1	Tracing groundwater recharge sources in a mountain-plain transitional area using
2	stable isotopes and hydrochemistry
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12	Abstract
13	Mountain-plain transitional landscapes are especially important as groundwater
14	recharge zones. In this study, the oxygen and hydrogen stable isotopic composition
15	($\delta^{18}O$ and δD) of water and hydrochemical information were employed to quantify
16	contribution ratios of different sources of groundwater recharge in the Ashikaga area of
17	central Japan. The study area is situated between the Ashio Mountains and the Kanto
18	plain, and the Watarase River flows into the region parallel to the mountain-plain
19	boundary. There was an obvious isotopic altitude effect in and around the study area
20	(-0.25‰ per 100 m for δ^{18} O and -1.7‰ per 100 m for δ D), and the isotopic signatures
21	of water from the Watarase River and local precipitation could be clearly distinguished.
22	In addition, it was possible to identify the occurrence of mountain block recharge using
23	hydrochemistry, specifically the chloride ion. End-member mixing analysis using δ
24	values and Cl ⁻ concentration revealed spatial variation in the contribution ratios of the

1 river water, mountain block groundwater and local precipitation. Seepage from the 2 Watarase River contributed a significant amount of water to aquifers along its channel. 3 The river-recharged water reached 5 km from the channel in the south (i.e., plain) side 4 and 1.6 km or less in the north (i.e., mountain) side. Remarkable mountain block 5 recharge was observed in the foothills near the axis of the syncline, which has layers of 6 chert and sandstone that likely hinder river channel seepage into the mountain-side 7 aquifers. Major factors controlling the river-water contribution ratio include the distance 8 from the river channel, topography, and hydrogeological settings. The results presented 9 here should facilitate integrated management of groundwater and surface water 10 resources.

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Key words: Groundwater recharge, river channel seepage, mountain block recharge,
end-member mixing analysis (EMMA), isotopic tracer, groundwater-river interaction

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15 **1. Introduction**

16 A large amount of groundwater recharge occurs along mountain fronts in semiarid 17 regions (Wilson and Guan, 2004). Even under humid climates, deep groundwater in 18 plains aquifers is recharged along the margins of plains adjacent to mountains (Mikita et 19 al., 2011; Yamanaka et al., 2011a). Clearly identifying recharge sources and their 20 corresponding contributions is prerequisite to sustainable management of aquifers to 21 ensure that the needs of humans and ecosystems are met. Groundwater recharge in 22 mountain-plain transitional areas can generally be divided into direct recharge of local 23 precipitation and mountain-front recharge. Mountain-front recharge is subdivided into 24 surface and subsurface components, with the former representing seepage from surface

water bodies (e.g., river seepage) and the latter consisting of mountain block recharge,
 which is the subsurface water flow from the mountain block to an adjacent basin
 (Manning and Solomon, 2005; Aishlin and McNamara, 2011).

4 Numerous studies have demonstrated that river water seepage is an important 5 recharge mechanism along river reaches (Vanderzalm et al., 2011). So-called "induced 6 flow" from rivers can occur to replenish water levels in aquifers that have been depleted 7 as a result of groundwater abstraction (Sanz et al., 2011). However, overexploitation of 8 water resources in the upstream and midstream portions of a river can cause 9 groundwater decline and quality deterioration in downstream regions (Xi et al., 2010). 10 In addition, mountain block recharge can be restricted by geological structures such as 11 faults and folds (Wilson and Guan, 2004). For example, a normal fault can act as a 12 conduit for the transfer of precipitation in mountain areas to a plain (Yuan et al. 2011). 13 Geological folding diverts groundwater along syncline axes from mountain aquifers 14 toward rift valley aquifers (Laronne Ben-Itzhak and Gvirtzman, 2005).

15 The hydrochemistry and isotope tracer approaches have been widely used for 16 determination of recharge fluxes based on conceptually simple models (de Vries and 17 Simmers, 2002). Stable isotopic compositions of groundwater recharged through direct 18 infiltration of precipitation will reflect that of local precipitation (Clark and Fritz, 1997). 19 If rivers retain the depleted isotopic signature of their headwaters, the difference in 20 stable isotope signatures of rivers and local precipitation can be used to determine the 21 relative contribution of these two sources of groundwater recharge (Scanlon et al., 2002; 22 Lambs, 2004; Kalbus et al., 2006). Hydrochemical data can be used to complement 23 isotopic information. Such an approach is especially useful in areas for which 24 hydrometric data are not available.

We present here a case study of groundwater recharge and groundwater-river interaction in a mountain-plain transitional area with syncline geological settings. The specific objectives of the present study are to identify recharge sources of groundwater and quantitatively evaluate their contribution ratio using stable isotopes and hydrochemistry, with a special emphasis on the spatial extent of river-recharged groundwater and its controlling factors.

7

8 **2. Material and methods**

9 2.1. Study area

10 The Ashikaga area of central Japan (Fig. 1) was selected as the study site. This area 11 is situated at the northern margin of the Kanto Plain, which is the largest plain in Japan. 12 The area is bounded on the north by the Ashio Mountains. The Watarase River, which 13 has a total length of 106.7 km, originates in Mt. Sukai (elevation, 2,144 m above mean 14 sea level (a.m.s.l.)) in the Ashio Mountains (Fig. 1a) and runs across the central part of 15 the study area. The river flows to the southwest and then turns southeast along the 16 margin of the plain. The study area is located in the middle reaches of this river (Fig. 17 1b). There are several tributaries to the Watarase within the study area that originate in 18 low mountains with elevations that range from 223 to 681 m a.m.s.l. Since the discharge 19 rates of tributaries are far less than the Watarase River and all the river banks are built 20 using pebble and concrete, the river seepage of tributaries was neglected and the 21 Watarase River seepage was the focus in this study.

