Copper-catalyzed [3 + 2] Annulation of Azides with a (Difluorovinyl)zinc Complex, Fluoroacetylene Equivalent

Takeshi Fujita, Masafumi Takeishi, and Junji Ichikawa*

Division of Chemistry, Faculty of Pure and Applied Sciences, University of Tsukuba, Tsukuba, Ibaraki 305-8571, Japan Supporting Information Placeholder



ABSTRACT: The copper-catalyzed [3 + 2] annulation of organic azides with (2,2-difluorovinyl)zinc chloride–TMEDA was achieved via C–F bond cleavage. Thus, a series of 1-substituted 4-fluorotriazoles were synthesized in high yields. In this reaction, the difluorovinylzinc complex functions as an easy-to-handle equivalent of fluoroacetylene (FC=CH) to undergo cycloaddition with azides. This work offers a facile and practical method for the use of fluoroacetylene, which has been considered to be highly reactive and difficult to handle and control for synthetic applications.

The methods for chemical transformation of various types of fluorine-containing compounds have been developed to achieve considerable progress in pharmaceutical, agrochemical, and materials sciences.^{1,2} However, despite the synthetic utility of alkynes, the reactivities of fluoroalkynes that bear a fluorine substituent on the sp carbon remain unclear owing to their instability, difficult preparation, and difficult isolation. Among them, fluoroacetylene (FC=CH), the second smallest alkyne, was first synthesized in 1959 via the pyrolysis of fluoromaleic anhydride at high temperature (650 °C) under vacuum (5–7 mmHg).³ However, the required harsh reaction conditions hampered the synthetic versatility of this method. Thereafter, a metalation-elimination protocol using fluorinated haloalkenes has been adopted for the synthesis of fluoroacetylene.⁴ Although this method enabled the in situ generation of gaseous fluoroacetylene in solution and its application to synthetic use, its reactions are difficult to control because the metalated intermediates and produced fluoroacetylene are unstable. For example, Sauvêtre have reported the generation of fluoroacetylene via lithiation of 1,1-difluoroethylene (Scheme 1a).⁵ In this case, β -fluorine elimination from the intermediary 2,2-difluorovinyllithium proceeded at temperature above -80°C to afford fluoroacetylene.⁶

Recently, we have developed 2,2-(difluorovinyl)zinc chloride–TMEDA (1) stabilized by the coordination of N,N,N',N'tetramethylethylenediamine (TMEDA), which serves as a difluorovinyl nucleophile in palladium- and copper-catalyzed couplings (Scheme 1b).⁷ We envisaged that the difluorovinylzinc complex could also serve as an easy-to-handle equivalent of fluoroacetylene, which would be of considerable synthetic utility. Thus, we revisited fluoroacetylene chemistry⁸ and succeeded in constructing 4-fluorinated triazole rings by treating aryl, benzylic, allylic, and alkyl azides with difluorovinylzinc complex 1, where Huisgen-type [3 + 2] annulation^{9,10} occurred (Scheme 1c).

Scheme 1. Fluoroacetylene and Its Equivalent



First, we selected 4-(azidomethyl)biphenyl (**2a**) as a model compound for the examination of annulation with (difluorovi-

nyl)zinc complex 1, which was prepared according to the reported procedure⁷ via the deprotonation of 1,1difluoroethylene and subsequent transmetalation and ligation (Table 1). Although only a trace amount of annulated product was provided upon the reaction of 2a and 1 without a catalyst (Entry 1), the use of copper catalysts in 1,4-dioxane at room temperature efficiently afforded the corresponding [3 + 2]annulation product, 4-fluorotriazole 3a (Entries 2-7). Among copper catalysts screened, CuCl was determined to be a prospective catalyst, which afforded 3a in 81% yield (Entry 7). To improve the yield of 3a, we screened ligands on CuCl. While phosphine ligands suppressed annulation (Entries 8 and 9), nitrogen ligands were effective (Entries 10-13). Specifically, 1,10-phenanthroline improved the yield of **3a** up to 86% (Entry 12).

