

1 **Effect of hydrogel particle mechanical properties**
2 **on their disintegration behavior using a gastric digestion simulator**

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20 **Abstract**

21 The interest in designing novel foods whose digestibility can be controlled based on life stage
22 and health conditions continues to grow. Physical digestion is important for solid foods as their
23 breakdown and resulting size reduction can promote enzymatic reactions. Our human gastric
24 digestion simulator (GDS) enables the simulation and direct observation of food particle
25 disintegration induced by simulated antrum contraction waves. The objectives of this study were
26 to verify the disintegration performance of the GDS compared with previously reported *in vivo*
27 data and evaluate the effects of the mechanical properties of hydrogel particles on their *in vitro*
28 gastric disintegration behavior. Agar beads with four fracture forces were prepared and mixed
29 with meal containing locust bean gum to adjust viscosity same as their *in vivo* data. The half
30 residence time of intact beads was longer for hard agar beads than for soft agar beads, and a
31 similar disintegration trend to *in vivo* data was obtained. Moreover, as solid food models, 5-mm
32 hydrogel cubes with different fracture stresses and fracture strains were prepared by varying the
33 agar and native type gellan gum concentrations. The hydrogel cubes disintegrated because of
34 fracture and abrasion during *in vitro* gastric digestion in the presence of simulated antrum
35 contraction waves. The degree of hydrogel cube disintegration was affected by their fracture
36 strain rather than their fracture stress and was suppressed when their fracture strain was greater
37 than 30%. Our findings may provide a better understanding of the gastric digestion behavior of
38 solid foods with different mechanical properties.

39

40 **Keywords:** *In vitro* gastric digestion, Hydrogel, Gastric digestion simulator, Antral contraction
41 waves, Mechanical properties, Disintegration behavior

42

43 1. Introduction

44 The stomach plays an important role in the digestion of foods in the human digestive tract.
45 The main functions of the stomach include storage, mixing, disintegration, and emptying. Solid
46 food is mechanically broken down by chewing, roughly reducing its size to <5.0 mm (Jalabert-
47 Malbos et al., 2007). The bolus sent from the esophagus to the stomach is then temporarily
48 stored in the stomach for less than 3 h (Camilleri et al., 1985; Gardner, Ciociola, & Robinson,
49 2002). The gastric content comprising food particles, digestive fluids, and digestive enzymes is
50 mixed in the presence of peristaltic motion on the gastric wall. The food particles in the gastric
51 content also disintegrate because of physical movements (antral contraction waves, ACWs) and
52 chemical reactions (digestive enzymes, pH). Because of the gastric disintegration process, most
53 of the digesta with particle diameters less than approximately 2 mm is emptied from the antrum
54 of the stomach (Kelly, 1980; Guo et al., 2014). Investigating the disintegration behavior of solid
55 foods during gastric digestion is a key factor in controlling digestibility and the delivery of the
56 nutrients embedded within foods.

57 There is an increasing demand for food products whose texture is appropriately designed for
58 elderly, obese, and functional dyspepsia patients. The mechanical properties of solid foods, such
59 as hardness and elasticity, are important parameters for controlling food digestibility in the
60 above-mentioned people. The mechanical properties of hydrogels can be readily varied by
61 adjusting the formulation and/or concentration of the gelling agents (e.g., polysaccharides and
62 proteins). Hydrogels are also commonly used as solid food models in oral food processing
63 research. For example, Ishihara et al. (2014) found that the first size reduction of gellan
64 hydrogels was similar for instrumental compression tests using artificial tongue and *in vivo*
65 human tests. Kohyama et al. (2016) also identified that the mechanical properties of different

66 types of hydrogels had a strong influence on natural eating behaviors during oral processing in
67 humans. However, the effects of the mechanical properties of hydrogels on their disintegration
68 during gastric digestion remain unclear.

69 Numerous *in vitro* and *in vivo* studies on the gastric digestion of solid foods have been
70 reported over the past two decades (Kong & Singh, 2008; Dupont et al., 2018). The most
71 common *in vivo* method uses magnetic resonance imaging (MRI), which allows rapid
72 measurements of multiple parameters of gastric function in a single scan (Hoad et al., 2015).
73 This *in vivo* method is ideal for studying the gastric digestion of solid foods but has drawbacks
74 such as ethical constraints and in some cases being a burden on subjects. Different *in vitro*
75 digestion models mimicking the gastric digestion process have been proposed as alternatives to
76 *in vivo* methods. A conventional *in vitro* digestion model involves shaking tubes or flasks to mix
77 food particles with artificial digestive fluids containing digestive enzyme(s) (McClements & Li,
78 2010). However, this model does not evaluate the disintegration of food particles appropriately
79 because ACWs are absent.

80 *In vitro* dynamic models that can consider ACWs have been developed since the mid-1990s
81 (Guerra et al., 2012; Dupont et al., 2018). The TNO Gastro-Intestinal Model-1 (TIM-1),
82 developed by Minekus et al. (1995), allows contraction movements of the soft, flexible gastric
83 vessel walls driven by periodically controlled hydrostatic pressure outside the walls. The
84 contraction movement enhances the mixing of the gastric content. The Dynamic Gastric Model
85 (DGM) mechanically processes gastric content through the movement of a piston and barrel
86 simulating the rhythmic ACWs of the human stomach (Vardakou et al., 2011). However, these
87 dynamic digestion models can be expensive for daily use in the food industry. Chen et al. (2016)
88 developed a ‘Rope-Driven’ *in vitro* Human Stomach Model (RD-IV-HSM), with the aim of

89 investigating the effects of gastric morphology on digestion behavior. The RD-IV-HSM modeled
90 the whole gastric morphology using a liquid silicone molding process, and the contraction
91 movements by fastening/relaxing ropes wrapped around the antrum of the modeled stomach. The
92 RD-IV-HSM has reproduced the size distribution of a semi-solid meal during the digestion
93 process; however, it was not effective in breaking down larger food particles into the smaller
94 sizes required for gastric emptying ($< \sim 2$ mm). An advanced dynamic *in vitro* human stomach
95 (new DIVHS) system based on the RD-IV-HSM has been developed (Wang et al., 2019). The
96 human gastric simulator (HGS) mimics the ACWs using mechanically operated rollers; however,
97 the ACW-induced motion of the gastric contents cannot be directly observed (Kong & Singh,
98 2008; Dupont et al., 2018). Recently, *in vitro* stomach digestion devices based on a similar
99 concept have also been proposed (Barros et al., 2016; Liu et al., 2019).

100 Our group has developed an *in vitro* model named the gastric digestion simulator (GDS) that
101 simplifies the major features of the stomach including gastric peristalsis, which mainly
102 progresses in the antrum (distal stomach), and allows operation of quantitatively simulated
103 ACWs and real-time observation of digestion behavior (Kozu et al., 2014). To study physical
104 gastric digestion, Kozu et al. (2015) performed GDS and flask-shaking experiments using agar
105 cubes as a solid food model. It was reported that agar cubes were only broken down in the GDS
106 experiments, which suggests that simulated ACWs contribute to the disintegration of solid foods.
107 However, quantitative evaluations of the physical forces generated by simulated ACWs and the
108 effect of the mechanical properties of solid foods on the disintegration of food particles remain
109 lacking.

110 *In vivo* studies focusing on the contraction force and the force experienced by the target solid
111 food particles during gastric digestion have been reported. Vassallo et al. (1992) measured the

112 force generated by ACWs directly using a reaction force catheter. Marciani et al. (2001)
113 observed the degree of gastric disintegration in subjects who ingested agar beads with several
114 different fracture forces using MRI. Kamba et al. (2000) analyzed the absorption of a marker drug
115 in subjects who ingested press-coated Teflon tablets with several different fracture forces.
116 However, the data obtained from these *in vivo* studies varied widely. We believe that the result
117 reported by Marciani et al. (2001) is the most useful because it provided direct observation of
118 food disintegration in the stomach.

119 To verify the disintegration performance of the GDS it is necessary to compare the *in vitro*
120 data obtained from GDS experiments with the above-mentioned *in vivo* data. Additionally, the
121 quantitative impacts of the mechanical properties of solid foods on the disintegration mechanism
122 remain unclear. The first objective of this study was to validate the GDS device for reproducing
123 human gastric disintegration of solid foods using similar food samples (agar beads with a range
124 of fracture forces 0.53–0.90 N in LBG meals) against the *in vivo* data. The second objective was
125 to evaluate the effect of the mechanical properties of hydrogel particles on their disintegration
126 behavior caused by the simulated ACWs of the GDS using 5×5×5 mm hydrogel cubes
127 containing agar or a mixture of agar and native type gellan gum as a model solid food.

128

129 **2. Materials and methods**

130 *2.1. Gastric digestion simulator (GDS)*

131 The GDS used for this study (Kozu et al., 2014) was equipped with a vessel that models the
132 antrum and rollers that generate ACWs, which provide mechanical forces on the gastric contents
133 (Fig. 1a). The speed (2.5 mm/s) and generation frequency (1.5 cycle/min) of the ACWs that act
134 on the sidewalls of the GDS vessel were controlled based on literature data for the ACWs of

135 healthy adults (Sun et al., 1995). The standard values of the ACWs obtained from *in vivo* studies
136 were 1.5–5.0 mm/s and 1–3 cycles/min (Pal et al., 2004; Marciani et al., 2001; Ajaj et al., 2004;
137 Sun et al., 1995). A temperature control unit maintained the temperature inside and around the
138 GDS vessel at normal human body temperature (~ 37 °C). As shown in Fig. 1b, each roller
139 contains two foam rubber layers with a 12.5-mm thick exterior layer made of ethylene propylene
140 rubber foam (E-4070) and a 2.5-mm-thick interior layer made of polyurethane rubber foam
141 (SM55;) (INOAC CORPORATION, Tokyo, Japan).

142 The contraction force generated by the GDS rollers was measured using manometry, which
143 was conducted using a digital manometer (testo 510, Testo Co., Ltd., Osaka, Japan; Fig. 1c). The
144 maximum pressure (P_{\max}) generated in a 26-mm-diameter silicone balloon was measured by
145 placing the manometer at a position where the occluded clearance in the GDS vessel was a
146 minimum. The balloon was also compressed using a texture profile unit (TPU-2C, Yamaden Co.,
147 Ltd., Tokyo, Japan) equipped with a 40-mm-diameter flat cylindrical probe at a deformation
148 speed of 2.5 mm/s. When the balloon was gradually compressed, the maximum force (F_{\max})
149 applied to the balloon and P_{\max} in the balloon were recorded to analyze the correlation between
150 the values. The F_{\max} value was used to express the maximum contraction force generated by the
151 motion of the rollers.

