PL} , respectively.
- Fig. 3-21. *In situ* high pressure PL spectra of the C₆₀ photopolymer. The PL peak energies which are obtained in the same way as described in Fig. 3-9 are plotted as a function of pressure in the right figure.
- Fig. 3-22. In situ high pressure Raman spectra of the C_{60} photopolymer. Thick and thin arrows indicate the p- A_g and $A_g(2)$ mode, respectively. The frequency of the p- A_g mode is plotted as a function of pressure in the right figure.
- Fig. 3-23. Schematic pictures of the two factors in the formation of intermolecular bonds.
- Fig. 3-24. The comparison of the energy gap shift (ΔE_g) between a C₆₀ crystal and 2D HPHT polymers. Open and solid circles denote the ΔE_g based on the absorption measurement of a C₆₀ crystal under pressure [73] and the theoretically obtained ΔE_g of 2D polymers [75], respectively.
- Fig. 3-25. Conversion of the fictive pressures (left axis) to the Raman frequencies (right axis) in accordance with the pressure coefficient of the p-A_g mode of the C₆₀ photopolymer, 6.3 cm⁻¹/GPa. Squares and circles correspond to the PL and Raman results, respectively.
- Fig. 3-26. A pressure-temperature phase diagram for HPHT C₆₀ polymers [78].
- Fig. 3-27. (a) PL and (b) Raman spectra at ambient condition of C₆₀ single crystals which were subjected to HPHT treatments.
- Fig. 3-28. Temperature and photo-irradiation time dependence of the PL spectra of C₆₀ samples. (A), (B), and (C) correspond to the results of the PIHP treatments at room temperature, 75 °C, and 100 °C, respectively. All PL spectra were recorded at ambient condition after the photo-irradiation at HPHT condition.
- Fig. 3-29. Illustration of the PL analysis for the PIHP treatments at room temperature using two PL elements, I_{Bef} and I_{Aft}. (a) PL spectra of the pristine C₆₀ single

- crystal (I_{Bef}) and (b) PL spectra of C_{60} sample which was subjected to photo-irradiation under HPHT condition for 90 minutes (I_{Aft}). In the left figure, dotted and chain lines denote I_{Bef} and I_{Aft} components, respectively.
- Fig. 3-30. PL analyses for the PIHP treatment at 75 °C, and 100 °C. These analyses were performed in the same way described in Fig. 3-29.
- Fig. 3-31. The ratios, I_{Aft}/I_{Bef.}, as functions of the photo-irradiation time at room temperature (Δ), 75 °C (•), and 100 °C (■). The dashed lines indicate fits to the data using exponential functions to obtain time constants τ. Arrhenius plots are shown in the inset.
- Fig. 3-32. In situ high pressure Raman scattering spectra of the hex-I_xC₆₀ up to 11 GPa. Raman spectra relevant to iodine molecules and C₆₀ molecules are shown by dots in (a) and (b), respectively. The up and down arrows in parentheses denote the pressure increase and decrease process, respectively. Solid lines denote Gaussian components fitted to data.
- Fig. 3-33. Pressure dependence of the peak frequency of the peak-A and A_g(2) mode. Open and closed notations indicate pressure increase and decrease runs, respectively.
- Fig. 3-34. The photo-irradiation time dependence of Raman spectra of the (a) C₆₀ single crystal, (b) fcc-I_xC₆₀ and (c) hex-I_xC₆₀ at ambient pressure. Solid lines denote Lorentz functions fitted to Raman data represented by dots.
- Fig. 3-35. Intensity ratios, I_{p-Ag}/(I_{p-Ag}+I_{Ag(2)}), as functions of photo-irradiation time at ambient pressure. The ratios obtained from C₆₀ single crystal, fcc-I_xC₆₀ and hex-I_xC₆₀ are denoted by circles, squares and triangles, respectively.
- Fig. 3-36. Photo-irradiation time dependence of Raman spectra of the fcc-I_xC₆₀ at ambient pressure.
- Fig. 3-37. Graphitic particle size, La, as a function of the photo-irradiation time. The diameter, La, was estimated using equation 3.1.
- Fig. 3-38. Raman spectra of the fcc- I_xC_{60} which were obtained after the PIHP treatments up to ~37 GPa. The 488.0 nm (a) and 514.5 nm (b) lines were used for the photo-irradiation.
- Fig. 3-39. Raman spectra of the hex- I_xC_{60} which were obtained after the PIHP treatments up to ~30 GPa. The 457.9 nm (a), 514.5 nm (b) and the 647.1 nm