Table 1. Screening of Conditions for [3 + 2] Annulation of **2a** with **1**.

H F、	••• + N ₃ ∕		[Cu] (15 mol %) gand (15 mol %)	N-N	Ph
F 1 (L = TME	DA)	Ph 2a	1,4-dioxane rt, 18 h	F	3a
entry	[Cu]		ligand		3a (yield %) ^{a}
1	None		-		trace
2	Cu ₂ O		-		trace
3	$CuSO_4$		-		2
4	CuOAc		-		80
5	CuI		-		70
6	CuBr		-		78
7	CuCl		-		81
8	CuCl		PPh3 ^b		2
9	CuCl		dppe		14
10	CuCl		pyridine ^b		80
11	CuCl		2,2'-bpy		83
12	CuCl		1,10-phen		86 (83)
13	CuCl	2,9	-diMe-1,10-phe	n	84

^aYield was determined by ¹⁹F NMR spectroscopy using PhCF₃ as an internal standard. Isolated yield was shown in parentheses. ^b30 mol %.



Using the determined optimal conditions, the scope of reaction with respect to azides 2 was investigated (Table 2). Benzylic azides 2b and 2c bearing methyl and methoxy groups on 4-positions successfully underwent copper-catalyzed [3 + 2] annulation with 1 to afford corresponding 1-benzylic 4fluorotriazoles 3b and 3c in 87% and 69% yields, respectively. Because the vinyl group was tolerated under the conditions used, benzylic triazole 3d bearing a vinyl group was synthesized in 81% yield. The annulation of benzylic azides 2e-2gbearing a chlorine substituent at para, meta, or ortho positions proceeded effectively regardless of the position of the substituent, which led to corresponding triazoles 3h-3k in 82%, 84%, and 82% yields, respectively. Although the reactions of aryl azides 2h-2k required 2.5 equiv of (difluorovinyl)zinc complex 1, triazoles 3h-3k bearing aryl ether, alkyl ether, ester, and amide moieties were obtained in good to high yields. Not only allylic azide 2l but also alkyl azide 2m participated in [3 + 2] annulation, which led to the synthesis of 1-allylated and 1-alkylated 4-fluorotriazoles 3l and 3m in 81% and 75% yields, respectively.





^aIsolated yield. ^b1 (2.5 equiv) was used. ^c1 (2.5 equiv), CuCl (30 mol %), and 1,10-phen (30 mol %) were used.

We assumed that the [3 + 2] annulation began with transmetalation between (difluorovinyl)zinc complex 1 and the copper(I) salt, which probably generated (difluorovinyl)copper A (Scheme 2, path a). Fokin advocated a mechanism involving two copper components for the Huisgen reaction of azides with terminal alkynes.¹⁰ In his mechanism, the two copper components initially activate the alkyne moiety and then promote the formation of intermediary metalacycles in an oxidative cyclization-like manner. Thus, the in situ-generated (difluorovinyl)copper A would undergo the oxidative cyclization-reductive elimination sequence with the aid of another copper, followed by β -fluorine elimination to afford **3**. There remains another possible pathway involving fluoroacetylene B generated via β-fluorine elimination at an early stage. The protonation of triazolyl copper C probably proceeds with B as the final step (Scheme 2, path b).





To gain mechanistic insight, a competition experiment was conducted using deuterated (triisopropylsilyl)acetylene 5-*d* (D/H = 99/1). The treatment of azide 2a with (difluorovinyl)zinc complex 1 in the presence of the copper catalyst and 5-*d* afforded fluorinated triazole 3a (D/H = <1/>99) derived from 1, and nonfluorinated triazole 6 (D/H = 24/76) derived from 5-*d*; 5-deuterated 3a was not observed (Scheme 3). If the annulation of 2a with 1 involves the protonation step (from C to 3), a substantial H/D scrambling would be observed via deuteration with 5-*d*. Thus, this result suggests that the [3 + 2] annulation of azides 2 with (difluorovinyl)zinc complex 1 probably does not proceed via the generation of fluoroacetylene **B**.