The study area is located in a humid temperate zone that shows broad variations in temperature. Summer in the region is characterized by warm and humid conditions, while winter is cold and relatively arid. Based on observed data collected by the Japan Meteorological Agency (http://www.jma.go.jp/jma/index.html) from 1981-2010, the monthly mean air temperature ranges from 2.8 °C (January) to 25.7 °C (August), and the annual precipitation is approximately 1,200 mm. About 75% of the annual precipitation occurs during the rainy season from May to October. The precipitation data were collected from Ashikaga Station (36 °18.1' N, 139 °28.4' E and 28 m a.m.s.l.), while the temperature data were from Sano Station (36 °21.8' N, 139 °34.2'E and 68 m a.m.s.l.).

Forest is the dominant land cover for the mountains, while residential areas are mainly located in the lowlands along the Watarase River. Rice paddy fields are also distributed along the Watarase River and its tributaries, but their total area accounts for only 10% of the entire study area and most of them are far from the sampled wells. In addition, irrigation occurs during rainy seasons from June to September. Therefore, seepage from paddy fields is expected to be a minor recharge source for most of the sites in our study area.

14 No facilities consuming large amounts of water were found in the study area, 15 except for seven water purification plants (AT1 - AT7 in Fig. 1) with 17 municipal wells. 16 These plants are distributed along the Watarase River and supply tap water to local 17 residents. The municipal water supply system in Ashikaga City is solely dependent on 18 groundwater. In addition to tap water, many residents also utilize water drawn from 19 private wells on their own land for domestic purposes. Because these wells are private, 20 hydrometric measurements are not available. However, the well depths of some wells 21 were inquired from local owners and most of the wells are shallow wells with depth 22 between 5-12 m except one deep well (AW6 in Fig. 1), of which depth is 150 m. In 23 addition, the municipal wells of AT1, AT2, AT3 and AT5 are also shallow wells with 24 depth no more than 11 m. Although municipal wells AT4, AT6 and AT7 are deep wells

with depth 40-80 m, their well screens are multi-depths and the uppermost screens are in the shallow aquifer. It should also be noted that the drainage system in the study area is well developed; therefore, domestic wastewater can be rapidly drained to rivers and away from the study area without re-entering local groundwater systems. Accordingly, the seepage of domestic wastewater to groundwater in the study area was not considered in this study.

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8 2.2. Geographical and hydrogeological settings

9 As shown on the surficial geology map (Fig. 2a), the mountains in the study area 10 are characterized by a series of distinct NE-SW trending anticline and syncline tectonic 11 movements. The lithology of the mountains consists mostly of chert, sandstone with 12 interbedded slate or mudstone, and some limestone. Plains and valleys in the region are 13 overlain by recent alluvial sediments (sand and gravel) and volcanic soil (loam). The 14 geologic cross section (Fig. 2b) shows that the thickness of the aquifer increases from 15 55 to more than 150 m from northwest to southeast in the plain. The aquifer is mainly 16 composed of gravel and sand, and the aquitard consists of red and blue clay. The 17 bedrock is composed of chert.

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19 2.3. Sampling and analysis

20 Precipitation samples were collected monthly during a 1-year period from 21 November 2010 to October 2011. Field surveys were carried out six times 22 approximately bimonthly during the period. Overall, 12 groundwater samples were 23 collected from private wells (AW1-AW12 in Fig. 1) and four river water samples 24 (AR1-AR4 in Fig.1) were collected at each sampling point. Water temperature,

electrical conductivity (EC) and pH were measured in situ with portable instruments.
Additionally, groundwater samples from 17 municipal wells utilized by seven water
purification plants (AT1-AT7 in Fig. 1) were collected monthly from December 2010 to
October 2011. All samples were subjected to isotope analysis. In addition,
hydrochemical analysis was conducted using 23 groundwater samples collected from
private wells and seven river water samples in September 2010 and April 2011, and 8
groundwater samples from municipal wells in September 2010.

8 The isotopic and chemical compositions of the water samples were analyzed at the 9 Terrestrial Environment Research Center, University of Tsukuba. The hydrogen and 10 oxygen stable isotopic composition of the water samples were determined using a liquid 11 water isotope analyzer (L1102-i, Picarro, Santa Clara, CA, USA) based on 12 wavelength-scanned cavity ring-down spectroscopy (WS-CRDS; Gupta et al., 2009). The isotopic ratios of ¹⁸O and D were expressed as δ^{18} O and δ D, respectively, relative to 13 the Vienna Standard Mean Ocean Water. The analytical errors were 0.1% for δ^{18} O and 14 1‰ for δD (Yamanaka and Onda, 2011) for samples collected from November 2010 to 15 July 2011, while they were 0.2‰ for δ^{18} O and 2‰ for δ D for samples collected after 16 17 August 2011 owing to replacement of the analyzer. Cation and anion concentrations 18 were measured using an ion chromatograph (LC-10AP, Shimadzu, Tokyo, Japan) with 19 an analytical precision of ± 1 mg/L. The bicarbonate concentration was measured by the 20 titration method using sulfuric acid. The accuracy of the hydrochemical analyses was 21 checked using an electrical balance. Samples with relative errors greater than 5% were 22 reanalyzed.

3. Results and discussion

2 3.1. Isotopic signatures

3 The isotopic compositions of precipitation change temporally in response to 4 rainout history and various factors such as temperature effect and amount effect (Clark 5 and Fritz, 1997). The d-excess of precipitation had statistically significant, negative 6 correlation with temperature (r = -0.87, p < 0.001). A similar tendency was reported by 7 Waseda and Nakai (1983), who stated that it was caused by seasonal shifting of the 8 water vapor source between the Pacific and the Sea of Japan. In contrast to the high 9 variability in δ values of precipitation, those of river water and groundwater remained 10 nearly constant (Fig. 3). Although the d-excess of river water and groundwater showed 11 weak variation, its phase was not in agreement with that of precipitation. These features 12 suggest that the mean residence times of river water and groundwater are considerably 13 long (at least > 1 year).