- (c) lines were used for the photo-irradiation.
- Fig. 3-40. Illustration of the peak fitting with Gaussian components to Raman spectra of the fcc-I_xC₆₀ and hex-I_xC₆₀ after PIHP treatments. The Raman spectra of the fcc-I_xC₆₀ and hex-I_xC₆₀ correspond to those shown in Fig. 3-38 (b) and Fig. 3-39 (b), respectively.
- Fig. 3-41. The frequency of the p-A_g mode of (a) fcc-I_xC₆₀ and (b) hex-I_xC₆₀ as functions of pressure applied at PIHP treatments. (a) \Box and \circ correspond to the results of the PIHP treatments with 514.5 nm and 488.0 nm lines, respectively. Δ denotes the results of the PIHP treatment for the C₆₀ single crystal with the 488.0 nm line. (b) \bullet , \Box and \circ correspond to the results of the PIHP treatments with 647.1 nm, 514.5 nm and 457.9 nm lines, respectively. Δ is same as in the case of (a).
- Fig. 3-42. Changes in the Raman spectra of the I_xC₆₀ compounds which were subjected to pressure treatments (a) without and (b) with photo-irradiation. The original Raman spectra of the I_xC₆₀ compounds are shown in the upper graph in (b).
- Fig. 3-43. Raman spectra of K₃C₆₀ at ambient condition after subjecting to the PIHP treatment at 200 °C. Experimental conditions are presented in the upper part of each graph.
- Fig. 3-44. Raman spectra of K₆C₆₀ at ambient condition after subjecting to photo-irradiation under HPHT conditions. Experimental conditions are presented in the upper part of each graph. In the left graph, two different PIHP treatments were performed onto the same sample continuously at room temperature (RT).
- Fig. 3-45. *In situ* high pressure Raman scattering spectra of Rb₃C₆₀ and K₃C₆₀ in the frequency range from 1350 cm⁻¹ to 1520 cm⁻¹.
- Fig. 3-46. *In situ* high pressure Raman scattering spectra of Rb₆C₆₀ and K₆C₆₀ in the frequency range from 1350 cm⁻¹ to 1550 cm⁻¹.
- Fig. 3-47. Peak frequencies of the $H_g(7)$, $H_g(8)$, and $A_g(2)$ mode of the C_{60} single crystal [77] and A_xC_{60} (A=Rb, K x=3, 6) as functions of pressure. Explanations of notation and pressure coefficients are given in the figure.
- Fig. 3-48. In situ high pressure Raman scattering spectra of Rb₃C₆₀ and K₃C₆₀ in the

- frequency range from 230 cm⁻¹ to 720 cm⁻¹.
- Fig. 3-49. *In situ* high pressure Raman scattering spectra of Rb₆C₆₀ and K₆C₆₀ in the frequency range from 230 cm⁻¹ to 720 cm⁻¹. Asterisks indicate the plasma lines of an Ar⁺ laser.
- Fig. 3-50. Peak frequencies of the $H_g(1)$ and $A_g(1)$ modes of the C_{60} single crystal and A_xC_{60} (A=Rb, K x=3, 6) as functions of pressure. Explanations of notation and pressure coefficients are given for $A_g(1)$ modes in the right figure.
- Fig. 3-51. Stereographic representation of the $H_g(1)$, $A_g(1)$, and $A_g(2)$ modes of the C_{60} molecule [81].
- Fig. 3-52. Normalized frequencies of $H_g(1)$, $A_g(1)$, and $A_g(2)$ mode are plotted as functions of normalized volumes of C_{60} single crystal and A_xC_{60} compounds (A=Rb, K x=3, 6). The Grüneisen parameters are tabulated for $A_g(1)$ and $A_g(2)$ modes.
- Fig. 4-1. In situ high pressure PL spectra of the T polymer. [76]