Scheme 3. Competition Experiment Using Deuterated Acetylene 5-*d*



In addition, the reaction of fluoroacetylene **B**, prepared via the lithiation of 1,1-difluoroethylene (**4**) followed by β fluorine elimination,^{5,6} was examined. After the treatment of **4** with *sec*-BuLi at -100 °C, the reaction mixture was kept at -60 °C for 2 h, and then the temperature was increased to room temperature (Scheme 4). After stirring at room temperature for another 30 min, fluoroacetylene **B** was obtained as a THF–ether solution but only in 26% yield, which was characterized by ¹⁹F NMR spectroscopy (δ -17.8 ppm relative to C₆F₆, singlet).¹¹ Additionally, when the obtained fluoroacetylene **B** in solution was treated with azide **2a** in the presence of the copper catalyst, triazole **3a** was obtained but only in 71% isolated yield (Scheme 4). Although this reaction apparently proceeds via the [3 + 2] annulation of azides **2** with **B**, the efficiency of the entire reaction remains quite low (<20% from **4**) and not practical. Moreover, the decrease in yield (71%) compared to 86% (Table 1, Entry 12) supports the fluoroacety-lene-free mechanism in the [3 + 2] annulation of **2** with **1** (Scheme 2, path a).

Scheme 4. Generation of Fluoroacetylene B and Its Reaction with Azide 2a



Further chemical transformations of triazoles 3 were examined. Because triazole rings are known to serve as a directing group,¹² the constructed fluorotriazole rings were applied to promote the rhodium-catalyzed C-H bond activation of a benzene ring on triazole nitrogen according to the annulation using pyrazole derivatives reported by Miura and Satoh (Scheme 5).¹³ In the presence of $[Cp*RhCl_2]_2$ as a catalyst and $Cu(OAc)_2 \cdot H_2O$ as an oxidant, the [5 + 2] annulation of 1benzylated 4-fluorotriazole 3a with diarylacetylene 7 effectively proceeded via domino C-H bond activation to afford azepine derivative 8a in 57% yield. Similarly, 1-arylated 4fluorotriazole **3i** underwent [4 + 2] annulation under the same conditions to afford pyridine derivative 8i in 87% yield. Thus, fluorine-containing tricyclic compounds were readily synthesized by the combination of the copper-catalyzed [3 + 2] annulation and rhodium-catalyzed [5+2]/[4+2] annulation.

Scheme 5. Annulation of Triazoles **3** with Alkyne **7** via the Rhodium-Catalyzed Domino C–H Bond Activation



In summary, we achieved the copper-catalyzed [3 + 2] annulation of azides with (2,2-difluorovinyl)zinc chloride– TMEDA. We offer a facile and practical method for the use of a fluoroacetylene equivalent, which has been considered to be highly reactive and difficult to handle and control for synthetic applications. Of note, the protocol provides an efficient method for the synthesis of 4-fluorotriazoles bearing a wide variety of substituents on the 1-position as promising candidates for pharmaceuticals and agrochemicals, which are difficult to prepare by conventional methods.^{14,15}

ASSOCIATED CONTENT

Supporting Information

Experimental details, characterization data, and NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*E-mail: junji@chem.tsukuba.ac.jp.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

This work was financially supported by JSPS KAKENHI Grant Number JP19H02707 (J.I.) in Grant-in-Aid for Scientific Research (B), JSPS KAKENHI Grant Number JP18H04234 (J.I.) in Precisely Designed Catalysts with Customized Scaffolding, JSPS KAKENHI Grant Number JP20K21186 (J.I.) in Grant-in-Aid for Challenging Research (Exploratory), and JSPS KAKENHI Grant Number JP18K05116 (T.F.) in Grant-in-Aid for Scientific Research (C). This study was also supported by the Cooperative Research Program of Institute for Catalysis, Hokkaido University (Grant #20A1005). We thank Prof. M. Nishida (Hokkaido University) for a fruitful discussion. We acknowledge TOSOH FINECHEM CORPORATION for a generous gift of 1,1difluoroethylene.

