Decrease in electrical contact resistance of Sb-doped n⁺-BaSi₂ layers and spectral response of an Sb-doped n⁺-BaSi₂/undoped BaSi₂ structure for solar cells

Komomo Kodama, Ryota Takabe, Suguru Yachi, Kaoru Toko, Takashi Suemasu^{*} Institute of Applied Physics, Graduate School of Pure and Applied Sciences, University of Tsukuba, Tsukuba, Ibaraki 305-8573, Japan

We investigated how the electron concentration *n* in a 300-nm-thick Sb-doped n⁺-BaSi₂ layer grown by molecular beam epitaxy affected the contact resistance $R_{\rm C}$ to surface electrodes (Al, indium-tin-oxide). As the *n* of n-BaSi₂ increased, $R_{\rm C}$ decreased and reached a minimum of 0.019 Ω cm² at $n = 2.4 \times 10^{18}$ cm⁻³ for the Al electrodes. This value was more than one order of magnitude smaller than that obtained for Al/B-doped p-BaSi₂. We believe that this significant decrease in $R_{\rm C}$ came from Sb segregation. Furthermore, the internal quantum efficiency (*IQE*) spectrum was evaluated for an Sb-doped n⁺-BaSi₂(20 nm)/undoped BaSi₂(500 nm)/n⁺-Si(111) structure. Its *IQE* reached as high as ~50% over a wide wavelength range under a small bias voltage of 0.1 V applied between the top and bottom electrodes.

Email: suemasu@bk.tsukuba.ac.jp

1. Introduction

In many studies, various solar cells materials such as CdTe, chalcopyrites, kesterite, and perovskite have been assessed because of their low cost and high efficiency [1-6]. However, these materials contain elements that are scarce, toxic, or both. Thin-film Si solar cells have also been studied extensively; however, it is not easy to achieve efficiencies as high as 20% [7-17]. Thus, it is important to explore other materials for thin-film solar cells. Among such materials, we have focused on barium disilicide (BaSi₂), a semiconductor consisting of only abundant elements [18,19]. BaSi₂ is attractive for use in thin-film solar cells because of its bandgap of 1.3 eV, a matching solar spectrum [20], a high absorption coefficient of 3×10^4 cm⁻¹ at 1.5 eV comparable to those of chalcopyrites [21-23], and superior minority-carrier diffusion length ($L \sim 10 \ \mu$ m) [24,25] and lifetime ($\tau \sim 10 \ \mu$ s) [26-29]. Because of these properties, an energy conversion efficiency η of over 25% can be expected for a 2-µm-thick BaSi₂ pn-junction diode [30]. Recently, we have achieved a conversion efficiency near 10% with B-doped p-BaSi₂/n-Si heterojunction solar cells [31].

To fabricate BaSi₂ homojunction solar cells, we plan to use undoped BaSi₂ as an absorber layer in a solar cell. Undoped BaSi₂ has both a high α and a large *L*, both of which facilitate the collection of photogenerated carriers in an external circuit even though *L* degrades somewhat because of extrinsic effects such as crystal imperfections. In a solar cell, the topmost layer should be a heavily impurity-doped n- or p-type layer to decrease the contact resistance $R_{\rm C}$, which directly affects the fill factor *FF* of the solar cell. As $R_{\rm C}$ increases, *FF* decreases and thus η decreases. In many studies, $R_{\rm C}$ have been assessed in solar cell materials [32-38]. In our previous work, we investigated the $R_{\rm C}$ between Al electrodes and B-doped p-BaSi₂. The $R_{\rm C}$ reached a minimum of 0.35 Ω cm²[31] when the hole concentration *p* of p-BaSi₂ was 4.0 × 10¹⁸ cm⁻³. In this study, we first investigated the dependence of $R_{\rm C}$ between surface electrodes (Al or ITO) and Sb-doped n-BaSi₂ layers at various electron concentrations. We then evaluated the

spectral response of an Sb-doped n^+ -BaSi₂(20 nm)/undoped BaSi₂(500 nm) structure to verify how the heavily doped n^+ -BaSi₂ surface layer affected the spectral response, especially at short wavelengths.

