Difluorocarbene-Based [4 + 1] Cycloaddition Strategy for Synthesizing Derivatives of 5-Fluorinated Thiazoles and Oxazoles

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1 When treated with difluorocarbene, which was
2 generated from $FSO_2CF_2CO_2SiMe_3$ with a 1,8-2 generated from FSO₂CF₂CO₂SiMe₃ with a 1,8-
3 bis(dimethylamino)naphthalene catalyst, *N*-
4 (thioacyl)amidines underwent [4 + 1] cycloaddition to afford 5 the corresponding amino-substituted 5,5-difluorothiazolines. 6 Both dehydrofluorination and a Hofmann elimination/S_N2[']-
6 Both dehydrofluorination and a Hofmann elimination/S_N2^{'-}
7 type reaction sequence enabled the aromatization of the
8 obtained products, affording 5-fluo 8 obtained products, affording 5-fluorothiazoles. The $[4 + 1]$
9 cycloaddition strategy was also applied to *N*-acylamidines. 10 affording the corresponding 5-fluorooxazole derivatives. $\frac{11}{12}$ Keywords: Difluorocarbene, Fluorine, Heterocycles

 Thiazoles and oxazoles are found in the core structures of various beneficial compounds. For example, firefly 15 luciferin¹ and aztreonam² possess the thiazole ring and act as luminescent compounds and antibiotics, respectively (Figure 1). Retosiban is an oxytocin receptor antagonist that exhibits

18 the oxazole substructure.³

Firefly luciferin

19

Aztreonam (antibiotics)

20 **Figure 1.** Beneficial thiazole derivatives and oxazole derivatives.

22 Fluorine is frequently introduced into organic molecules
23 to improve or change their biological activities and spectra.⁴ 23 to improve or change their biological activities and spectra.⁴ 24 Thus, the synthesis of fluorinated thiazoles and oxazoles may 25 aid in the development of innovative and high-performance
26 medicines and agrochemicals. However, the direct medicines and agrochemicals. However, the direct 27 introduction of a fluorine atom onto such heterocyclic rings 28 is problematic due to low yields and a lack of 29 regioselectivity.^{5,6} regioselectivity.^{5,6}

30 We established an organocatalyzed process for 31 difluorocarbene generation and its applications, 7 some of 32 which undergo three-membered thiirane ring formation $(2 +$ 33 1] cycloaddition) using difluorothiocarbonyl ylide intermediates.⁸ To overcome the drawbacks of fluorine 35 introduction into five-membered thiazole and oxazole rings, 36 we adopted a $[4 + 1]$ cycloaddition strategy using π -extended 37 (thio)carbonyl ylides (Scheme 1). $9,10$ Thus, on treatment with 38 difluorocarbene, *N*-(thio)acylamidines 1 ($X = S$) and 2 ($X =$

39 O) would form difluoro(thio)carbonyl ylides.¹¹ Nucleophilic 40 five-membered ring closure of the vlide intermediates $(5-$ 40 five-membered ring closure of the ylide intermediates (5-
41 *endo-trig* cyclization) may yield 5.5-difluorothiazolines 3 41 *endo*-*trig* cyclization) may yield 5,5-difluorothiazolines **3** 42 and 5,5-difluorooxazolines 4,¹² whose aromatization would 43 form 5-fluorothiazoles **5** and 5-fluorooxazoles **6**. Notably, the 44 pendant $NMe₂$ group may aid the five-membered ring 45 formation by resonance (5-*exo-trig* cyclization) and serve as 45 formation by resonance (5-*exo*-*trig* cyclization) and serve as 46 a functional group to allow aromatization.¹³

48 **Scheme 1.** $[4 + 1]$ Cycloaddition strategy to afford 5-fluoro-
49 thiazole and 5-fluorooxazole derivatives. thiazole and 5-fluorooxazole derivatives.