REFERENCES

- For general chemical transformations of fluorine-containing compounds, see: (a) Banks, R. E.; Smart, B. E.; Tatlow, J. C. Organofluorine Chemistry, Principles and Commercial Applications, Plenum Press, New York, 1994. (b) Uneyama, K. Organofluorine Chemistry, Blackwell Publishing, Oxford, 2006. (c) Bégué, J.-P.; Bonnet-Delpon, D. Bioorganic and Medicinal Chemistry of Fluorine, John Wiley & Sons, Hoboken, 2008. (d) Nenajdenko, V. ed. Fluorine in Heterocyclic Chemistry vol. 1 and 2, Springer, Heidelberg, 2014.
- (2) For selected recent reviews on chemical transformations of fluorine-containing compounds via C-F bond activation, see: (a) Amii, H.; Uneyama, K. C-F Bond Activation in Organic Synthesis. Chem. Rev. 2009, 109, 2119-2183. (b) Stahl, T.; Klare, H. F. T.; Oestreich, M. Main-Group Lewis Acids for C-F Bond Activation. ACS Catal. 2013, 3, 1578-1587. (c) Ahrens, T.; Kohlmann, J.; Ahrens, M.; Braun, T. Functionalization of Fluorinated Molecules by Transition-Metal-Mediated C-F Bond Activation To Access Fluorinated Building Blocks. Chem. Rev. 2015, 115, 931-972. (d) Unzner, T. A.; Magauer, T. Carbon-Fluorine Bond Activation for the Synthesis of Functionalized Molecules. Tetrahedron Lett. 2015, 56, 877-883. (e) Shen, Q.; Huang, Y.-G.; Liu, C.; Xiao, J.-C.; Chen, Q.-Y.; Guo, Y. Review of Recent Advances in C-F Bond Activation of Aliphatic Fluorides. J. Fluorine Chem. 2015, 179, 14-22. (f) Zhang, X.; Cao, S. Recent Advances in the Synthesis and C-F Functionalization of gem-Difluoroalkenes. Tetrahedron Lett. 2017, 58, 375-392. (g) Jaroschik, F. Picking One out of Three: Selective Single C-F Activation in Trifluoromethyl Groups. Chem.-Eur. J. 2018, 24, 14572-14582. (h) Fujita, T.; Fuchibe, K.; Ichikawa, J.

Transition-Metal-Mediated and -Catalyzed C–F Bond Activation by Fluorine Elimination. *Angew. Chem., Int. Ed.* **2019**, *58*, 390–402.