2. Experimental methods

Here, we used an ion-pumped molecular beam epitaxy (MBE) system equipped with an electron-beam evaporation source for Si as well as standard Knudsen cells for Ba and Sb. Details of the growth procedure of undoped BaSi2 and Sb-doped n-BaSi2 were reported previously [39,40]. To measure $R_{\rm C}$, we used high-resistivity Czochralski (CZ) p-Si(111) substrates (resistivity $\rho = 1000 - 10000 \,\Omega$ cm) to minimize current flowing through the substrate. After thermally cleaning the substrate at the substrate temperature $T_{\rm S} = 900$ °C, we deposited Ba at 500 °C to form an approximately 5-nm-thick BaSi₂ template layer by reactive deposition epitaxy. This template worked as seeds that control the crystal orientation of the BaSi₂ overlayers [41]. Next, we formed 300-nm-thick Sb-doped n-BaSi₂ epitaxial layers by MBE at various *T*_S values of 460 (sample A), 480 (sample B), 500 (sample C), and 520 °C (sample D). The electron concentration n of Sb-doped n-BaSi₂ depends on $T_{\rm S}$ [40]. Then, a 3-nm-thick a-Si passivation layer was deposited in situ at $T_s = 180 \text{ °C}$ [42]. Finally, 150-nm-thick striped Al or ITO electrodes (length l = 0.6 mm, width W = 7.0 mm, electrode interval d = 1.0 mm) were formed by sputtering. Carrier concentration was measured by the van der Pauw method at room temperature (RT). $R_{\rm C}$ was measured by the transfer length method at RT [43]. We compared the smallest $R_{\rm C}$, $R_{\rm C} = 0.35 \ \Omega \ {\rm cm}^2$, measured from Al/p-BaSi₂ ($p = 4.0 \times 10^{18} \ {\rm cm}^{-3}$, sample E) [31]. The depth profile of Sb atoms was measured by secondary ion mass spectrometry (SIMS) using C_{S}^{+} ions.

Next, we investigated the photoresponse properties of the Sb-doped n⁺-BaSi₂/undoped BaSi₂ structure (sample F). A 500-nm-thick undoped BaSi₂ layer was epitaxially grown on a low-resistivity Cz n-Si(111) substrate ($\rho < 0.01 \ \Omega \ cm$) at $T_S = 580 \ ^{\circ}$ C, followed by a 20-nmthick Sb-doped n⁺-BaSi₂ ($n = 2.4 \times 10^{18} \ cm^{-3}$) layer and a 3-nm-thick a-Si passivation layer. We used the low-resistivity Si substrate to exclude the contribution of photogenerated carriers in the n-Si substrate and to form a good electrical contact at the BaSi₂/Si heterointerface. Finally, 80-nm-thick ITO electrodes with a diameter of 1 mm and 150-nm-thick Al rear electrodes were formed by sputtering. The reflectance *R* and external quantum efficiency (*EQE*) spectra were evaluated at RT using a lock-in technique with a xenon lamp and a single monochromator (Bunko Keiki, SM-1700A and RU-60N) with a focal length of 25 cm. Internal quantum efficiency (*IQE*) was obtained by dividing the *EQE* by (1–*R*). Light intensity was calibrated using a pyroelectric sensor (Melles Griot, 13PEM001/J). The crystalline quality of the grown layers was characterized by reflection high-energy electron diffraction (RHEED) and X-ray diffraction (XRD) analyses using Cu Kα radiation.

3. Results and discussion

Figure 1 shows the θ -2 θ XRD and RHEED patterns of samples A, B, C, and D. The RHEED patterns were observed along the Si $[11\overline{2}]$ azimuth. The diffraction peaks appeared only from the (100)-oriented BaSi₂ planes, such as (200), (400), and (600), showing that *a*-axis-oriented BaSi₂ epitaxial films were formed. However, the RHEED patterns were not streaky but spotty, meaning that the surface is rough. This surface roughness developed because the T_S for Sb-doped BaSi₂ was much lower [40] than the optimum T_S (~600 °C) for the undoped BaSi₂ absorber layers. The *n* values of samples A, B, C, and D were 6.9×10^{17} , 1.8×10^{18} , 2.4×10^{18} , and 2.8×10^{17} cm⁻³, respectively. Now, using sample C as an example, we will show how to obtain R_C . Figure 2 shows the relationship between the measured resistance r(x) between the Al or ITO electrodes and the electrode distance *x* for sample C ($n = 2.4 \times 10^{18}$ cm⁻³, $\mu = 100$ cm²/V s). Ignoring the lateral resistance of the electrode when the current flows in the *x* direction,