152 The correlation between the contraction force generated in the GDS vessel and P_{\max} was
153 analyzed using the texture profile unit and manometry method (Fig. S1). The contraction force
154 was estimated to be 8.5 ± 0.1 N ($n = 5$) when the minimum clearance between a pair of rollers
155 was 11.2 ± 0.1 mm ($n = 10$). The estimated contraction force was converted to mechanical stress
156 for comparison with the *in vivo* data reported in previous research. The calculated mechanical
157 stress ranged from 16.0 to 86.3 kPa. Marciani et al. (2001) reported a fracture force of 0.65 N for

158 12.7-mm-diameter agar beads in the human stomach. Kamba et al. (2000) reported a fracture
159 force of 1.89 N for Teflon-coated tablets (7 mm long and 4 mm wide) containing a marker drug
160 that was released only when the tablets received a force greater than its fracture force. These
161 fracture force values obtained from *in vivo* experiments correspond to a range of mechanical
162 stress of 5.1–67.5 kPa. The contraction force value generated in the GDS vessel was therefore
163 compared with these *in vivo* data.

164 2.2. Composition of simulated digestive fluids

165 α -Amylase from *Bacillus subtilis* (#10070) (59.3 U/mg) and pepsin from porcine gastric
166 mucosa (#P7000) (714 U/mg) were purchased from Sigma-Aldrich, Inc. (St. Louis, MO, USA).
167 All salts and chemicals used for preparing simulated saliva fluid (SSF) and simulated gastric
168 fluid (SGF) (Table 1) were purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan).
169 The compositions of SSF and SGF shown in Table 1 were adopted with slight modification of
170 the SSF and SGF proposed by Minekus et al. (2014). The pH of the SGF used in this study
171 (Table 1) was based on the literature data for the United States Pharmacopeia (USP) dissolution
172 apparatus II (USP 26, 2003), which employs a solution with a pH close to that of the gastric juice
173 secreted in human stomach.

174 2.3. Comparison of *in vitro* gastric digestion using GDS and *in vivo* human gastric digestion

175 To investigate whether the GDS simulates the disintegration environment of the human
176 stomach, we prepared spherical agar beads whose composition and size were the same as those
177 used for *in vivo* digestion in the human stomach (Marciani et al., 2001) and the bead
178 disintegration patterns for the two experiments were compared. Agar powder (#010-15815) was
179 purchased from Wako Pure Chemical Industries, Ltd. Several agar beads with different agar

180 concentrations in the range 1.5–3.0 wt% were prepared. The hot agar hydrosol injected into an
181 acrylic template was slowly cooled for 2 h at 8 °C. Locust bean gum (LBG) (#G0753) purchased
182 from Sigma-Aldrich, Inc. was added to the meal so that the viscosity of the meal was matched to
183 that used for the *in vivo* experiments conducted by Marciani et al., 2001. The LBG meal was
184 prepared by dispersing 10.5 g of LBG powder in 1 L of Milli-Q water with vigorous stirring
185 overnight. The viscosity of the prepared LBG meal was 0.06 Pa·s, which is similar to the value
186 previously reported for the *in vivo* experiments (Marciani et al., 2001). The oral phase was not
187 considered because all the agar gel beads ingested by subjects without chewing were intact after
188 swallowing based on the experimental procedure of this *in vivo* human study. Because the
189 capacity of the GDS vessel is approximately 550 mL, our GDS experiments were performed
190 using 10 agar beads, 100 mL of LBG meal, and 330 mL of SGF for 150 min at 37 °C. The
191 number of agar beads that remained intact (N) was counted every 10 min.

192 *2.4. Effect of hydrogel particle mechanical properties on their disintegration in the GDS*

193 *2.4.1. Preparation of hydrogel samples*

194 Agar powder (#010-15815) was purchased from Wako Pure Chemical Industries, Ltd. A
195 native-type gellan gum was kindly provided by San-Ei Gen FFI, Inc. (Osaka, Japan). Hydrogels
196 were prepared by dissolution of different concentrations of agar and native-type gellan gum in
197 Milli-Q water using a magnetic stirrer for 30 min at 90 °C, and subsequent cooling of the
198 hydrosol to 8 °C over 2 h. The concentrations of the gelling agents are presented in Table 2.

199 *2.4.2. Measurement of hydrogel mechanical properties*

200 The mechanical properties (fracture stress and fracture strain) of the hydrogel samples were
201 measured using a texture profile unit equipped with a flat cylindrical probe (16 mm diameter).

202 Hydrogel samples cut into cylinder shapes (16 mm diameter, 10 mm high) were compressed up
203 to 90% deformation at a probe speed of 2.5 mm/s. The mechanical properties of the hydrogel
204 samples prepared in this study are shown in Table S1. Three-dimensional curve fittings of the
205 fracture stress and fracture strain values of the prepared hydrogels were performed using gnuplot
206 software (Geeknet, Inc., Mountain View, CA, USA). The functions presented in Fig. S2 allowed
207 hydrogel samples with arbitrary mechanical properties to be obtained.

208 2.4.3. *In vitro* gastric digestion using the GDS

209 To independently analyze the influence of two mechanical properties (fracture stress and
210 fracture strain) on gastric disintegration, all hydrogel samples were prepared by fixing the
211 fracture stress or fracture strain (Table 2). The hydrogel samples are described relative to the
212 concentrations of agar and native gellan gum that they contain. For example, A0.7G0.6 indicates
213 the hydrogel sample contains 0.7 wt% agar and 0.6 wt% native gellan gum.

214 The conditions and procedure for the GDS experiments were based on our previous study
215 (Kozu et al., 2014). In brief, 100 g of the hydrogel was shaped into 5-mm cubes. The cubic shape
216 is more realistic than a spherical shape as the masticated food model for *in vitro* gastric digestion
217 experiments. The size of the hydrogel cubes was based on the size of solid particles transferred
218 to the stomach through the esophagus ($< \sim 5.0$ mm) (Jalabert-Malbos et al., 2007). The 5×5×5
219 mm hydrogel cubes containing agar or a mixture of agar and native type gellan gum were mixed
220 with 30 mL of SSF (pH 7, 37 °C) for 2 min to simulate mastication. A total of 260 mL of SGF
221 (pH 1.3, 37 °C) was added to the above mixture, and then the model gastric content was
222 introduced into the GDS vessel. The pH of the above-mentioned gastric content increased to
223 approximately 2.0 during GDS experiments. Each *in vitro* gastric digestion experiment using the
224 GDS was performed at 37 °C for up to 180 min. The progressive speed and generation frequency

225 of the ACWs of the GDS were set to 2.5 mm/s and 1.5 cycles/min, respectively. The
226 disintegration behavior of the hydrogel cubes in the GDS vessel was monitored and recorded
227 through the transparent window using a digital video camera.

228 *2.4.4. Observation and classification of digested hydrogel particles*

229 At the end of the *in vitro* gastric digestion experiment, the digesta was transferred to the top
230 of a stack of metal mesh sieves with mesh sizes of 0.60, 1.18, 2.36, and 3.35 mm. The hydrogel
231 particles retained in the gastric vessel were carefully rinsed with Milli-Q water. The hydrogel
232 particles on each sieve were gently washed with Milli-Q water to prevent further particle
233 breakdown during the operation. After the washing steps, the metal sieves were wiped to remove
234 the excess water, and each sieve was weighed to evaluate the particle size distribution of the
235 digesta.

236 *2.5. Statistical analysis*

237 The data were analyzed using SPSS Statistic 24 software. One-way analyses of variance were
238 performed to test significant differences in the mechanical properties, the half residence time
239 ($t_{1/2}$), and the ratio of small particles ($0.60 < d \leq 2.36$ mm) to the initial amount of hydrogel
240 particles at $p < 0.05$.

241

242 **3. Results and discussion**

243 *3.1. Comparison of in vitro and in vivo gastric digestion data for agar beads in LBG meals*

244 *In vitro* gastric digestion experiments on agar beads in LBG meals were conducted using the
245 GDS. The results obtained in this study were compared with the results of *in vivo* human gastric
246 digestion reported by Marciani et al. (2001). The spherical agar beads with different agar

247 concentrations in the range 1.5–3.0 wt% prepared in this work had a diameter of approximately
248 13 mm (Fig. 2), which is similar to those used for the *in vivo* study (12.7 mm diameter) and their
249 fracture forces ranged from 0.53 to 0.90 N (Marciani et al., 2001).

250 The (disintegrated) agar beads after the GDS experiments (150 min) are also shown in Fig. 2.
251 The agar beads with the lowest agar concentration and fracture force were largely disintegrated
252 compared with the agar beads with higher agar concentrations and fracture forces. The beads
253 initially packed near the bottom of the GDS vessel. Four of the beads were compactly aligned at
254 the bottom of the GDS vessel, while the minimum clearance was above 26 mm without
255 contraction of GDS rollers. When the ACWs were generated on the sidewalls of the GDS vessel,
256 the beads present in the occluded area were affected by the compression force and shear force.
257 Fracture of the beads was primarily observed because of the compression force caused by
258 interaction between neighboring beads.

259 The agar beads that remained intact (N) at a specific digestion time can be estimated using Eq.
260 1. The parameter k was calculated by carrying out curve fitting using Eq. 1. The half residence
261 time ($t_{1/2}$) of these beads was then calculated using Eq. 2 (Marciani et al., 2001):

$$262 \quad N = N_0 e^{-kt} \quad (1)$$

$$263 \quad t_{1/2} = \frac{\ln 2}{k} \quad (2)$$

264 where N_0 is the initial number of agar beads and t is the time. Eq. 1 was corrected to Eq. 3 below,
265 which subtracted the blank value of the intact bead number because of the geometry of the GDS
266 vessel:

$$267 \quad N = N_0' e^{-kt} \quad (3)$$

268 The batch-type process of the GDS used here does not empty the disintegrated particles. Because
269 of the vertical layout of the GDS vessel, at least five agar beads present in the upper region of the

270 GDS vessel were not compressed by the ACWs, even after disintegration of the beads present at
271 the bottom of the GDS vessel ($N'_0 = N_0 - 5$). Fig. S3 depicts the variations of N and the fitting
272 curves for different agar concentrations using Eq. 3. The half residence time of intact beads
273 significantly increased ($p < 0.05$) between the agar concentrations of 1.89 and 2.39 wt% (fracture
274 force of the agar beads between 0.65 and 0.78 N) (Fig. 3a). A similar trend was reported for the
275 *in vivo* human gastric digestion study (Fig. 3b) (Marciani et al., 2001). Although the absolute
276 value of $t_{1/2}$ was different for the GDS and *in vivo* cases, the threshold of the half residence time
277 of the agar beads was the same. This indicates that although replicating the complex movement
278 of the human stomach during food digestion was not fully achieved, the similar trends observed
279 among the GDS and *in vivo* data are useful for investigating the disintegration behavior of solid
280 foods during gastric digestion.