14 As expected from the seasonal changes in the d-excess of precipitation, the local 15 meteoric water line (LMWL) which was drawn based on 1-year observed data differed remarkably between the warm period (April-September; $\delta D = 8.65\delta^{18}O + 14.8$, $r^2 = 0.98$) 16 and the cool period (October-March: $\delta D = 8.42\delta^{18}O+23.8$, $r^2 = 0.98$). The LMWL in the 17 18 warm period is very similar to the global meteoric water line (GMWL; $\delta D = 8\delta^{18}O+10$; 19 Craig, 1961), with a slightly higher slope (Fig. 4). The observed isotopic compositions 20 of river water and groundwater (except for AW7) were plotted along the LMWL for the 21 warm period, indicating that precipitation in the warm period is a more effective recharge source than precipitation during the cool period. The arithmetic means of δ^{18} O 22 23 and δD of water from the Watarase River (AR1 in Fig. 1) were lowest among all samples collected during the study period. The weighted (based on monthly 24

precipitation) mean δ^{18} O and δ D of precipitation were -9.26‰ and -62.5‰, respectively. 1 2 Total precipitation during the period was 236 mm (20%) higher than the climatic normal, in part because an extraordinary amount of rainfall occurred in May, July and 3 4 September, 2011. Therefore, the 1-year observation period may be insufficient to 5 provide a representative isotopic signature of the local precipitation. If the climatic 6 normals (1981-2010) of monthly precipitation are used to calculate the weighted mean δ 7 values, significantly higher values ($\delta^{18}O = -8.58\%$ and $\delta D = -57.4\%$) are obtained. 8 Since the extraordinary amount of rainfall should have reduced the observed δ values of 9 precipitation, long-term mean values in this area are expected to be higher.

10 A close relationship existed between the mean δ values of river water and the mean 11 elevation of the catchment corresponding to each sampling point (Fig. 5), which reflects 12 the well-known altitude effect (Dansgaard, 1964; Clark and Fritz, 1997). Accordingly, 13 the low δ values of the Watarase River water (AR1) are due to the high elevation of its recharge zone. The isotopic lapse rate was calculated to be -0.25% per 100 m for δ^{18} O 14 15 and -1.7‰ per 100 m for δD . The global average isotopic lapse rate reported by Poage and Chamberlain (2001) is -0.28‰ per 100 m for δ^{18} O, which is similar to the results 16 17 reported here. Based on our isotopic lapse rate and the mean elevation of the entire 18 study area, the long-term mean δ^{18} O and δ D of the local precipitation are approximately 19 -8.26‰ and -56.6‰, respectively. Although these values are higher than the observed 20 weighted mean values for the study period, they are consistent with our expectations. 21 Thus, we adopted these estimated values rather than those based on only 1-year 22 observations as the isotopic signature of the local precipitation.

23 The arithmetic mean values of groundwater δ^{18} O and δ D at each well during the 24 study period ranged from -9.48‰ to -7.83‰ and from -65.0‰ to -55.5‰, respectively

1 (Table 1). With the exception of AW7, the mean δ values of groundwater were between 2 those of Watarase River water and the estimated local precipitation values. The isotopic 3 composition of groundwater at AW7 (with the lowest d-excess) was plotted on the 4 right-hand side of the meteoric water lines at a distance, suggesting that it was affected by kinetic fractionation, probably due to evaporation from water or ground surfaces. 5 The lowest δ^{18} O and δ D values were found in well AT3, which is only 0.68 km from the 6 Watarase River. The δ^{18} O and δ D of groundwater on the south side of the Watarase 7 8 River ranged from -9.48 to -8.45‰ and from -65.0 to -58.9‰, respectively. Conversely, 9 the values on the north side ranged from -9.22 to -7.83‰ and from -63.0 to -55.5‰, 10 respectively. The values were generally more negative on the south (i.e., plain) side than 11 the north (i.e., mountain) side, which is seemingly contrary to the altitude effect. As 12 shown in Figure 6, the Watarase River water signal decreased with distance from the 13 Watarase River for well waters on the south side due to the dispersion and diffusion 14 processes. However, the relationship between δ values and the distance from the 15 Watarase River is not very obviously for well waters on the north side. Relatively high δ 16 values were found at AT5 and AW8 near the Watarase River and relatively low δ values 17 were found at AW9 far from the Watarase River on the north side. Since most of the 18 groundwater samples including AT5 and AW8 were from shallow wells, the samples 19 could represent well mixed water signals of recharge sources very well. Although the 20 depth of AW9 is unknown, these results suggest that the Watarase River contributes a 21 significant amount of water to aquifers, especially on its south side and some processes, 22 which restrict Watarase River seepage flow from the river channel into its north side 23 aquifers probably occur.

1 3.2. Hydrochemical characteristics

2 The EC reflects the total dissolved ion concentrations in water bodies, and to a 3 certain extent, the length of flow paths and residence times underground (Song et al., 4 2006). The EC values of groundwater ranged from 11.9 to 44.4 mS/m, while those of 5 river water ranged from 8.3 to 18.1 mS/m (Table 1). The lowest EC for groundwater 6 was found at AW11, which was a very shallow well situated in the foothills, suggesting 7 a shallow flow path along the hillslope with a relatively short residence time. The 8 highest EC for groundwater was found at AW7, where the d-excess value was lowest, 9 indicating an evaporative enrichment effect. The second highest EC of 36.6 mS/m was 10 found at AW9, indicating the relatively longer residence time for the groundwater at this 11 site.

The groundwater pH ranged from 6.11 to 7.53, with the highest value being observed at the deepest well (AW6, 150 m). The depths of the other private wells and some municipal wells (AT1, AT2, AT3, and AT5) were less than 12 m, while that of AW9 is unknown. Lower pH values (e.g., < 6.0) were found in very shallow wells in the foothills (AW 11).