r(x) can be written as

$$r(x) \approx 2r_c + r_{\text{semi}}(x), \qquad (1)$$

where $r_{\text{semi}}(x)$ is the resistance of the semiconductor and $2r_{\text{C}}$ is the intercept of the vertical axis when we plot r(x) with respect to x. Because the second term on the right side of Eq. (1) is proportional to x, r(x) increases with increasing x. As shown in Fig. 2, the line for the ITO electrodes has a slightly higher slope and thus a larger resistance than that for Al electrodes. This behavior probably occurred because the ITO electrode had larger lateral resistance than Al. From the plots in Fig. 2, $r_c = 2.26$ and 1.22Ω were obtained for the ITO and Al electrodes, respectively. R_{C} is given by

$$R_c = r_c \times W \times L_t, \quad (2)$$

where L_t is the transfer length, which is half of the intercept of the horizontal axis of the straight line in Fig. 2. From Eq. (2), R_c was calculated to be 0.057 and 0.019 Ω cm² for the ITO and Al electrodes, respectively, in sample C.

Figure 3 summarizes the dependence of $R_{\rm C}$ on *n* for all the samples obtained by the method described above. $R_{\rm C}$ decreased with increasing *n*. The smallest $R_{\rm C}$, 0.019 $\Omega \cdot \rm{cm}^2$, was obtained from Al electrodes in sample C. This value is more than one order of magnitude smaller than the smallest $R_{\rm C}$ (0.35 Ω cm²) obtained from Al/B-doped p-BaSi₂ ($p = 4.0 \times 10^{18}$ cm⁻³, $\mu = 28$ cm²V⁻¹s⁻¹) [31]. We will now discuss the reason for this behavior. According to Ref. 36, the sheet resistance $R_{\rm sheet}$ is given by

$$R_{\rm sheet} = R_{\rm C}/L_{\rm t}^2 \,. \tag{2}$$

For sample C, R_{sheet} is 0.40 Ω from $L_{\text{t}} = 0.218$ mm and $R_{\text{C}} = 0.019 \ \Omega \text{ cm}^2$, while R_{sheet} equals 6.79 Ω using $L_{\text{t}} = 0.227$ mm and $R_{\text{C}} = 0.35 \ \Omega \cdot \text{cm}^2$ for sample E. Such a large difference in R_{sheet} between samples C and E cannot be anticipated from their similar ρ values: 0.0644 and 0.0557 Ω cm, respectively, which were determined by Hall measurements. It is known that B atoms in B-doped p-BaSi₂ do not segregate and so are almost uniformly distributed [44]. Figure 4 shows the depth profile of Sb atoms and secondary ions (Ba + Si) in sample C. The Sb atoms segregate toward the surface of sample C, and the Sb concentration exceeds 10^{20} cm⁻³ near the electrode. Segregation of Sb has been observed in various semiconductors such as Si, Ge, and III–V semiconductors [45-52]. Hence, *n* is actually much higher in the surface region than the *n* found from the Hall measurement ($n = 2.4 \times 10^{18}$ cm⁻³), leading to the significant decrease in R_{sheet} and thus R_{C} in sample C. Owing to the segregation of Sb in BaSi₂, an Sb-doped n⁺-BaSi₂ layer should be placed on the top of the BaSi₂ solar cell. Hence, an Sb-doped n⁺-BaSi₂/undoped BaSi₂ absorber layered structure is useful for practical applications.

We next assessed the spectral response of sample F, with an Sb-doped n⁺-BaSi₂/undoped BaSi₂ structure. Figure 5 shows a schematic of sample F for this purpose and the RHEED patterns obtained after the growth of each layer. Because BaSi₂ has a large α , shortwavelength light is absorbed near the surface, in the Sb-doped n⁺-BaSi₂ layer. Therefore, defective n⁺-BaSi₂ deteriorates the solar-cell performance. Figure 6 shows the θ -2 θ XRD pattern, revealing *a*-axis-oriented epitaxial growth up to the 20-nm-thick Sb-doped n⁺-BaSi₂. Figure 7 shows the *IQE* spectrum of sample F. A bias voltage of -0.1 V was applied to the top ITO electrode with respect to the back side-surface Al electrode so that the photogenerated holes in the BaSi₂ were extracted into the ITO electrode. The IQE increased sharply at wavelengths shorter than approximately 1000 nm, corresponding to the band gap of BaSi₂, and reached approximately 50% over a wide wavelength range even under this small bias voltage. We observed no pronounced reduction in *IQE* at short wavelengths, even when a highly Sbdoped layer was inserted. In a BaSi₂ homojunction solar cell, the built-in potential across the pn junction will be much higher than 0.1 V, producing larger IQEs. On the basis of these findings, we conclude that the Sb-doped n⁺-BaSi₂/undoped BaSi₂ absorber layered structure is suitable for practical BaSi₂ solar cells.