281 The physical forces generated in the human stomach that contribute to breaking down solid
282 foods are still not fully understood. It is currently believed that three forces are effective for the
283 disintegration of solid food particles in the human stomach: 1) the contraction force generated by
284 ACWs; 2) the shear force generated by changes of the gastric morphology; 3) the shear force
285 generated by the retropulsive fluid flow in the antrum while the pylorus is shut (Faas et al., 2001;
286 Indireskumar et al., 2000; Marciani et al., 2001). The contraction force generated in the GDS
287 vessel was approximately 8 N higher than that generated in the human stomach (see Sect. 2.1);
288 however, similar agar bead disintegration trends were observed. The findings obtained in this
289 section suggest that the contraction force generated from ACWs in the GDS does not act
290 sufficiently on the agar beads.

291 The disintegration of the large agar beads was primarily the result of brittle fracture because
292 the disintegrated pieces (e.g. Fig. 2 (d)) could be fit together to restore the original shape (Beer et

293 al., 2012). For the design of our GDS, it is appropriate to compare the force experienced by the
294 brittle solid food with *in vivo* data rather than the contraction force generated by the ACWs. It is
295 desirable that the contraction force generated by the ACWs in the GDS is adjusted to be the same
296 order as those from the *in vivo* data (e.g., 0.8 N (liquids) and 2.2 N (solids) according to Vassallo
297 et al. (1992)). Therefore, we adjusted the contraction force generated by the ACWs to be <10 N
298 (see Sect. 2.1). As the force acting on the food particles in the stomach is mainly affected by
299 their size, shape, packing, and interactions, it is useful to compare the *in vitro* results with the *in*
300 *vivo* results reported by Marciani et al. (2001) using similar food samples. Conversely, the
301 contraction force generated by the ACWs of the RD-IV-HSM, which is another *in vitro* gastric
302 model, was 3.37 ± 0.59 N; however, none of the agar beads with fracture forces in the range
303 0.15–0.65 N fractured during the 1.5 h digestion process (Chen et al., 2016). The study reported
304 that 3.37 N may not sufficiently act on the large particles of agar beads, which was similar to the
305 GDS findings. In comparison, the half residence time of intact beads significantly increased ($p <$
306 0.05) when the fracture force of the agar beads was increased from 0.65 N to 0.78 N during both
307 the *in vivo* experiments and the *in vitro* experiments using the GDS as shown in Fig. 3. This
308 implies that most of the compression forces acting on neighboring particles are in the range
309 0.65–0.78 N (agar conc. 1.89–2.39 wt%). During the GDS digestion process, the contraction
310 force generated by ACWs converts to compression forces acting on some of the neighboring
311 particles, while some particles may escape compression because of slippage caused by their
312 smooth spherical shape. In the case of brittle fracture, the compression forces acting on
313 neighboring particles (0.65–0.78 N) were able to easily breakdown the beads with low fracture
314 forces (< 0.65 N), but had little effect on the beads with high fracture forces (> 0.78 N).

315 Somewhat larger compression forces acting on some neighboring particles may result in the
316 disintegration of the beads with high fracture forces (> 0.78 N).

317 Although replicating the complex movement of the human stomach during food digestion is
318 difficult, these results indicate that the GDS can simulate the disintegration behavior trends of
319 solid foods in the human stomach. Although there are some differences in the absolute half
320 residence time of agar beads with a given fracture force between the *in vitro* and *in vivo* data, the
321 similar trends observed are useful for investigating the disintegration behavior of solid foods
322 during gastric digestion. Of course, the absolute values of the half residence time results for the
323 GDS likely would have been closer to the *in vivo* data if a GDS equipped with emptying and
324 other more complex functions was used.

325 *3.2. Effect of hydrogel mechanical properties on their disintegration in the GDS*

326 *3.2.1. Direct observation of digestion behavior and size distribution of digested particles*

327 The hydrogel samples were initially cut into 5-mm cubes and settled on the bottom of the
328 GDS vessel. The (disintegrated) hydrogel cubes (A1.4, A1.1G0.7, A0.4G1.1) before or after the
329 GDS experiments (180 min) are also shown in Fig. S4. In the case of A1.4, many fractured
330 hydrogel cubes were observed. In the case of A1.1G0.7 and A0.4G1.1, fewer fractured cubes and
331 some small corner or surface pieces caused from slight abrasion were observed. Fig. 4 depicts
332 the gastric content variation observed at the start and the end of *in vitro* gastric digestion using
333 the GDS and the change of particle size distribution during GDS experiments in the case of
334 hydrogel cubes (A1.4, A1.1G0.7, A0.4G1.1) for which the fracture strains were different while
335 the fracture stresses were the same, maintained at ca. 40 kPa (Table 2). Because of the size
336 reduction, small hydrogel particles tended to distribute and pack more densely in the lower

337 region of the gastric content, resulting in a decrease in the packing height of the hydrogel
338 particles. From the change in packing height shown in Fig. 4 (i, ii), we found that the small
339 particles ($0.60 < d \leq 2.36$ mm) of A1.4 showed more disintegration than A1.1G0.7 and A0.4G1.1
340 after 180 min. The wet weight of the fraction between 0.60 mm and 2.36 mm increased with time,
341 which corresponds to the size of particles that solid food disintegrated to approximately 1–2 mm
342 in diameter and that were emptied from the pylorus during human digestion (Kelly et al., 1980,
343 Guo et al., 2014). In the case of A1.4, which had a fracture strain of ca. 30%, the weight ratio of
344 the small particles ($0.60 < d \leq 2.36$ mm) to the initial amount of hydrogel particles increased to
345 22.1 % and the wet weight of the largest fraction ($d > 3.35$ mm) decreased to 74.8 g after 180
346 min (Fig. 4 (a, iii)). Compared with the result of the flask-shaking experiments (Kozu et al.,
347 2015), the effect of the largest hydrogel particles ($d > 3.35$ mm) breaking down into small
348 particles ($0.60 < d \leq 2.36$ mm) using the GDS is clear.

349 *3.2.2. Relationship between hydrogel mechanical properties and disintegration*

350 Fig. 5 shows the relationship between the mechanical properties of all hydrogel samples and
351 their disintegration using the GDS ($n=3$). When the fracture strain exceeded the threshold value
352 (between ca. 30% and 40%), the degree of disintegration was markedly reduced: the ratio of
353 small particles ($0.60 < d \leq 2.36$ mm) to the initial amount of hydrogel particles decreased
354 significantly ($p < 0.05$) as described in Fig. 5(a). In the region in which fracture strain is small
355 (ca. 30%), it was found that when fracture stress exceeded a certain value (40–60 kPa), the
356 degree of hydrogel particle disintegration decreased significantly ($p < 0.05$) (Fig. 5(b)). In the
357 region in which fracture strain is large (ca. 40% and 65%), the fracture stress had little influence
358 on disintegration; the ratio of small particles ($0.60 < d \leq 2.36$ mm) to the initial amount of
359 hydrogel particles showed little change ($p > 0.05$) and fracture stress varied from ca. 20 kPa to

360 60 kPa. These size-reduction trends could also be seen in the gradual increase of the weight ratio
361 of small particles ($0.60 < d \leq 2.36$ mm) to the initial amount of hydrogel particles during GDS
362 digestion experiments (Fig. S5).

363 *3.2.3. Possible mechanisms for the disintegration of hydrogel particles*

364 A possible mechanism for the gastric disintegration of hydrogels with different mechanical
365 properties in GDS experiments is shown schematically in Fig. 6. We assume that there are two
366 types of fracture mechanism (brittle fracture and ductile fracture). Brittle fracture shows no
367 apparent plastic deformation before fracture, while ductile fracture shows an extensive plastic
368 deformation before fracture (Beer et al., 2012). The concept of the brittle-ductile transition of
369 double network hydrogels has been reported as being applicable to various species of polymeric
370 materials. This could explain how the brittle hydrogels change into ductile hydrogels because of
371 increasing the amount of ductile component (Ahmed et al., 2014).

372 At equivalent fracture strain (ca. 30%), the fracture stress of A1.0, A1.4, A1.7, and A1.9
373 increased with agar concentration. Measurement of the mechanical properties of these hydrogel
374 samples showed the typical stress-strain curves for brittle materials (data not shown). The key
375 factor determining the disintegration is the fracture stress (σ_f) compared with the compression
376 stress acting on neighboring hydrogel particles (σ_a). Guo et al. (2015) investigated the
377 disintegration of whey protein emulsion gels with different fracture forces (soft and hard gels)
378 using HGS and showed that the soft gel broke down faster than the hard one. In our experiments,
379 when the fracture stress exceeded a threshold (between 40 kPa and 60 kPa), the degree of
380 hydrogel particle disintegration markedly decreased (Fig. 5(b)). The compression force acting on
381 neighboring hydrogel particles can be estimated at approximately 1.0–1.5 N calculated from the
382 above-mentioned threshold and the contact area of the hydrogel samples (25 mm²: surface area

383 of one face of a 5-mm hydrogel particle cube). A similar disintegration pattern was obtained in
384 the experiments with agar beads in LBG meals (see Sect. 3.1), the difference in the absolute
385 value of the compression force acting on neighboring hydrogel particles may be due to the
386 different sample shapes (sphere or cube).

387 However, in the case of samples that had a fracture strain above 40%, the influence of
388 fracture stress on disintegration was hardly observed. The hydrogel samples showed the typical
389 stress-strain curves for ductile materials (data not shown) and could sustain an extensive plastic
390 deformation without fracture. Observation after 180 min of GDS digestion (e.g. Fig. S4(c))
391 showed that deformation because the ACWs did not exceed the maximum plastic deformation of
392 the ductile hydrogel particles and that only slight abrasion of the surface or corners occurred.

393

394 **4. Conclusions**

395 Based on the comparison of the GDS results and *in vivo* data (Marciani et al., 2001) using
396 agar beads with different fracture forces in LBG meals, we concluded that the fracture of solid
397 foods caused by the simulated ACWs of the GDS was comparable to that of the human stomach.
398 Our GDS results demonstrated that two fracture mechanisms (brittle fracture and ductile
399 fracture) occurred for hydrogel cubes during gastric digestion. In the case of the low fracture
400 strain hydrogels, the degree of hydrogel disintegration was affected by their fracture stress and
401 was decreased when their fracture stress was greater than a threshold value because mainly
402 brittle fracture occurred. In the case of the high fracture strain hydrogels, little effect of fracture
403 stress on disintegration was found because ductile fracture did not occur when there was
404 insufficient plastic deformation. This study provides useful insights for understanding the gastric

405 digestion behavior of food hydrogels with different mechanical properties and better design of
406 novel solid foods whose digestibility can be controlled based on life stage and health conditions.