17 The trilinear diagram suggests that Ca-HCO₃ is the dominant hydrochemical facies 18 in the study area (Fig. 7), which is common to ordinary groundwater systems. However, for AW9 and AW6, the proportion of Ca^{2+} relative to the total cations (in meq/L) is 19 20 lower and that of Na^+ is higher than for the other wells. These characteristics are similar 21 to those for tributaries of the Watarase River and may reflect geology specific to their 22 recharge areas. Groundwater in the foothills (AW9, AW11 and AW12) and from a deep well (AW6), as well as water from the tributaries (AR2, AR3 and AR4) was 23 characterized by poor SO42- content. Conversely, water samples from the Watarase 24

1 River and the adjacent groundwater had relatively high SO_4^{2-} contents. Because the 2 absolute SO_4^{2-} concentration was greater in groundwater than in water from the 3 Watarase River, the SO_4^{2-} likely originated from sediments deposited by the river rather 4 than the river water itself.

5 NO_3^- was detected in all river water and groundwater samples except for AW7 and 6 AW9. These findings suggest a certain measure of contamination by fertilizers, sewage 7 and acid deposition. The absence of NO_3^- indicates that the water has never been 8 contaminated or that natural attenuation of nitrate (i.e., denitrification) has occurred in 9 the system.

10

11 3.3 End-member mixing analysis

12 End-member mixing analysis (EMMA) based on the mass balance of tracers has 13 conventionally been used for hydrograph separation (Burns et al., 2001; Doctor et al., 14 2006). EMMA can also be used to evaluate the contribution ratio of each recharge 15 source of groundwater (Wakui and Yamanaka, 2006; Nakaya et al., 2007; Qin et al., 16 2011). This technique assumes the following (Barthold et al., 2011): (1) groundwater is 17 a mixture of source substances with a fixed composition; (2) the mixing process is 18 linear and completely dependent on hydrodynamic mixing; (3) the substances used as 19 tracers are conservative; and (4) the source substances have extreme concentrations.

Previous studies have shown that oxygen and hydrogen stable isotope compositions of water have the potential for use in tracing both sources and flow paths in groundwater systems (Nakaya et al., 2007). Chloride is a very conservative tracer, and Subyani (2004) demonstrated its usefulness as a groundwater tracer to evaluate groundwater recharge in alluvial sediments.

1 Fig. 8 shows a plot of annual (arithmetic) mean δ values against Cl⁻ in river water 2 and groundwater. Data for all groundwater samples except AW7 are plotted within a 3 triangle. Watarase River water is clearly the end-member corresponding to the lower left 4 vertex of the triangle. Local precipitation had estimated long-term mean δ values and 5 low chloride concentrations and was therefore likely the end-member forming the upper 6 left vertex. Because the chloride concentration of precipitation is not highly variable in 7 space, the chloride concentration of 0.5 mg/L for precipitation at a nearby city (Miyoshi, 8 2012; personal communication) was used. The δ values and Cl⁻ concentration of this 9 end-member were similar to those of samples collected from the tributaries (except 10 AR4). The third end-member appears to have moderate δ values and a high Cl⁻ 11 concentration, which is similar to AW9. As mentioned above, the hydrochemical 12 characteristics of groundwater at AW9 suggest a deep, long flow path. Therefore, this 13 end-member is considered to be the mountain block recharge component. The medium δ 14 values of this end-member indicate that mountain block groundwater was recharged at 15 an elevation between those of the upper Watarase River watershed and the present study 16 area. The reasons for the high Cl⁻ in the mountain block groundwater are discussed later. 17 It should be noted that the AW7 groundwater and AR4 river water differed from 18 the other samples. The isotopic signatures of groundwater at AW7 suggest evaporative 19 enrichment with kinetic fractionation (Section 3.1). However, the high Cl⁻ concentration 20 of this sample cannot be explained solely by such enrichment, indicating mixing with 21 high Cl⁻ water. According to Miyazaki et al. (2005), the Cl⁻ concentration of paddy field 22 water ranged from 0.15 to 1.46 mmol/L (5.3 to 51.7 mg/L) at an area adjacent to the 23 present study area. Moreover, the d-excess of paddy field water is generally low because 24 of direct evaporation from water surfaces (Wakui and Yamanaka, 2006). Thus, it is

likely that groundwater at AW7 is affected by paddy field water. Indeed, paddy fields comprise one of the dominant land uses in areas north of AW7. Anaerobic environments below paddy fields may have enhanced denitrification. Although river water at AR4 had similar characteristics, its d-excess was not very low. Therefore, this water may be affected by sewage rather than paddy field water. Consequently, data for AW7 are not suitable for EMMA.

Except for AW7, the groundwater δ values and Cl⁻ concentrations can be used for EMMA assuming three end-members: Watarase River water, local precipitation and mountain block groundwater. The contribution ratio (*R*) from each of the potential sources was estimated using the following equations:

11
$$R_{r} = \frac{(\delta_{g} - \delta_{m})(c_{p} - c_{m}) - (c_{g} - c_{m})(\delta_{p} - \delta_{m})}{(\delta_{r} - \delta_{m})(c_{p} - c_{m}) - (c_{r} - c_{m})(\delta_{p} - \delta_{m})}$$
(1)

12
$$R_{p} = \frac{(\delta_{g} - \delta_{m})(c_{r} - c_{m}) - (c_{g} - c_{m})(\delta_{r} - \delta_{m})}{(\delta_{p} - \delta_{m})(c_{r} - c_{m}) - (c_{p} - c_{m})(\delta_{r} - \delta_{m})}$$
(2)

$$13 R_m = 1 - R_r - R_p (3)$$

14 where δ is the δ^{18} O or δ D; *c* is the CI⁻ concentration; and *g*, *r*, *p* and *m* denote 15 groundwater at each location, Watarase River water, local precipitation and mountain 16 block groundwater, respectively. Although δ_m and c_m cannot be determined exactly, 17 those for groundwater at AW9 were used as approximations. If the plots of groundwater 18 fall outside of the three end-member model, the contribution ratio for the outliers can be 19 estimated by the method described by Liu et al. (2004). Both δ^{18} O vs. Cl and δ D vs. Cl 20 were applied.