4. Conclusions

We investigated the dependence of $R_{\rm C}$ on the electron concentration of Sb-doped n-BaSi₂ and achieved a minimum $R_{\rm C}$ of 0.019 Ω cm² for Al/Sb-doped n-BaSi₂ at $n = 2.4 \times 10^{18}$ cm⁻³. This $R_{\rm C}$ was more than one order of magnitude smaller than the $R_{\rm C}$ obtained for Al/B-doped p-BaSi₂, 0.35 $\Omega \cdot$ cm². We explained this significant reduction in $R_{\rm C}$ in Al/Sb-doped n-BaSi₂ as being due to Sb segregation, which was verified by SIMS. The concentration of Sb atoms exceeded 10^{20} cm⁻³ near the electrode, significantly decreasing $R_{\rm C}$. We achieved an IQE > 50% over a wide wavelength range under a small bias voltage of 0.1 V in the n⁺(20 nm)/undoped BaSi₂(500 nm)/n⁺-Si(111) structure.

Acknowledgements

This work was financially supported by JSPS KAKENHI Grant Numbers 15H02237 and 17K18865. R.T. was financially supported by a Grant-in-Aid for JSPS Fellows (15J02139).

References

- [1] M. G. Panthani, V. Akhavan, B. Goodfellow, J. P. Schmidtke, L. Dunn, A. Dodabalapur,
- P. F. Barbara, and B. A. Korgel, J. Am. Chem. Soc. 130, 16770 (2008).
- [2] W. Jaegermann, A. Klein, and T. Mayer, Adv. Mater. 21, 4196 (2009).
- [3] S. Ahmed, K. B. Reuter, O. Gunawan, L. Guo, L. T. Romankiw, and H. Deligianni, Adv. Energy Mater. **2**, 253 (2012).
- [4] C. Steinhagen, M. G. Panthani, V. Akhavan, B. Goodfellow, B. Koo, and B. A. Korgel, J. Am. Chem. Soc. 131, 12554 (2009).
- [5] M. A. Green, A. Ho-Baillie, and H. J. Snaith, Nat. Photonics 8, 506 (2014).
- [6] H. S. Kim, S. H. Im, and N. G. Park, J. Phys. Chem. C 118, 5615 (2014).
- [7] R. G. Gordon, J. Proscia, F. B. Ellis, Jr., and A. E. Delahoy, Sol. Energy Mater. 18, 263 (1989).
- [8] P. Campbell, Sol. Energy Mater. 21, 165 (1990).
- [9] H. Sasaki, H. Morikawa, Y. Matsuno, M. Deguchi, T. Ishihara, H. Kumabe, T. Murotani, and S. Mitsui, Jpn. J. Appl. Phys. **33**, 3389 (1994).
- [10] J. Meier, S. Dubail, R. Platz, P. Torres, U. Kroll, J. A. A. Selvan, N. P. Vaucher, Ch. Hof,
- D. Fischer, H. Keppner, R. Flückiger, A. Shah, V. Shklover, and K.-D. Ufert, Sol. Energy Mater. Sol. Cells **49**, 35 (1997).
- [11] O. Vetterl, F. Finger, R. Carius, P. Hapke, L. Houben, O. Kluth, A. Lambertz, A. Mück,B. Rech, and H. Wagner, Sol. Energy Mater. Sol. Cells 62, 97 (2000).
- [12] A. Poruba, A. Fejfar, Z. Remes, J. Springer, M. Vanecek, and J. Kocka, J. Appl. Phys. 88, 148 (2000).
- [13] J. Müller, B. Rech, J. Springer, and M. Vanecek, Sol. Energy 77, 917 (2004).
- [14] M. Berginski, J. Hüpkes, M. Schulte, G. Schöpe, H. Stiebig, and B. Rech, J. Appl. Phys.101, 074903 (2007).