- (3) Middleton, W. J.; Sharkey, W. H. Fluoroacetylene. J. Am. Chem. Soc. 1959, 81, 803–804.
- (4) (a) Yakubovich, A. Y.; Smirnov, K. M.; Dybov, S. S. Synthesis of Vinyl Monomers. Fluoroacetylene; Preparation and Properties. *Khim. Nauka, i. Prom.* 1959, *4*, 551–552. (b) Riemshneider, R.; Weil, L.; Nolde, K. Zur Kenntnis Halogenierter Acetylene, 4. Mitt.; Monofluoracetylen aus Monofluoracetylentetrabromid. *Monatsh. Chem.* 1962, *93*, 952–954. (c) Viehe, H. G.; Franchimont, E. Darstellung und Reaktionen des Monofluoracetylens. *Chem. Ber.* 1962, *95*, 319–327. (d) Smirnov, K. M.; Tomilov, A. P. Preparation of Fluoroacetylene. *Zh. Vses. Khim. Obshchest.* 1974, *19*, 350–351.
- (5) Sauvêtre, R.; Normant, J. F. Une Nouvelle Preparation du Fluoroacetylene — Sa Reaction avec les Organometalliques. Synthese d'Alcynes et d'Enynes Divers. *Tetrahedron Lett.* **1982**, *23*, 4325–4328.
- (6) Tellier, F.; Descoins, C.; Sauvêtre, R. Stereospecific Synthesis of 1,5-Dien-3-ynes and 1,3,5-Trienes Application to the Stereochemical Identification of Trienic Sex Pheromones. *Tetrahedron* 1991, 47, 7767–7774.
- (7) (a) Fujita, T.; Ichitsuka, T.; Fuchibe, K.; Ichikawa, J. Facile Synthesis of β,β-Difluorostyrenes via the Negishi Coupling of Thermally Stable 2,2-Difluorovinyl Zinc–TMEDA Complex. *Chem. Lett.* 2011, 40, 986–988. (b) Ichitsuka, T.; Takanohashi, T.; Fujita, T.; Ichikawa, J. A Versatile Difluorovinylation Method: Cross-Coupling Reactions of the 2,2-Difluorovinylzinc–TMEDA Complex with Alkenyl, Alkynyl, Allyl, and Benzyl Halides. J. Fluorine Chem. 2015, 170, 29– 37.
- (8) For synthetic application of fluoroalkynes, see: (a) Viehe, H. G.; Merényi, D.-I. R.; Oth, J. F. M.; Valange, P. Formation of 1,2,3-Tri-t-butyltrifluorobenzene by Spontaneous Trimerization of t-Butylfluoroacetylene. Angew. Chem., Int. Ed. Engl. 1964, 3, 746. (b) Delavarenne, S. Y.; Viehe, H. G. Directiospezifische Additionen und Substitutionen beim Fluochloracetylen und beim 2-Fluor-1.1-dichlor-äthylen. Chem. Ber. 1970, 103, 1198-1208. (c) Hanamoto, T.; Koga, Y.; Kawanami, T.; Furuno, H.; Inanaga, J. Crystal Structure of a Dewar Benzene Derivative Formed from Fluoro(triisopropylsilyl)acetylene. Angew. Chem., Int. Ed. 2004, 43, 3582-3584. (d) Meiresonne, T.; Verniest, G.; De Kimpe, N.; Mangelinckx, S. Synthesis of 2-Fluoro-1,4-benzoxazines and 2-Fluoro-1,4-benzoxazepin-5-ones by Exploring the Nucleophilic Vinylic Substitution (S_NV) Reaction of gem-Difluoroenamides. J. Org. Chem. 2015, 80, 5111-5124.
- (9) For selected reviews on the Huisgen reaction, see: (a) Huisgen, R. 1,3-Dipolar Cycloadditions. Proc. Chem. Soc. 1961, 357-396. (b) Huisgen, R. 1,3-Dipolar Cycloadditions Past and Future. Angew. Chem., Int. Ed. Engl. 1963, 2, 565-632. (c) Moses, J. E.; Moorhouse, A. D. The Growing Applications of Click Chemistry. Chem. Soc. Rev. 2007, 36, 1249-1262. (d) Amblard, F.; Cho, J. H.; Schinazi, R. F. Cu(I)-Catalyzed Huisgen Azide-Alkyne 1,3-Dipolar Cycloaddition Reaction in Nucleoside, Nucleotide, and Oligonucleotide Chemistry. Chem. Rev. 2009, 109, 4207-4220. (e) Kappe, C. O.; Van der Eycken, E. Click Chemistry under Non-classical Reaction Conditions Chem. Rev. 2010, 39, 1280-1290. (f) Kacprzak, K.; Skiera, I.; Piasecka, M.; Paryzek, Z. Alkaloids and Isoprenoids Modification by Copper(I)-Catalyzed Huisgen 1,3-Dipolar Cycloaddition (Click Chemistry): Toward New Functions and Molecular Architectures. Chem. Rev. 2016, 116, 5689-5743.
- (10) For mechanistic studies on the Huisgen reaction, see: Worrell, B. T.; Malik, J. A.; Fokin, V. V. Direct Evidence of a Dinuclear Copper Intermediate in Cu(I)-Catalyzed Azide–Alkyne Cycloadditions *Science* **2013**, *340*, 457–460.