21 Considering the uncertain composition/concentration of end-members and the 22 resultant mixtures (i.e., groundwater samples), the standard error of the estimated 23 contribution ratio was computed by error propagation analysis (Phillips and Gregg, 1 2001). The standard deviation of the δ values and Cl⁻ concentration of groundwater 2 (including AW9 as mountain block recharge) and Watarase River water were used as a 3 measure of the uncertainty. The potential error of long-term mean δ values of local 4 precipitation, which may be due to seasonal and inter-annual variations, was estimated 5 based on the standard deviation of δ values for all river water samples. The uncertainty 6 in Cl⁻ concentration for local precipitation and some groundwater samples were 7 assumed to be 0.5 mg/L, which is the mean standard deviation of Cl⁻ concentrations of 8 known groundwater samples.

9 The estimated contribution ratios and their possible errors are shown in Fig. 9. The 10 errors ranged from 3 to 10% for all wells except AW11. The difference in estimated ratios between δ^{18} O and δ D ranged from 0 to 13%. Taken together, these findings 11 12 suggest that the results of EMMA are considerably reliable, with a 10% error range. 13 Although river seepage contribution ratio of 16% is found at AW11, which is located far 14 from the river, the potential error of $\pm 17\%$ at this well is relatively large. As mentioned 15 above, the hydrochemical characteristics of groundwater at AW11 indicate its shallow 16 flow path and relatively short residence time. Therefore, the observed δ values of the 17 water should be affected by the lower δ_p in the study period rather than the long-term 18 mean values used for the EMMA. Thus, this error can be attributed to the difference in the time scale deterring δ_p , and the contribution of the Watarase river water at AW11 19 20 should be nonexistent.

Even with the errors in the estimated contribution ratios (Fig. 9), the general characteristics of their spatial distribution patterns (Fig. 10) do not change very much, indicating that the results obtained are fairly insensitive to errors. These findings suggest that the contribution ratio from the Watarase River water was extremely high

1 (up to 94%) at wells adjacent to the river channel. However, the distribution of 2 contribution ratios of river seepage was asymmetric with respect to the Watarase River 3 channel. As shown in Figure 11, the contribution of river seepage decreased with 4 increasing distance from the river on both the south and north sides. The river recharged 5 systems as far as 5 km to the south (i.e., plain side) and 1.6 km to the north (i.e., 6 mountain side). The steeper slope of the ground near the mountain is expected to 7 produce faster groundwater flow from the mountain to the river and thus reduce 8 groundwater flow from the river to the north. Thus, the distance can be considered an 9 important factor controlling the contribution ratio of the river water to the groundwater, 10 while the relationship between these factors is modified by topographic conditions such 11 as slope.

12 Direct infiltration of local precipitation is greatest in the foothills (AW11 and 13 AW12) and areas far from the Watarase River in the plain (AW5). Even at areas along 14 the Watarase River (AT5, AT7 and AW8), the contribution ratio of direct infiltration is 15 \geq 50%. The absolute value of the recharge rate by local precipitation should not differ 16 greatly among locations; therefore, reducing the local precipitation contribution will 17 indicate the additional recharge by the river and/or mountain block groundwater. The 18 spatial distribution pattern of the contribution ratios obtained from EMMA indicates 19 where such additional recharge occurs and how the recharge waters flow.

The contribution of the mountain block groundwater to aquifers (mountain block recharge) was remarkably high at the central portion of the study area north of the Watarase River (AW9, AW8 and AT5). In contrast, the contribution of the river water at AW8 and AT5 was nonexistent or negligible, despite the proximity to the river. The AW9, AW8 and AT5 were situated around the estimated axis of the syncline and had

1 layers of chert and sandstone (Fig. 2), which suggests that the syncline structure 2 promoted the mountain block recharge at a particular area around the wells while 3 hindering the flow of water from the river channel to the mountain side. These results 4 agree with those of Laronne Ben-Itzhak and Gvirtzman (2005), who found that 5 groundwater could flow from mountain aquifers to rift valley aquifers along synclinal 6 axes. The high Cl⁻ concentration observed at AW9 may reflect dissolution of the rocks 7 or solute enrichment by transpiration (not evaporation), which does not induce isotopic fractionation. Since the Na⁺, Mg^{2+} and HCO_3^- concentrations were also high at AW9, 8 9 dissolution of the rocks may be a major cause of high CI. Although this appears to 10 violate the third assumption of the model, the EMMA results are still valid for tracing 11 the mixing of groundwater in the plain aquifer since such sediment is less easily 12 dissolved.

13 As mentioned above, the major factors controlling the contribution ratio of the 14 river water at each well are the distance from the river, slope, and hydrogeological 15 conditions such as syncline structure. The groundwater pumping rate at each well may 16 be another potential factor influencing the groundwater recharge and flow system (Kelly, 17 2002; Yamanaka et al., 2011b). Although Fig. 12 suggests a positive correlation between 18 pumping rate and the river water contribution ratio for some municipal wells (AT3, AT4, 19 AT6 and AT7), the higher contribution ratios are a reflection of the shorter distance from 20 the river (Fig. 11) rather than the sole effect of the pumping rate. Although groundwater 21 pumping may enhance the absolute rates of total recharge, EMMA did not provide 22 strong evidence of its effect on river channel seepage.

Based on the above results, the main recharge source for municipal wells (except
AT5) is the Watarase River. The contribution ratios of river water generally exceed 50%.