407

408 **Declaration of interests**

409 None.

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413

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530

531 **Figure captions**

532 **Table 1.** Composition of simulated salivary fluid (SSF) and simulated gastric fluid (SGF).

533 **Table 2.** Mechanical properties of the hydrogel samples containing agar or a mixture of agar and
534 native type gellan gum.

535 **Fig. 1.** Contraction force measured in the GDS. (a): Key components of the GDS. (b): Two
536 layered structure of the rollers. (c): Digital manometer with a silicone balloon.

537 **Fig. 2.** Photographs showing the appearance of the agar beads before digestion and after 150 min
538 of digestion. (a) 1.50 wt% agar. (b) 1.89 wt% agar. (c) 2.39 wt% agar. (d) 3.00 wt% agar.

539 **Fig. 3.** Half residence time of the agar beads for four agar concentrations. (a) *In vitro* (GDS):
540 n=3 for each agar concentration. (b) *In vivo* (Marciani et al., 2001): n=9 for each agar
541 concentration.

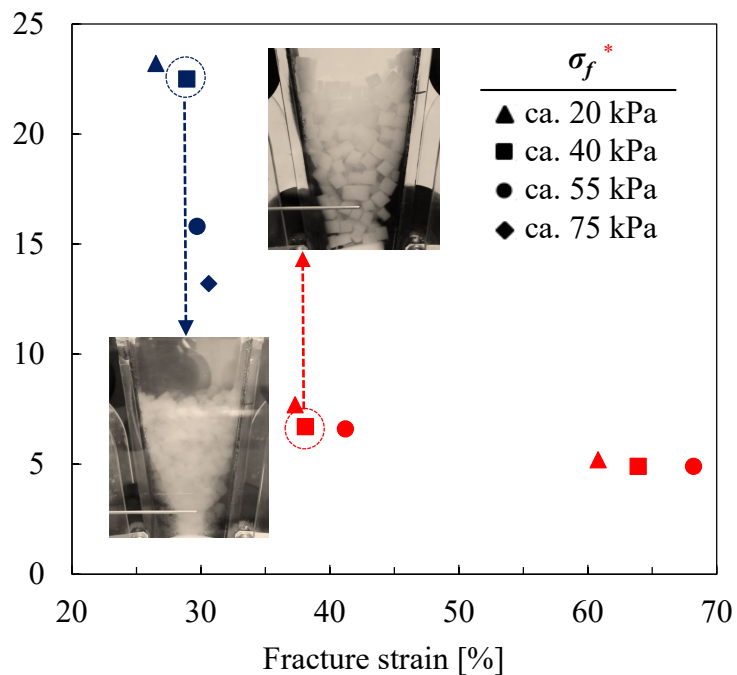
542 Values with different letters are significantly different ($p < 0.05$).

543 **Fig. 4.** Direct observation photographs of the hydrogel cubes during GDS digestion experiments.
544 (a) A1.4 (fracture stress 39.6 kPa, fracture strain 28.9%). (b) A1.1G0.7 (fracture stress 40.0 kPa,
545 fracture strain 38.1%). (c) A0.4G1.1 (fracture stress 37.5 kPa, fracture strain 63.9%). (i) Hydrogel
546 cubes before digestion. (ii) Hydrogel cubes after 180 min of digestion in the GDS. (iii) Size
547 distribution change of A1.4, A1.1G0.7 and A0.4G1.1 with digestion time in the GDS.

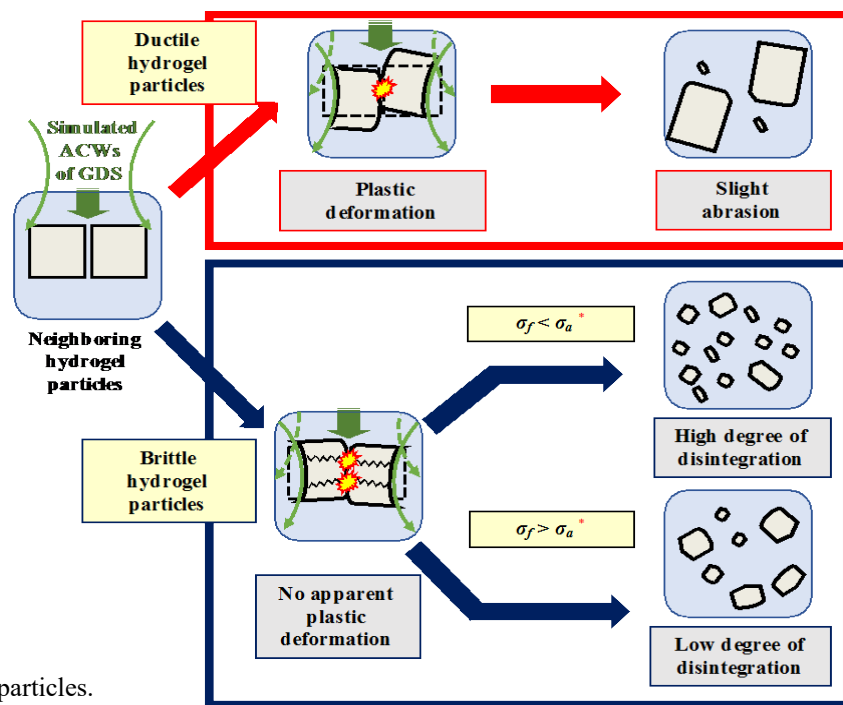
548 **Fig. 5.** Effects of mechanical properties on the gastric disintegration of hydrogel cubes. (a)
549 Horizontal axis: fracture strain. (b) Horizontal axis: fracture stress (A1.0, A1.4, A1.7, A1.9).

550 **Fig. 6.** Mechanisms for the gastric disintegration of hydrogels with different mechanical
551 properties during the GDS experiments. σ_f : fracture stress of hydrogels. σ_a : compression stress
552 acting on neighboring particles.

Weight ratio of 0.60 mm – 2.36 mm hydrogel particle fraction to the initial total amount [%]



* σ_f : fracture stress of hydrogels. σ_a : compression stress acting on neighboring particles.



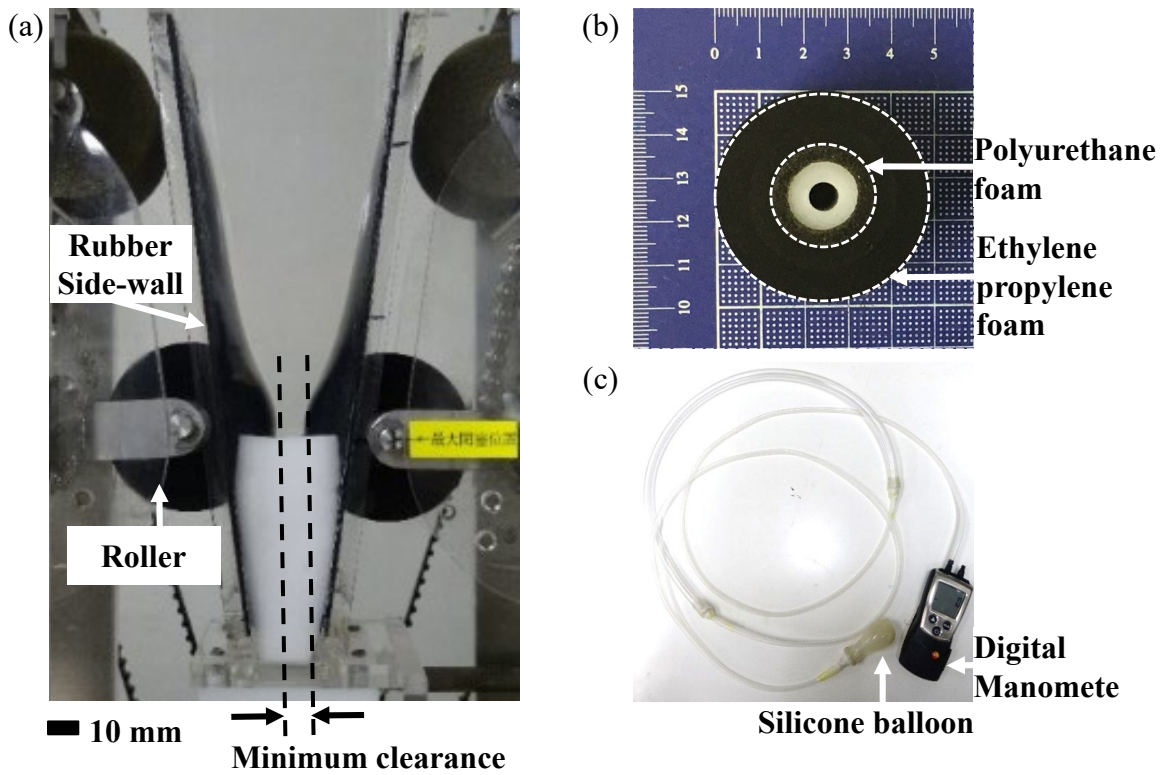
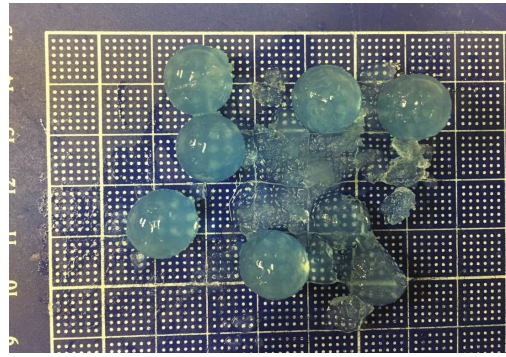
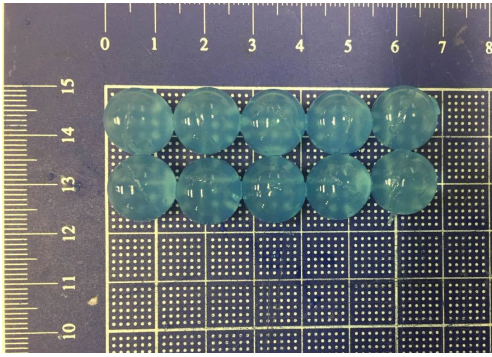