- (11) The value of ¹⁹F NMR of fluoroacetylene **B** was comparable to that of fluoro(triisopropylsilyl)acetylene (δ ca. -24 ppm relative to C₆F₆). See 8c.
- (12) Guerrero, I.; Correa, A. Metal-Catalyzed C-H Functionalization Processes with "Click"-Triazole Assistance. *Eur. J. Org. Chem.* 2018, 6034–6049, and references cited therein.
- (13) (a) Umeda, N.; Tsurugi, H.; Satoh, T.; Miura, M. Fluorescent Naphthyl- and Anthrylazoles from the Catalytic Coupling of Phenylazoles with Internal Alkynes through the Cleavage of Multiple C–H Bonds. *Angew. Chem., Int. Ed.* 2008, 47, 4019– 4022. (b) Umeda, N.; Hirano, K.; Satoh, T.; Shibata, N.; Sato, H.; Miura, M. Rhodium-Catalyzed Oxidative 1:1, 1:2, and 1:4 Coupling Reactions of Phenylazoles with Internal Alkynes through the Regioselective Cleavages of Multiple C–H Bonds. *J. Org. Chem.* 2010, *76*, 13–24. See also: (c) Qi, Z.; Yu, S.; Li, X. Rh(III)-Catalyzed Oxidative Annulation of 2-Phenylimidazo[1,2-a]pyridines with Alkynes: Mono versus Double C–H Activation. *J. Org. Chem.* 2015, *80*, 3471–3479.
- (14) For drugs containing a triazole ring, see: (a) Arroyo, S. Rufinamide. *Neurotherapeutics* 2007, *4*, 155–162. (b) Gin, A.; Dilay, L.; Karlowsky, J. A.; Walkty, A.; Rubinstein, E.; Zhanel, G. G. Piperacillin–Tazobactam: a β-Lactam/β-

Lactamase Inhibitor Combination. *Expert. Rev. Anti-infect. Ther.* **2007**, *5*, 365–383.

(15) For the synthesis of ring-fluorinated triazoles, see: (a) Worrell, B. T.; Hein, J. E.; Fokin, V. V. Halogen Exchange (Halex) Reaction of 5-Iodo-1,2,3-triazoles: Synthesis and Applications of 5-Fluorotriazoles. *Angew. Chem., Int. Ed.* 2012, *51*, 11791–11794. (b) Wang, D.; Sun, W.; Chu, T. Synthesis of 5-Fluorotriazoles by Silver-Mediated Fluorination of 5-Iodotriazoles. *Eur. J. Org. Chem.* 2015, 4114–4118. (c) Motornov, V. A.; Tabolin, A. A.; Novikov, R. A.; Nelyubina, Y. V.; Ioffe, S. L.; Smolyar, I. V.; Nenajdenko, V. G. Synthesis and Regioselective N-2 Functionalization of 4-Fluoro-5-aryl-1,2,3-NH-triazoles. *Eur. J. Org. Chem.* 2017, 6851–6860. (d) Jana, S.; Adhikari, S.; Cox, M. R.; Roy, S. Regioselective Synthesis of 4-Fluoro-1,5-disubstituted-1,2,3-triazoles from Synthetic Surrogates of α-Fluoroalkynes. *Chem. Commun.* 2020, *56*, 1871–1874.