1 The greatest contribution ratio for municipal wells was 94%, which was observed at the 2 well with the largest pumping rate (AT3 in Fig. 12). Therefore, the domestic water 3 supply system in this area depends strongly on the quantity and quality conditions of the 4 Watarase River. For example, if upstream river water is regulated by dams or 5 contaminated by accidents, water from municipal wells may be affected, while 6 hydraulic drawdown due to pumping at the wells can be mitigated by river seepage. 7 Conversely, even though AT5 is located very close to the river channel and has high 8 pumping rate, the contribution of Watarase River water to this aquifer is suppressed by 9 mountain block recharge. Thus, this well is not strongly influenced by Watarase River 10 conditions. Furthermore, at direct-recharge-dominated sites, shallow groundwater 11 should be more sensitive to artificial alteration of land use and cover. These findings 12 should be helpful for integrated management of groundwater and surface water 13 resources.

14

15 **4. Conclusions**

Isotope and hydrochemistry analysis indicated that direct infiltration by precipitation, Watarase River seepage and mountain block recharge are the three main recharge sources in the study area. The isotopic signatures revealed an obvious altitude effect, as well as mean residence times of river water and groundwater > 1 year.

EMMA using stable isotope and chloride tracers was shown to be useful for estimation of the contribution ratio of different recharge sources with sufficiently low error ($\leq 10\%$). The results demonstrated that the Watarase River contributes a great amount of water to aquifers along its channel, with the contribution ratio reaching as high as 94%. However, the spatial extent of river-recharged water differs, with the water 1 influencing aquifers for 5 km from the channel and 1.6 km on the plain side and 2 mountain side, respectively. Mountain block recharge is one of the dominant 3 components in a limited area in the foothills. A syncline structure was found to play an 4 important role in transmitting mountain block recharge, and mountain block recharge 5 was shown to suppress river channel seepage into the mountain side aquifers. The 6 distance from the river channel, topography and hydrogeological settings such as 7 syncline were identified as the three major factors controlling the river water 8 contribution ratio.

9 The major findings of this study will help improve the understanding of recharge 10 processes and dynamics and facilitate integrated management of surface and subsurface 11 water resources. This paper highlights the spatial structure of the groundwater-river 12 interactions and the influence of syncline structure settings on mountain block recharge. 13 To further increase the reliability of EMMA results, combined use of the other 14 approaches such as numerical simulation is necessary.

15

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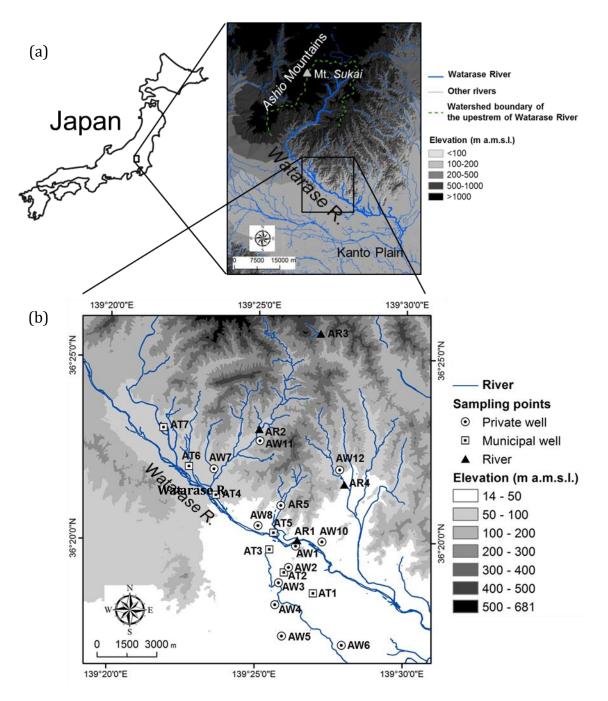
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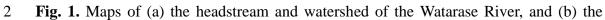
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17	







3 study area and sampling points.

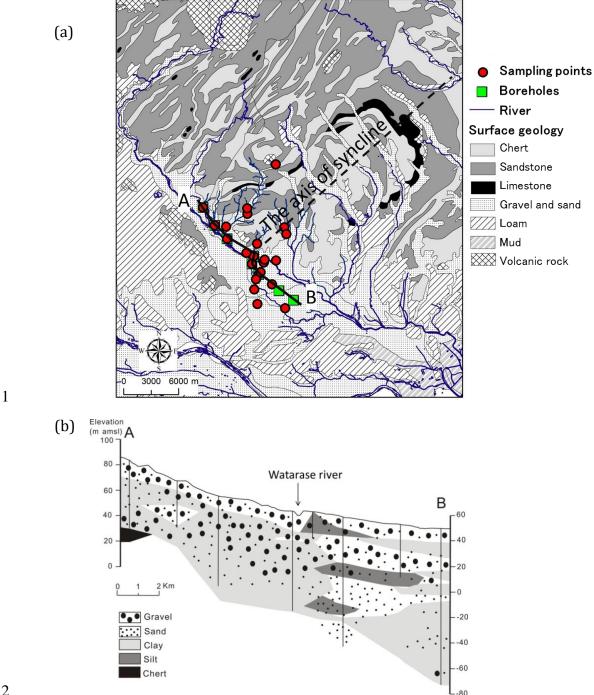
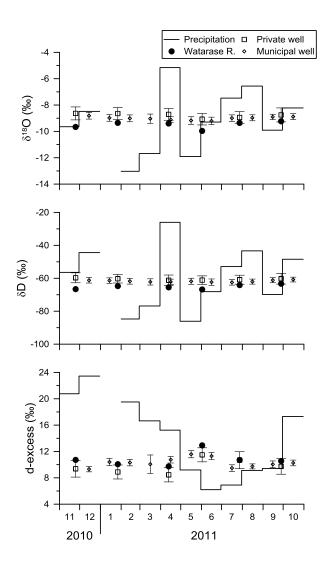




Fig. 2. Surficial geology map (a) of the study area (National Land Survey Division, Land and Water Bureau, Japan; http://tochi.mlit.go.jp/tockok/) with (b) cross-section A-B showing lithology of aquifers (drawn based on borehole logs).