Fig. 1

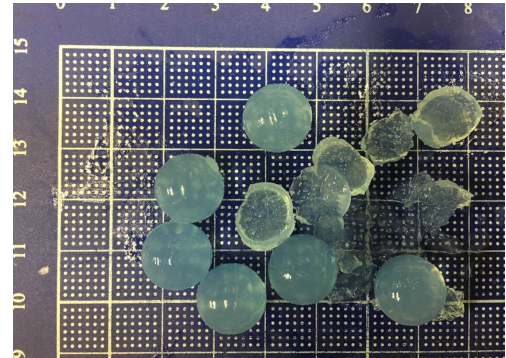
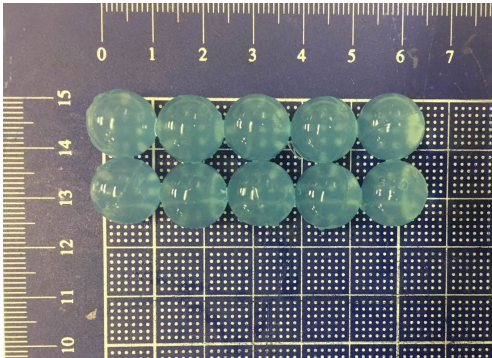
Before digestion

After 150 min digestion

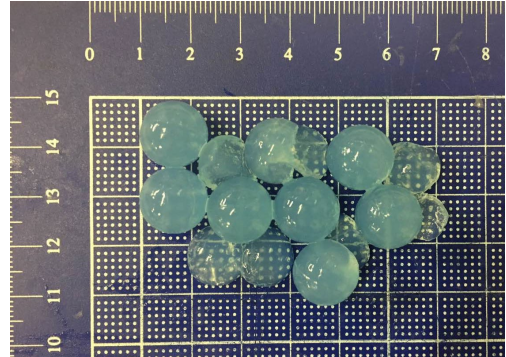
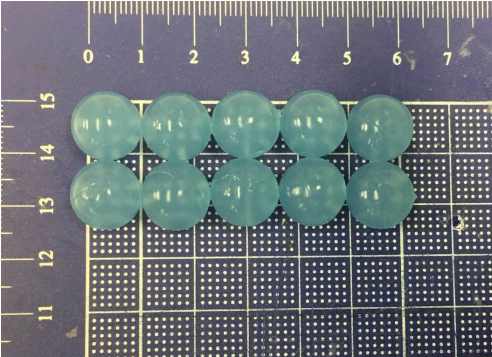
(a)



(b)



(c)



(d)

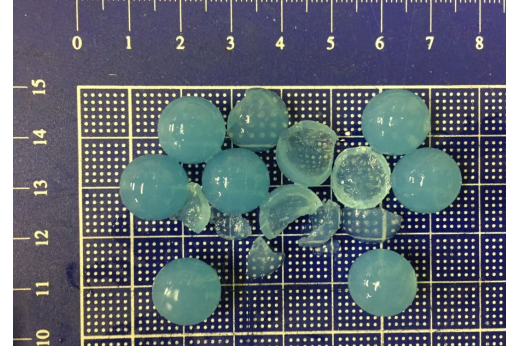
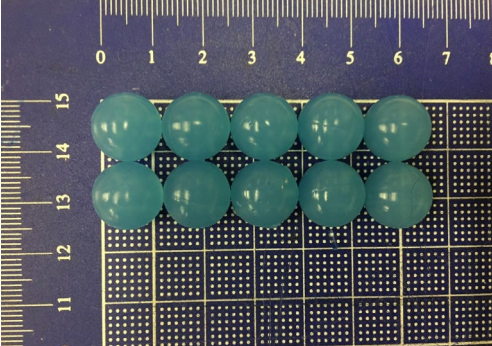