50 The required precursors were prepared according to the 51 literature (Table 1).¹⁴ Commercially available thioamides or 51 literature (Table 1).¹⁴ Commercially available thioamides or 52 amides were condensed with dimethyl acetals of *N*,*N*-53 dimethylformamide (Entries 1–4 and 8) or *N*,*N*dimethylacetamide (Entries 5–7). The required *N*-55 (thioacyl)amidines **1** and *N*-acylamidines **2** were afforded in 56 69%–quantitative yields.¹⁵

57 Using **1e** as the model substrate, the reaction conditions 58 were tuned (Table 2). A diluted solution of **1e** (0.05 M) was
59 treated with 2 equivalents of trimethylsily 2.2-difluoro-2treated with 2 equivalents of trimethylsilyl 2,2-difluoro-2-60 (fluorosulfonyl)acetate $(TFDA)^{16}$ in the presence of a 1,8-61 bis(dimethylamino)naphthalene (proton sponge) catalyst (5 mol%).7b 62 The expected adduct **3e** was afforded in all the 63 studied common solvents such as hexane (74% yield 64 determined by ¹⁹F NMR analysis, Entry 1) and toluene (86% 65 yield, Entry 2). However, the reaction showed poor 66 reproducibility when conducted in toluene. The product 67 gradually decreased in CCl_4 (Entry 3) and 1,2-dichloroethane 68 (Entry 4), whereas **3e** was produced slowly but in good yields 69 in nitromethane (Entry 5). Finally, tetrahydrofuran was 70 selected as a solvent because it afforded 5,5- 71 difluorothiazoline **3e** by treating the **1e** solution (0.2 M, Entry 72 8) with 1.05 equivalents of TFDA.¹⁷

1 **Table 1.** Preparation of *N*-(thioacyl)amidines and *N*-2 acylamidines

	avynamannos					
	х NH ₂ R ¹	$\ddot{}$	R^2 MeO NMe ₂ MeO	-2 MeOH CH ₂ Cl ₂ , RT		R^2 х NMe ₂ N R ¹
3	$(X = S, O)$		1.1 equiv			1 $(X = S)$ $2(X = 0)$
	Entry	X	R^1	R^2	t(h)	Yield $(\%)$ ^{<i>a</i>}
	1	S	Ph	H	3	1a 75
	2	S	C_6H_4p -Cl	Η	3	$1b$ 96
	3	S	C_6H_4p -CF ₃	Н	16	1c 84
	4	S	C_6H_4p -Me	Η	6	1d 85
	5	S	Ph	Me	1	1e quant
	6	S	C_6H_4p -Cl	Me	19	$1f$ 69
	7	S	C_6H_4p -Me	Me	15	$1g$ 69
	8	O	C_6H_4p -OMe	Н	23	2a quant

^a 4 Isolated yield.

S

Me

 Several (thioacyl)amidines **1** yielded the corresponding difluorothiazolines **3** under optimized conditions (Table 3). Benzthioamide-derived **1a** reacted with in situ-generated difluorocarbene to form **3a** in a 90% yield (Entry 1). Chlorinated, trifluoromethylated, and methylated difluorothiazolines **3b**–**d** were afforded in 72%–87% yields via the reactions of electron-deficient and -rich substrates **1b**–**d**, respectively (Entries 2–4). The obtained products **3** were separated via Kügelrohr distillation. Moreover, **3e**–**g** 14 with a methyl group on the ring (R^2) were synthesized from **1 e** – **g** , respectively, in 86 – 9 6 % y i e l d s

> FSO₂CF₂CO₂SiMe₃ (TFDA) 5 mol% Proton Sponge

S

F F

16 **Table 2.** Optimization of reaction conditions

17 Entry Solvent $c(M)$ ^{*a*} TFDA</sup> (equiv) *t* (min) Yield (%) *b,c* 1 Hexane 0.05 2.0 30 74 2 Toluene *^d* 0.05 2.0 30 86 *^e* 3 CCl4 0.05 2.0 5 58 4 DCE 0.05 2.0 15 61 5 MeNO₂ 0.05 2.0 30 84 6 THF 0.05 2.0 15 quant 7 THF 0.2 2.0 15 63 8 THF 0.2 1.05 15 quant N 60 °C **1e 3e** Ph NM_e $\mathsf{P}\mathsf{h}\mathop{\curvearrowright}\limits_{\mathsf{N}}$ $NMe₂$ Me

18 *a* Concentration of 1e. *b* ¹⁹F NMR yield based on an internal

19 standard $(CF_3)_2Cp$ -Tol₂. *c* Substrate **1e** was consumed in all of

20 the entries. *d* Reflux. *e* Poorly reproduced. Proton sponge = 1,8-

21 Bis(dimethylamino)naphthalene. DCE = 1,2-Dichloroethane.

22 (Entries 5–7). The oxygen analog **2a** afforded the 23 corresponding 5,5-difluorooxazoline **4a** in a 63% yield 24 (determined by 19 F NMR analysis, Entry 8).