4 Fig. 3. Temporal variations of δ^{18} O, δ D and d-excess in precipitation, Watarase river

5 water and groundwater during the study period.

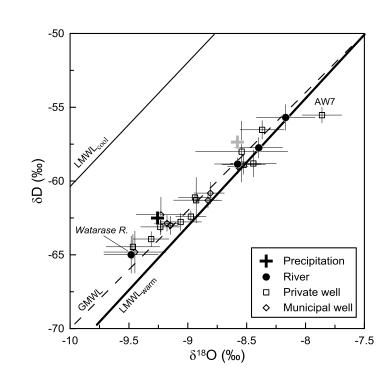


Fig. 4. Plot of δ¹⁸O vs. δD of river and groundwater samples. The global meteoric water
line and local meteoric water lines (in cool season and warm season) are added. Black
cross is weighted (by monthly precipitation amount) mean precipitation and gray cross
represent mean precipitation weighted by long-term mean precipitation amount.

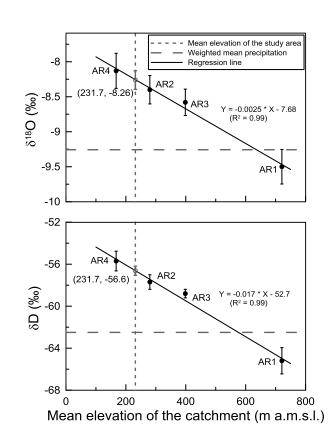


Fig. 5. Mean δ¹⁸O and δD of river water vs. mean elevation of the catchment. The gray
open circles represent estimated long-term mean δ values of precipitation corresponding
to the mean elevation of the study area and the gray error bars are estimated standard
deviation for long-term mean δ values of precipitation.

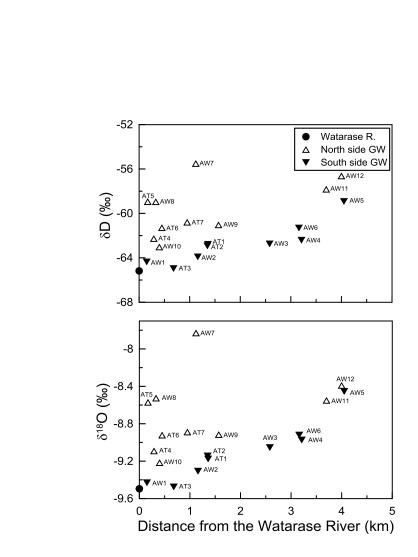


Fig. 6. Annual mean δ^{18} O and δ D of groundwater and their relationship with distance from the Watarase River. The annual mean $\delta^{18}O$ and δD of Watarase River water was

also plotted for reference.

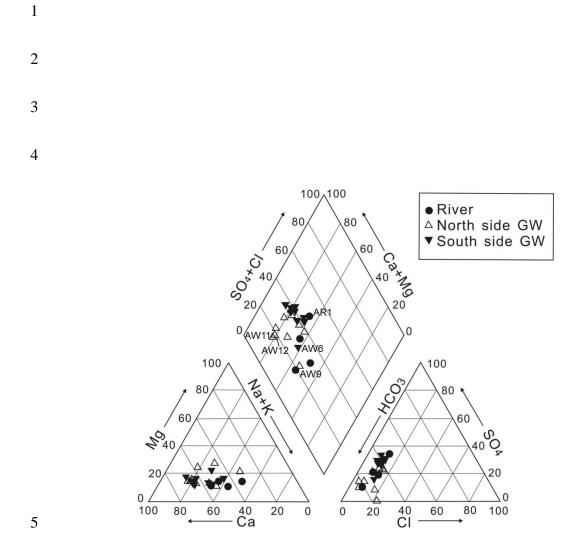


Fig. 7. Piper diagram of groundwater and river water samples in the study area.



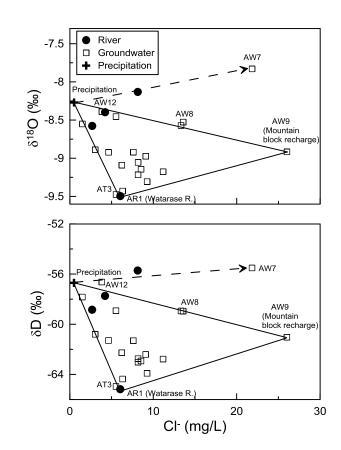
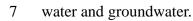
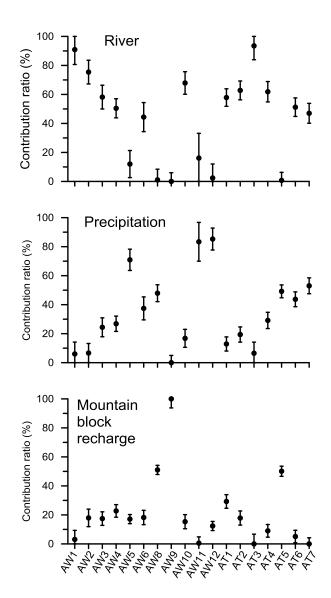




Fig. 8. Relationship of arithmetic mean annual δ^{18} O, δ D and Cl⁻ concentration in river







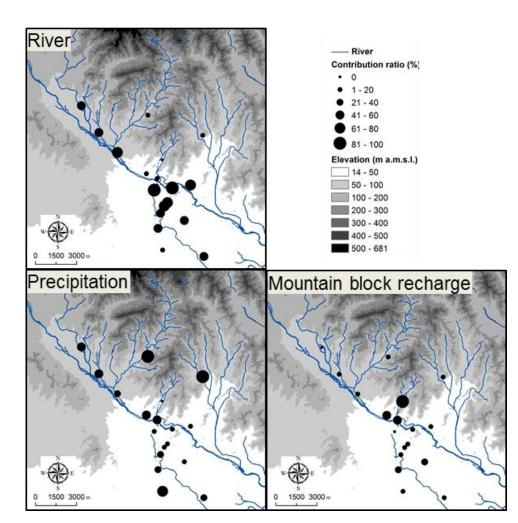
4 Fig. 9 The contribution ratios and corresponding estimation errors of recharge sources

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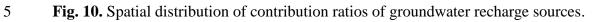
⁵ for wells.