Fig. 2

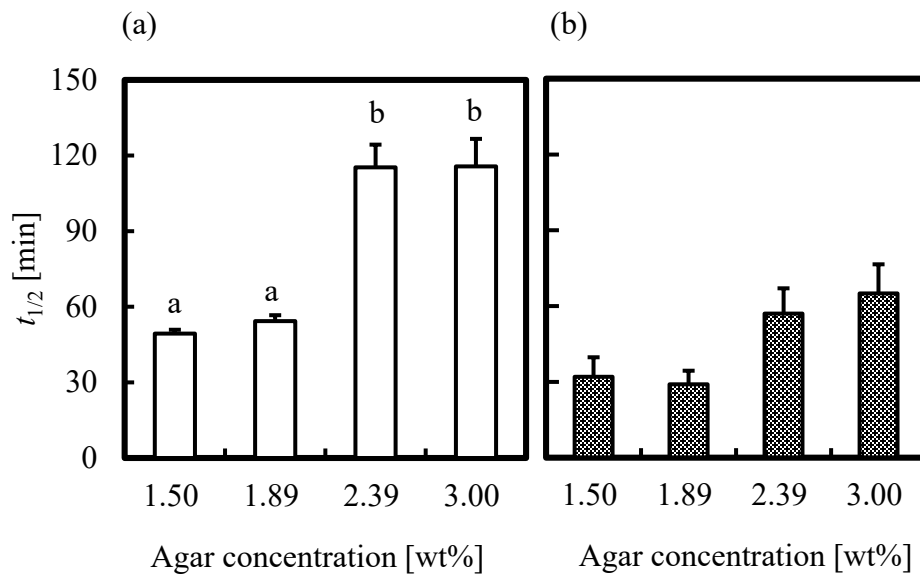


Fig. 3

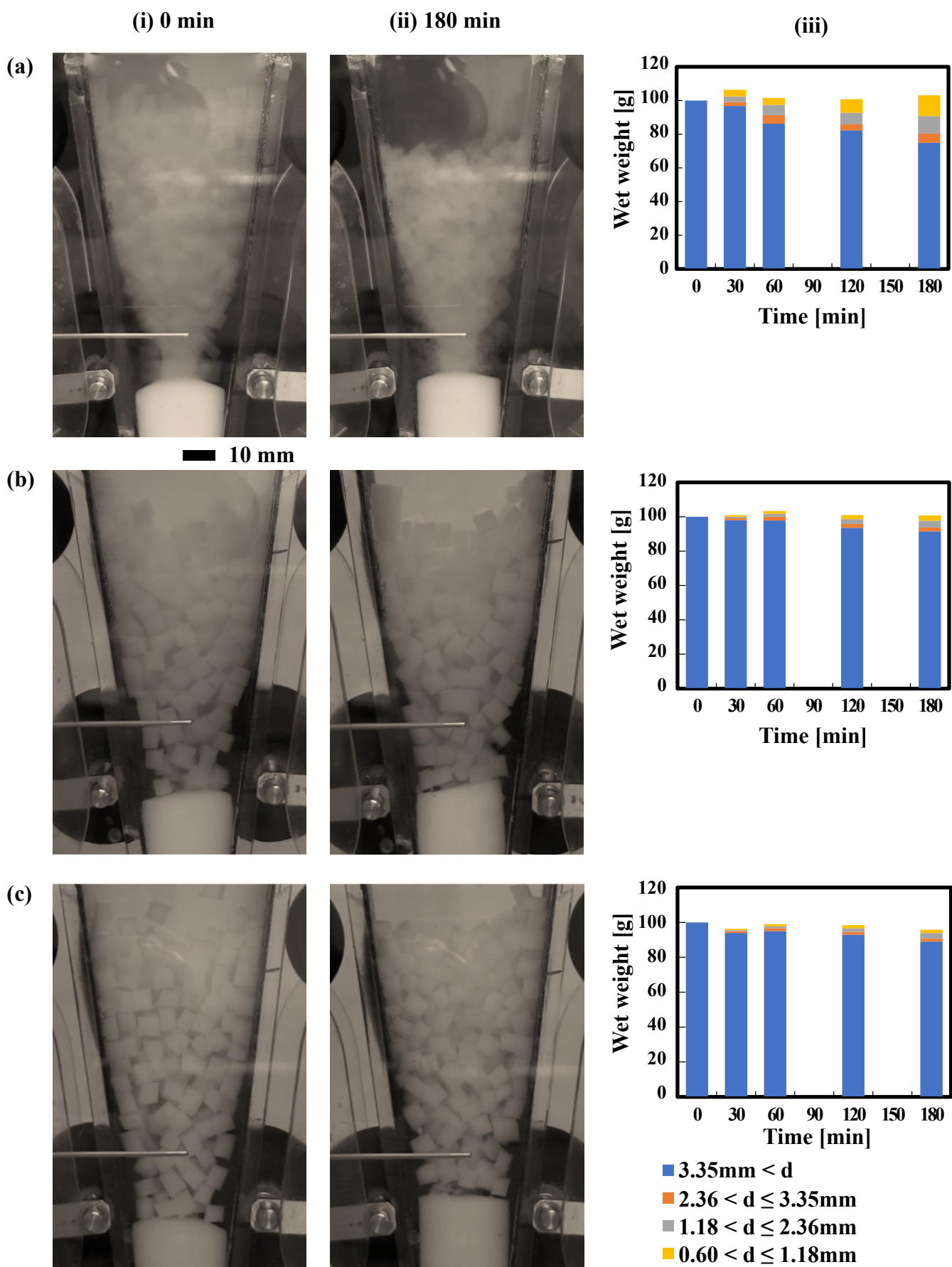


Fig. 4

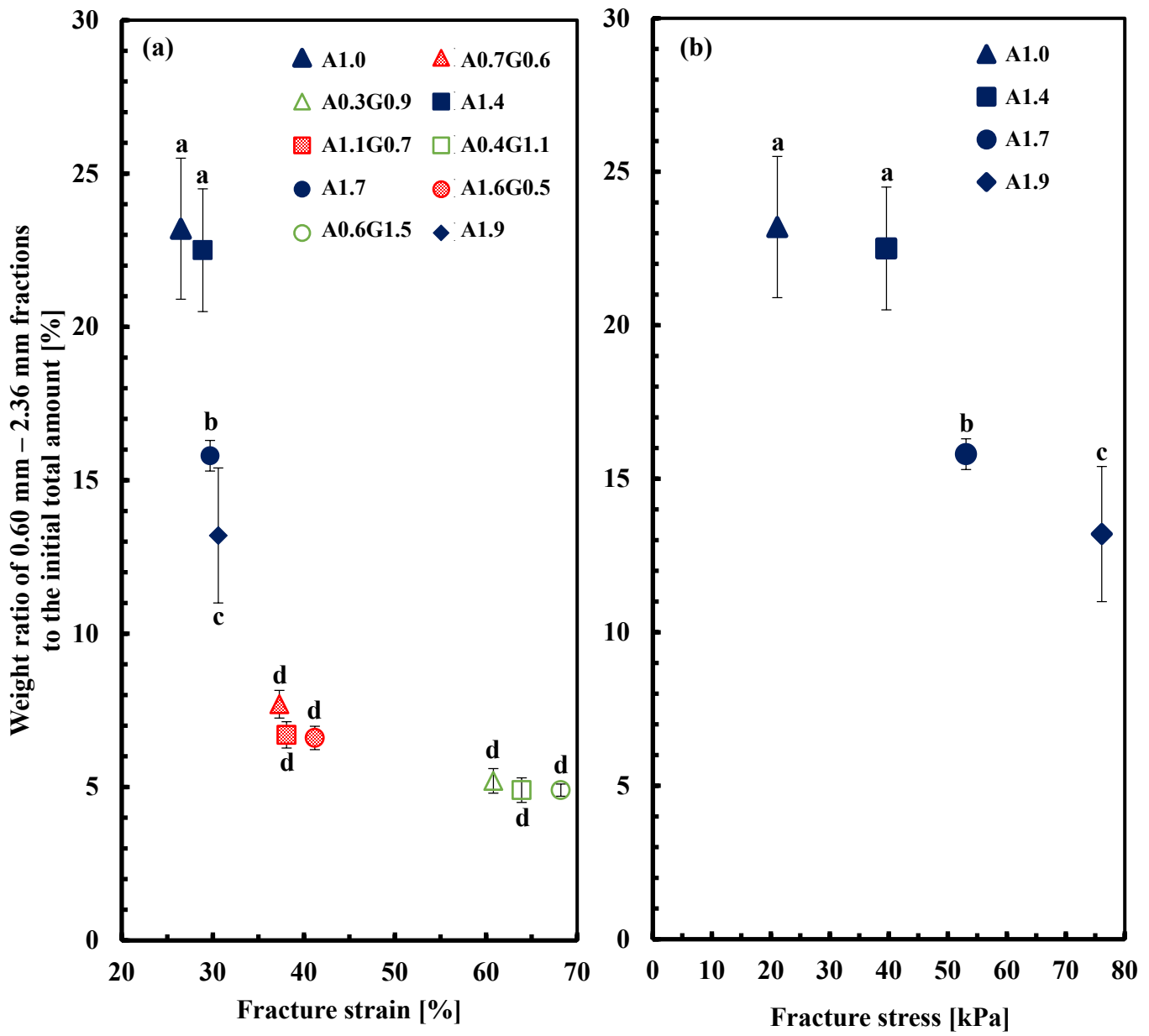


Fig. 5

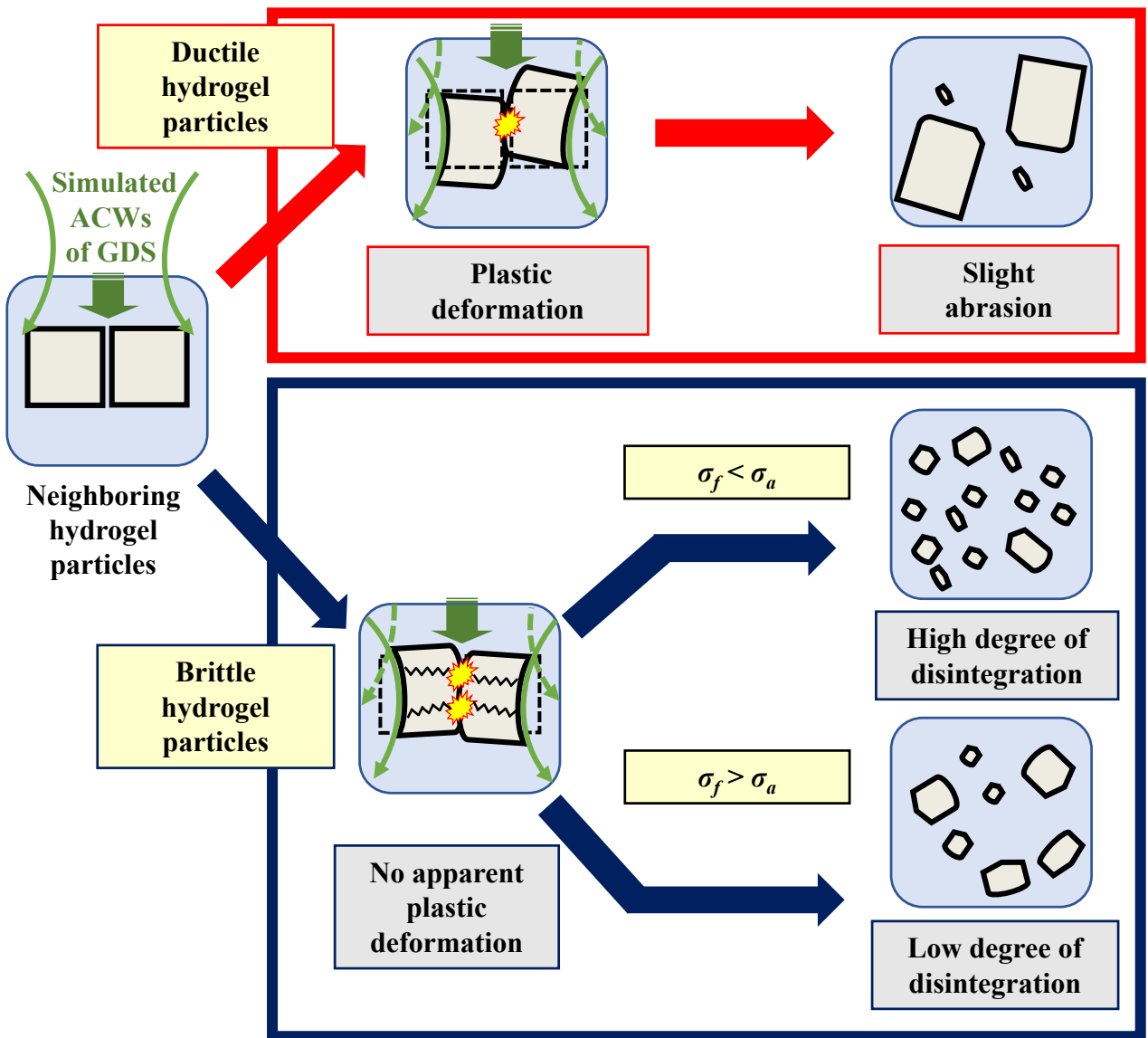


Fig. 6

Table 1 Composition of simulated salivary fluid (SSF) and simulated gastric fluid (SGF)

Constituent	SSF	SGF*
	pH 7.0	pH 1.3
	Conc. in SSF [g/L]	Conc. in SGF [g/L]
KCl	1.126	0.514
KH ₂ PO ₄	0.503	0.122
NaHCO ₃	1.142	2.100
NaCl	-	2.760
MgCl ₂ (H ₂ O) ₆	0.030	0.020
(NH ₄) ₂ CO ₃	0.006	0.074
CaCl ₂ (H ₂ O) ₂	0.221	0.022
α -Amylase	2.530 (150 U/mL)	-
Pepsin	-	5.602 (4000 U/mL)

*6 M HCl solution was used for pH adjustment.

Table 2 Mechanical properties of hydrogel samples

Sample	Concentration (wt%)		Fracture stress (kPa)		Fracture strain (%)	
	Agar	Native gellan gum	Calculated value ¹	Measured value	Calculated value ²	Measured value
A1.0	1.0	0	23.8	21.1±2.9 ^a	25.5	26.5±2.2 ^a
A0.7G0.6	0.7	0.6	20.6	23.4±1.6 ^a	42.9	37.3±0.6 ^b
A0.3G0.9	0.3	0.9	21.9	22.6±1.7 ^a	57.4	60.8±1.4 ^c
A1.4	1.4	0	43.7	39.6±3.0 ^b	28.4	28.9±2.8 ^a
A1.1G0.7	1.1	0.7	39.1	40.0±2.2 ^b	41.1	38.1±2.2 ^b
A0.4G1.1	0.4	1.1	36.0	37.5±1.8 ^b	57.9	63.9±2.4 ^c
A1.7	1.7	0	62.4	54.1±4.0 ^c	29.5	29.7±1.3 ^a
A1.6G0.5	1.6	0.5	61.2	56.7±3.6 ^c	39.9	41.2±1.9 ^b
A0.6G1.5	0.6	1.5	62.9	62.7±6.3 ^d	58.2	68.2±1.7 ^d
A1.9	1.9	0	76.7	76.1±7.8 ^e	33.7	30.6±2.4 ^a

All mechanical characteristics were measured at 37 °C with five replications.

¹ Calculated from equation in Fig. S3 (a)

² Calculated from equation in Fig. S3 (b)

^{a - e} Values with different superscripts are significantly different ($p < 0.05$) within the same groups.

Regarding the sample code, the values after A and G mean the concentrations of agar and native gellan gum, respectively.