- [15] D. Zhou and R. Biswas, J. Appl. Phys. 103, 093102 (2008).
- [16] A. Hongsingthong, T. Krajangsang, I. A. Yunaz, S. Miyajima, and M. Konagai, Appl. Phys.Express 3, 051102 (2010).
- [17] H. Sai, Y. Kanamori, and M. Kondo, Appl. Phys. Lett. 98, 113502 (2011).
- [18] J. Evers, G. Oehlinger, and A. Weiss, Angew. Chem., Int. Ed. 16, 659 (1977).
- [19] M. Imai and T. Hirano, J. Alloys Compd. 224, 111 (1995).
- [20] K. Morita, Y. Inomata, and T. Suemasu, Thin Solid Films 508, 363 (2006).
- [21] D. B. Migas, V. L. Shaposhnikov, and V. E. Borisenko, Phys. Status Solidi B 244, 2611(2007).
- [22] K. Toh, T. Saito, and T. Suemasu, Jpn. J. Appl. Phys. 50, 068001 (2011).
- [23] M. Kumar, N. Umezawa, and M. Imai, J. Appl. Phys. 115, 203718 (2014).
- [24] M. Baba, K. Toh, K. Toko, N. Saito, N. Yoshizawa, K. Jiptner, T. Sakiguchi, K. O. Hara,
- N. Usami, and T. Suemasu, J. Cryst. Growth 348, 75 (2012).
- [25] M. Baba, K. Watanabe, K. O. Hara, K. Toko, T. Sekiguchi, N. Usami, and T. Suemasu, Jpn. J. Appl. Phys. 53, 078004 (2014).
- [26] K. O. Hara, N. Usami, K. Toh, M. Baba, T. Toko, and T. Suemasu, J. Appl. Phys. 112, 083108 (2012).
- [27] K. O. Hara, N. Usami, K. Nakamura, R. Takabe, M. Baba, K. Toko, and T. Suemasu, Appl.Phys. Express 6, 112302 (2013).
- [28] R. Takabe, K. O. Hara, M. Baba, W. Du, N. Shimada, K. Toko, N. Usami, and T. Suemasu,
- J. Appl. Phys. 115, 193510 (2014).
- [29] T. Suemasu and N. Usami, J. Phys. D Appl. Phys. 50, 023001 (2017).
- [30] T. Suemasu, Jpn. J. Appl. Phys. 54, 07JA01 (2015).
- [31] S. Yachi, R. Takabe, K. Toko, and T. Suemasu, Appl. Phys. Lett. 109, 072103 (2016).
- [32] D. K. Schroder and D. L. Meier, IEEE Trans. Electron Devices 31, 637 (1984).

[33] S. Y. Lee, H. Choi, H. Li, K. Ji, S. Nam, J. Choi, S. W. Ahn, H. M. Lee, and B. Park, Sol.Energy Mater. Sol. Cells 120, 412 (2014).

[34] G. J. Bauhuis, P. Mulder, E. J. Haverkamp, J. C. C. M. Hujiben, and J. J. Schermer, Sol. Energy Mater. Sol. Cells **93**, 1488 (2009).

- [35] K. Kuribayashi, H. Matsumoto, H. Uda, Y. Komatsu, A. Nakano, and S. Ikegami, Jpn. J. Appl. Phys. 22, 1828 (1983).
- [36] K. Nishioka, T. Takamoto, T. Agui, M. Kaneiwa, Y. Uraoka, and T. Fuyuki, Sol. Energy Mater. Sol. Cells **90**, 1308 (2006).
- [37] M. A. Khan, M. S. Shur, and Q. Chen, Appl. Phys. Lett. 68, 3022 (1996).
- [38] L. L. Smith, R. F. Davis, M. J. Kim, R. W. Carpenter, and Y. Huang, J. Master. Res. **12**, 2249 (1997).
- [39] R. Takabe, K. Nakamura, M. Baba, W. Du, M. Ajmal Khan, K. Toko, M. Sasase, K. O. Hara, N. Usami, and T. Suemasu, Jpn. J. Appl. Phys. **53**, 04ER04 (2014).
- [40] M. Kobayashi, Y. Matsumoto, Y. Ichikawa, D. Tsukada, and T. Suemasu, Appl. Phys.Express 1, 051403 (2008).
- [41] Y. Inomata, T. Nakamura, T. Suemasu, and F. Hasegawa, Jpn. J. Appl. Phys. 43, L478 (2004).
- [42] R. Takabe, H. Takeuchi, W. Du, K. Ito, K. Toko, S. Ueda, A. Kimura, and T. Suemasu, J.Appl. Phys. **119**, 165304 (2016).
- [43] P. N. Vinod, J. Mater. Sci.: Mater. Electron. 22, 1248 (2011).
- [44] D. Tsukahara, M. Baba, K. Watanabe, T. Kimura, K. O. Hara, W. Du, N. Usami, K. Toko,
- T. Sekiguchi, and T. Sumasu, Jpn. J. Appl. Phys. 54, 030306 (2015).
- [45] K. D. Hobart, D. J. Godbey, M. E. Twigg, M. Fatemi, and P. E. Thompson, Surf. Sci. 334, 29 (1995).
- [46] A. Gulino, G. G. Condorelli, K. Fragala, and R. G. Egdell, Appl. Surf. Sci. 90, 289 (1995).