25 **Table 3.** Synthesis of 5,5-difluorothiazolines 26 and 5,5-difluorooxazolines

	R^2 х NMe ₂ R^1 1 or $2(0.2 M)$		TFDA (1.05 equiv) 5 mol% Proton Sponge		F X R^2 R^1 N NMe ₂	
			THF, 60 °C, 15 min			
27					3 or 4	
	Entry	1 or 2	R ¹	R ²	Yield $(\%)$ a,b	
	1	1a	Ph	H	3a 90	
	2	1b	C_6H_4p -Cl	H	3b 87	
	3	1c	C_6H_4p -CF ₃	H	3c 75	
	4	1d	C_6H_4p -CH ₃	H	3d 72 (82)	
	5	1e	Ph	Me	3e 96	
	6	1f	C_6H_4p -Cl	Me	3f 86	
	7	1g	C_6H_4p -Me	Me	3g86	
	8	2a	C_6H_4p -OMe	Н	4a (63)	

28 *a* Isolated yield $[19F] NMR$ yield based on an internal standard 29 *(CFx)Cn*-Tol₂ is shown in parentheses]. ^{*b*} The product was $(CF_3)_2Cp$ -Tol₂ is shown in parentheses]. ^{*b*} The product was 30 isolated via Kügelrohr distillation [115–120 °C (bath temp.)/0.8 31 mmHg].

 Furthermore, *N*-(thioacyl)amidines derived from aliphatic thioamides were less stable than those derived from aromatic thioamides. Thus, in situ-generated isopropyl precursor **1h** was treated with difluorocarbene to afford the corresponding isopropyl-substituted 5,5-difluorothiazoline **3h** in a 52% yield (determined by ¹⁹F NMR analysis, eq 1).

 The resulting products **3** thus obtained were then converted to 5-fluorothiazoles **5**. First, **3a** was treated with 4 equivalents of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) in toluene at 100 °C. Dehydrofluorination afforded the 43 aromatized 5a in an 80% yield (determined by ¹⁹F NMR 44 analysis, Scheme 2).¹⁸

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45 The quaternization of the dimethylamino group 46 facilitated the synthesis of aromatized 5-fluorothiazoles **5**, 47 where elimination followed by the S_N2' -type reaction with 48 nucleophiles produced the additional C–N or C–C bond (Scheme 3). 49 ¹⁹ When treated with dimethyl sulfate, **3e** with a 50 methyl group on the carbon α to the dimethylamino group 51 yielded the trimethylammonium salt (not shown), whose 3

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2 **Scheme 2.** Aromatization of 5,5-difluorothiazolines via 3 dehydrofluorination $(^{19}F$ NMR yield based on an internal 4 standard $PhCF_3$).

5 Hofmann elimination produced 4-methylidene-5,5- 6 difluorothiazoline **7e**. The obtained exo-methylene moiety of 7 **7e** was reactive toward nucleophiles such as pyrrolidine and

8 trimethylaluminium, affording 2,4-disubstituted 5-

- 9 fluorothiazole products **5b** (with pyrrolidine) and **5c** (with
- 10 AlMe3) in 84% and 73% yields, respectively.

12 **Scheme 3.** Aromatization of 5,5-difluorothiazolines via a 13 Hofmann elimination/SN2′-type reaction sequence (Isolated 14 yield).

 In conclusion, convenient access to 5-fluorothiazole and 5-fluorooxazole derivatives was devised using a difluorocarbene-based [4 + 1] cycloaddition strategy. Treating *N*-(thioacyl)amidines or *N*-acylamidines with difluorocarbene generated from $FSO_2CF_2CO_2SiMe_3$ with a proton sponge catalyst allowed cycloaddition to yield the 21 corresponding 5,5-difluorothiazolines and 5,5-
22 difluorooxazolines with an amino group. Both difluorooxazolines with an amino group. Both 23 dehydrofluorination and a Hofmann elimination/ S_N 2'-type reaction sequence successfully afforded the corresponding aromatized 5-fluorothiazoles.

 $rac{26}{27}$ Supporting Information is available on 28 http://dx.doi.org/10.1246/cl.******. 29

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