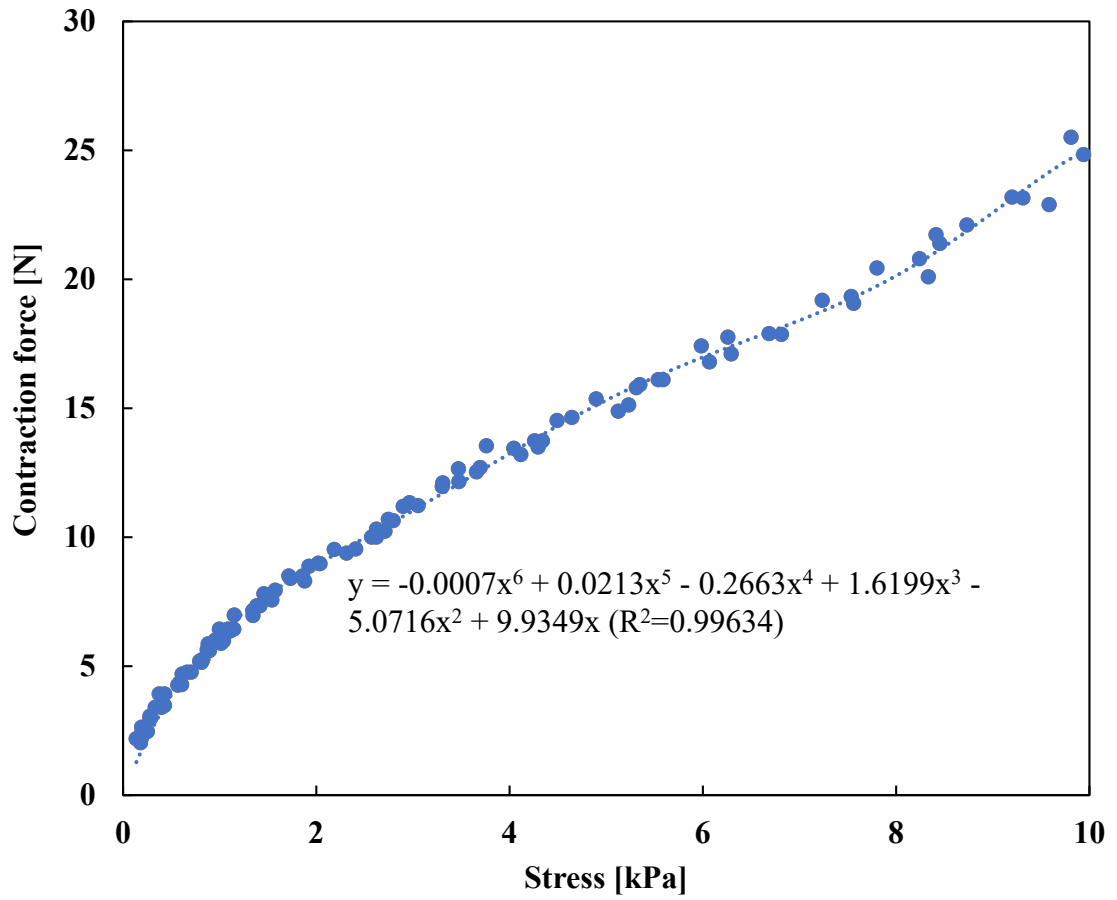
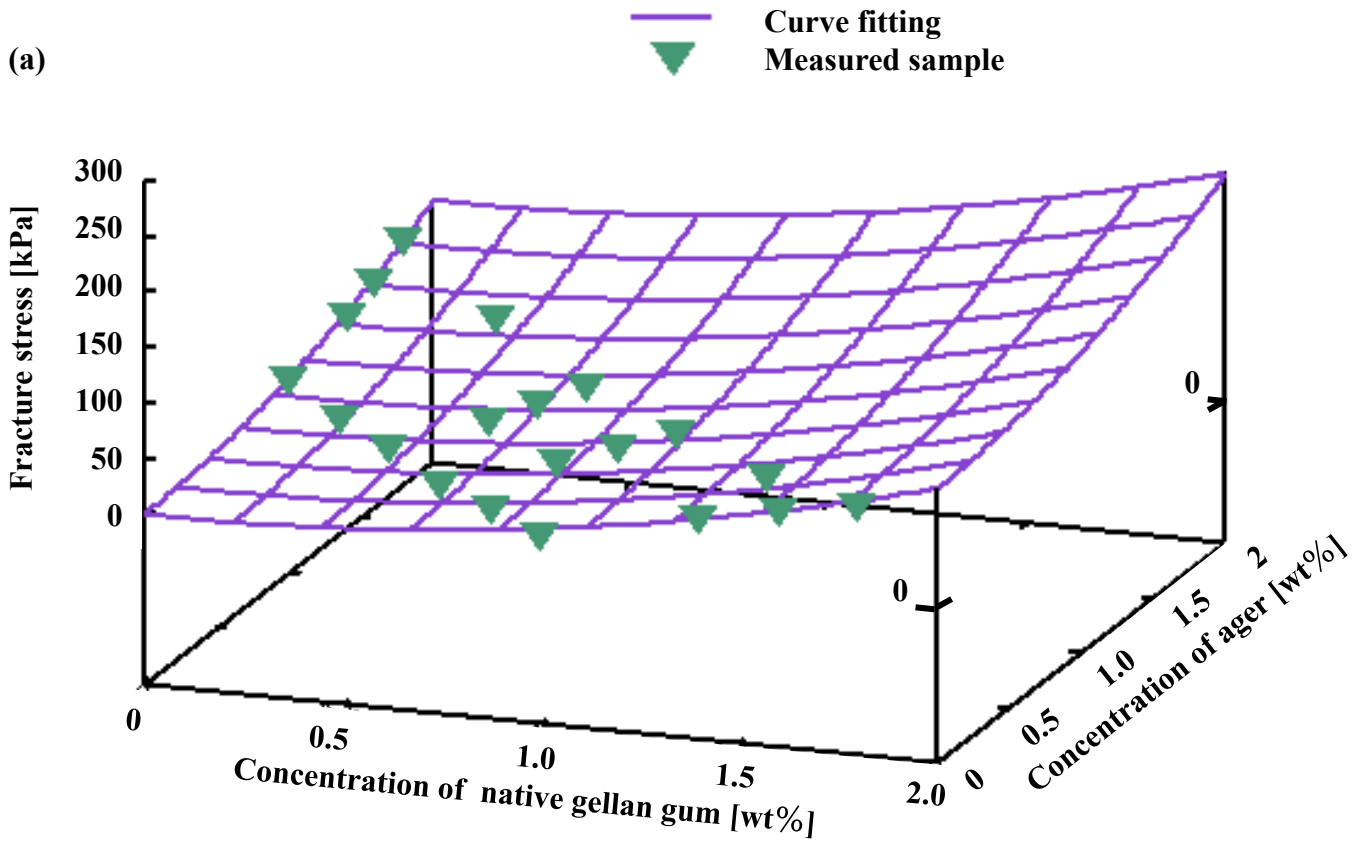


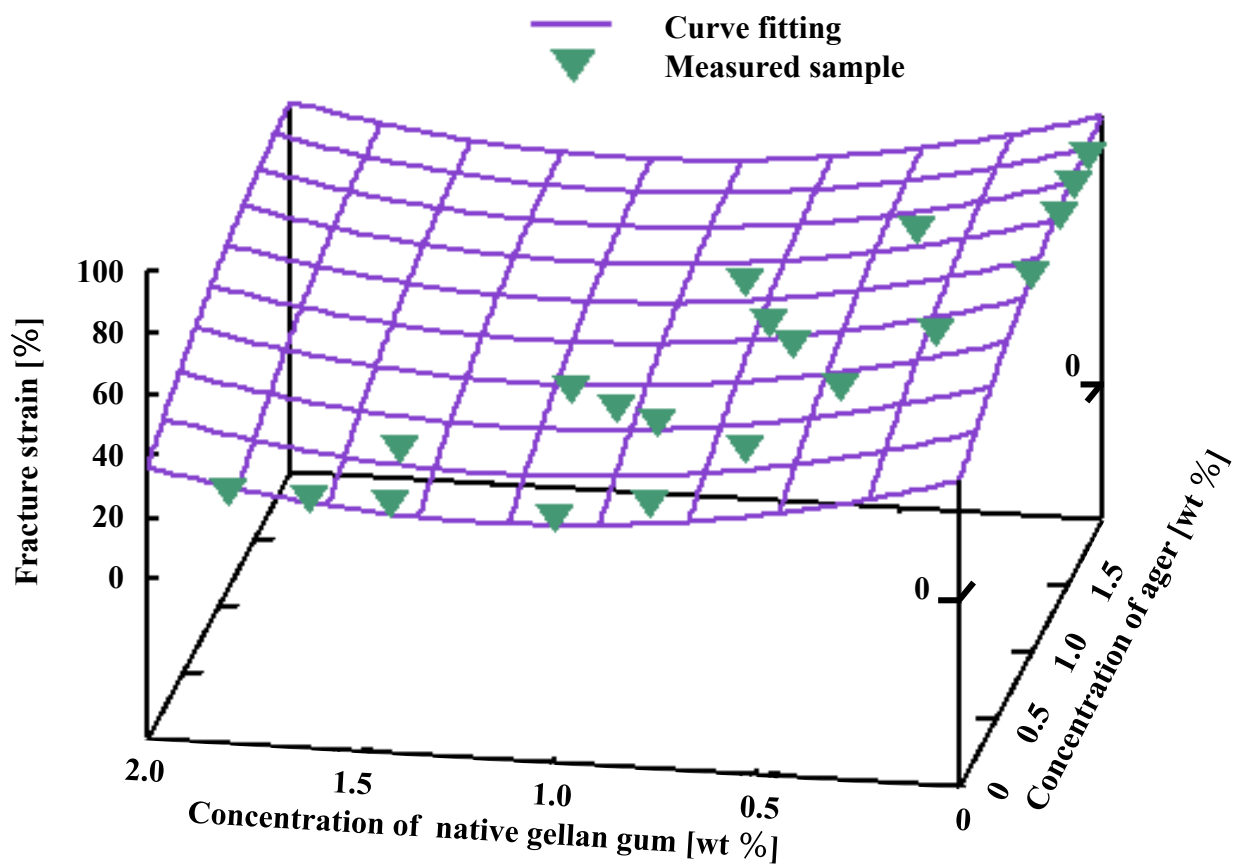
Fig. S1 Correlation between the contraction force of the GDS rollers and the maximum pressure of the balloon



Fracture stress $f_1(x,y) = 24.985x^2 + 18.373y^2 + 0.403xy - 2.31841x + 5.453y$
 $(R^2 = 0.986)$ (x: agar concentration; y: native gellan gum concentration)

Fig. S2 Curve fitting of the mechanical properties of each hydrogel sample. (a) Correlation between the concentrations of the two hydrogel agents and fracture stress

(b)



Fracture strain $f_2(x,y) = -5.580x^2 + 16.656y^2 + 26.840x - 39.227y + 48.075$
($R^2 = 0.953$) (x: agar concentration; y: native gellan gum concentration)

(Continues) Fig. S2 Curve fitting of the mechanical properties of each hydrogel sample. (b) Correlation between the concentrations of the two hydrogel agents and fracture strain.