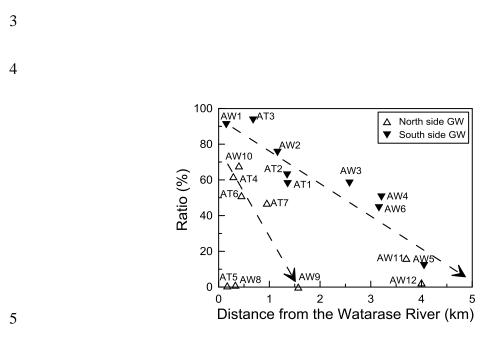


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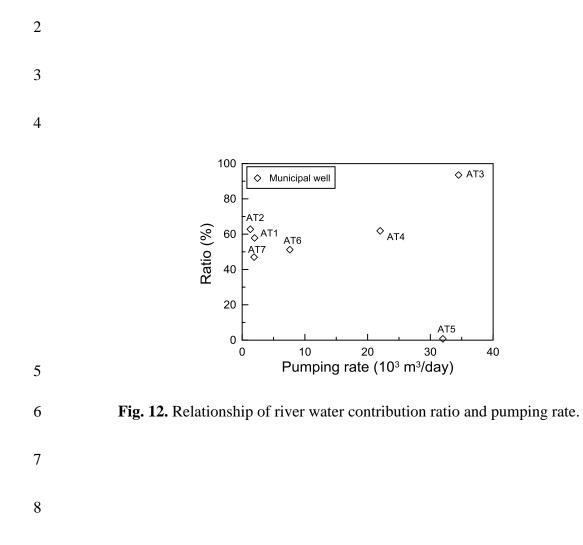






6 Fig. 11. Relationship of river water contribution ratio and distance from the Watarase

7 River.



	Hydrochemistry										Iso	Isotope	
ID	EC	pН	Na^+	\mathbf{K}^+	Ca ²⁺	Mg^{2+}	Cl	SO4 ²⁻	HCO ₃ ⁻	NO ₃ ⁻	$\delta^{18}O$	δD	
	ms/m		mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	mg/L	‰	‰	
AW1	17.1	6.40	13.5	1.6	20.1	2.9	6.3	23.7	55.5	9.3	-9.43	-64.4	
AW2	22.8	6.52	20.1	2.1	32.3	4.9	9.3	29.4	71.1	13.4	-9.31	-63.9	
AW3	29.4	6.62	16.4	2.3	42.2	6.8	8.2	45.7	101.5	14.6	-9.05	-62.7	
AW4	27.3	6.51	24.0	2.9	37.1	10.4	9.1	36.7	101.2	2.7	-8.97	-62.4	
AW5	19.9	6.29	11.5	2.8	32.5	3.9	5.5	28.4	76.9	3.3	-8.45	-58.9	
AW6	16.8	7.53	17.5	1.5	18.5	4.4	7.6	13.1	73.6	4.7	-8.92	-61.3	
AW7	44.4	6.73	23.9	1.7	34.5	13.5	21.8	18.1	168.9	0.0	-7.83	-55.5	
AW8	26.0	6.35	20.4	1.9	32.7	5.2	13.6	27.9	94.7	13.9	-8.53	-59.(
AW9	36.6	6.66	43.9	2.3	27.2	11.7	26.0	2.8	156.7	0.0	-8.92	-61.0	
AW10	21.2	6.36	19.7	2.7	25.2	3.4	8.2	25.5	74.2	10.8	-9.22	-63.0	
AW11	11.9	6.11	5.9	0.9	18.3	2.6	1.5	8.9	57.6	6.9	-8.55	-57.8	
AW12	15.3	6.46	8.6	2.6	23.9	6.6	3.8	10.2	94.7	9.6	-8.39	-56.6	
AT1			11.4	2.1	34.6	4.9	11.2	33.3	98.7	16.7	-9.18	-62.8	
AT2			8.3	1.6	33.4	5.3	8.5	37.3	106.7	15.3	-9.14	-62.9	
AT3			9.6	2.3	24.3	2.8	5.5	18.0	48.4	7.6	-9.48	-65.0	
AT4			9.3	1.6	23.9	3.3	6.2	18.5	58.2	8.3	-9.09	-62.3	
AT5			15.3	2.5	44.5	6.8	13.3	33.7	107.9	14.3	-8.57	-58.9	
AT6			7.1	1.7	23.1	3.4	4.6	15.9	61.9	9.9	-8.92	-61.3	
AT7			7.0	1.5	25.3	3.6	3.0	10.2	63.8	5.5	-8.89	-60.8	
AR1	12.6	7.37	15.6	1.7	22.3	2.6	6.0	19.5	38.6	6.0	-9.50	-65.2	
AR2	10.6	6.86	18.9	1.0	11.3	2.5	4.2	10.5	47.2	4.8	-8.40	-57.7	
AR3	8.3	7.10	12.5	0.8	11.2	1.4	2.7	3.8	41.4	7.8	-8.58	-58.8	
AR4	18.1	7.19	16.8	2.1	20.0	3.3	8.1	14.3	65.6	7.3	-8.13	-55.7	

Table 1 The analyzed hydrochemical and isotopic items of groundwater and river

2 samples

1