- Si
- [47] R. Kaspi and K. R. Evans, J. Cryst. Growth 175, 838 (1997).
- [48] D. Reinking, M. Kammler, M. HornVonHoegen, and K. R. Hofmann, Appl. Phys. Lett.

71, 924 (1997).

[49] O. J. Pitts, S. P. Watkins, C. X. Wang, V. Fink, and K. L. Kavanagh, J. Cryst. Growth **254**, 28 (2003).

[50] H. S. Wong, L. Chan, G. Sainudra, and Y. C. Yeo, IEEE Trans. Electron Devices 18, 703

(2007).

- [51] M. Oehme, J. Werner, and E. Kasper, J. Cryst. Growth 310, 4531 (2008).
- [52] D. V. Yurasov, A. V. Antonov, M. N. Drozdov, V. B. Schmagin, K. E. Spirin, and A. V.
- Novikov, J. Appl. Phys. 118, 145701 (2015).

Figure captions

Fig. 1 θ -2 θ XRD and RHEED patterns observed along the Si [11 $\overline{2}$] azimuth of samples A, B, C, and D. The asterisk (*) indicates the diffraction of the Si substrate used.

Fig. 2 Sample structure for $R_{\rm C}$ measurement and the relationships between resistance and Al or ITO electrode distance for sample C, Sb-doped n-BaSi₂ ($n = 2.4 \times 10^{18} \,{\rm cm}^{-3}$).

Fig. 3 $R_{\rm C}$ as a function of electron concentration for Al/n-BaSi₂ and ITO/n-BaSi₂. The smallest $R_{\rm C}$ (0.35 Ω cm²) obtained from Al/p-BaSi₂ [31] is shown for comparison.

Fig. 4 SIMS depth profile of Sb atoms and secondary ions (Ba + Si).

Fig. 5 Sample structure of sample F and RHEED patterns of each layer observed along the Si [110] azimuth.

Fig. 6 θ -2 θ XRD pattern of sample F. The asterisk (*) indicates the diffraction of the Si substrate used.

Fig. 7 *IQE* spectrum of sample F under a bias voltage of -0.1 V applied to the surface ITO electrode with respect to the back side-surface Al electrode.

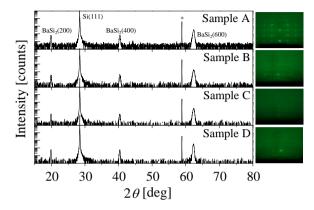


Fig. 1

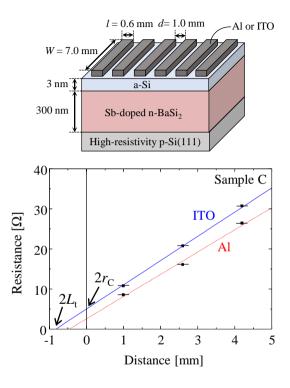


Fig. 2

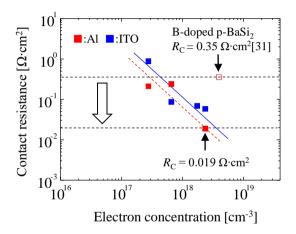


Fig. 3

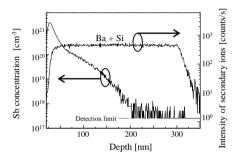


Fig. 4

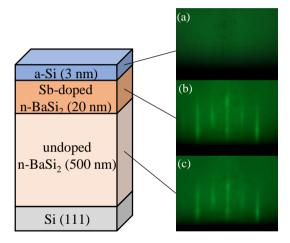


Fig. 5

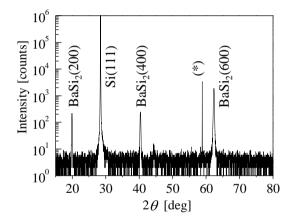


Fig. 6

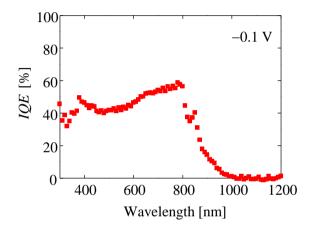


Fig. 7