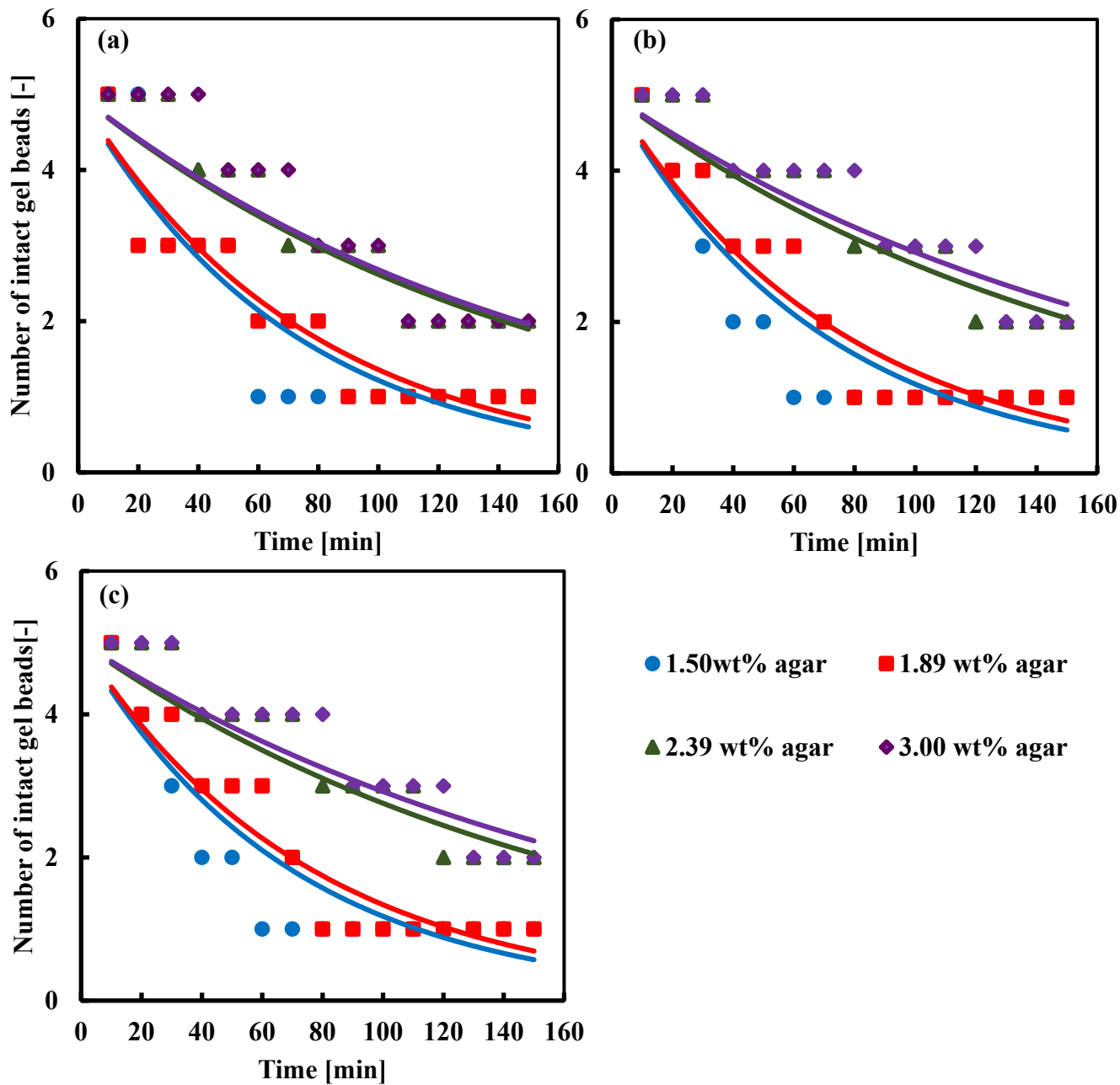


Fig. S3 Time-dependent change of the number of intact agar beads. (a) First, (b) second, and (c) third experiments.

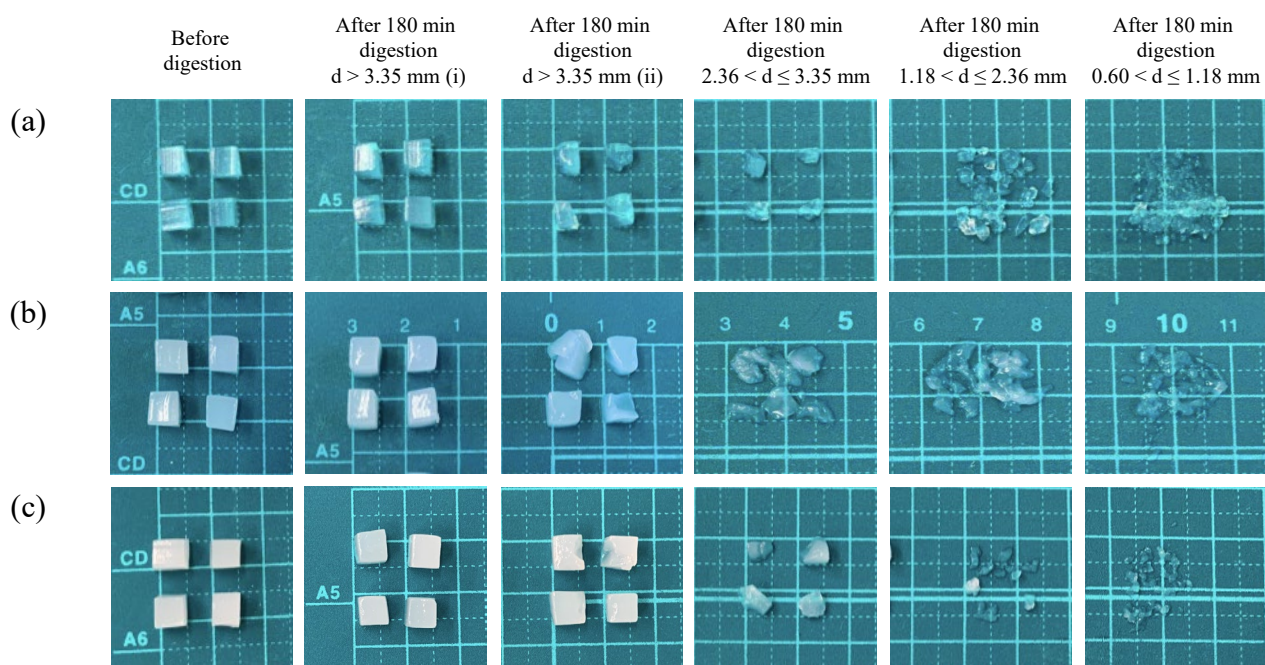


Fig. S4 Photographs showing the examples of the cubes before digestion and after 180 min of digestion. (a) A1.4. (b) A1.1G0.7. (c) A0.4G1.1. (i) Intact hydrogel cubes. (ii) Damaged hydrogel cubes.

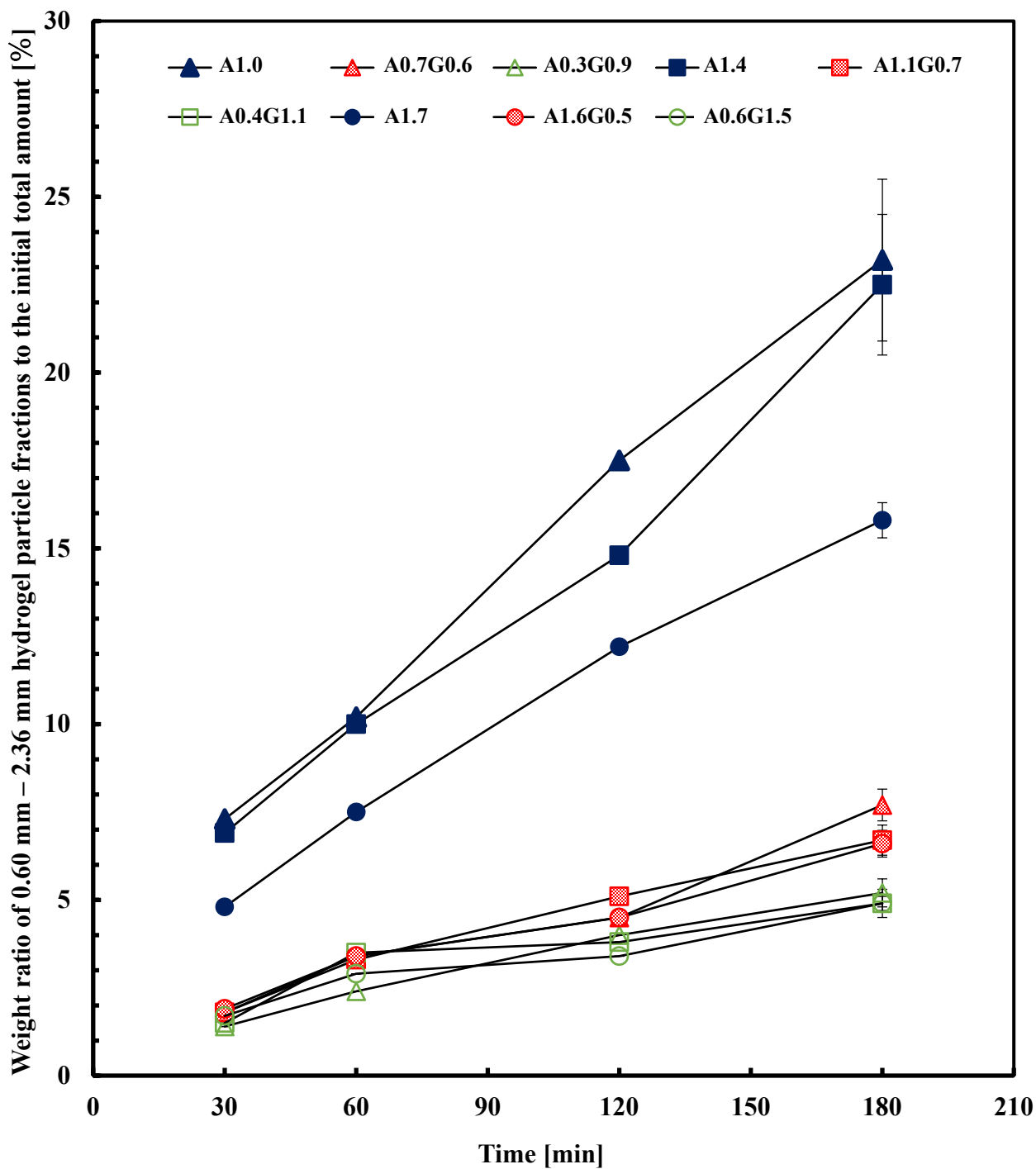


Fig. S5 Weight ratio change of 0.60 mm - 2.36 mm fractions digestion time in the GDS

Table S1 Mechanical properties of the hydrogel samples

Concentration [wt%]		Fracture stress [kPa]		Fracture strain [%]	
Agar	Native gellan gum	Measured value	Calculated value*	Measured value	Calculated value**
1.0	0	24.2	23.8	26.6	25.5
1.4	0	40.7	43.7	27.3	25.8
1.6	0	54.6	55.7	30.2	28.0
1.8	0	70.2	69.3	30.8	31.4
0.8	0.2	17.1	16.7	26.6	32.5
1.4	0.4	50.3	48.5	31.9	35.1
0.6	0.4	16.9	13.0	36.8	40.4
0.8	0.6	22.7	24.3	36.9	40.2
1.0	0.6	29.3	31.1	37.0	40.7
1.1	0.7	36.9	38.9	38.8	41.6
0.4	0.6	11.3	12.8	50.2	49.1
0.6	0.8	23.1	24.7	52.6	49.9
0.6	1.0	33.6	32.0	53.2	50.4
0.7	1.1	43.7	40.4	60.7	50.9
0.2	0.8	15.9	16.0	60.8	58.8
0.4	1.4	51.3	53.0	64.6	63.2
0	1.0	19.8	22.7	72.2	69.3
0	1.4	48.7	45.7	74.1	74.7
0	1.6	63.1	60.2	75.2	76.7
0	1.8	74.0	76.8	76.2	78.3

All mechanical characteristics were measured at 37 °C with five replications.

* Calculated from equation in Fig. S3 (a)

** Calculated from equation in Fig. S3 (b)