

An Attempt of Migita-Kosugi-Stille Type Polycondensation at Room Temperature



Polymer Synthesis

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ABSTRACT

In this research, preparation of conjugated polymers was attempted with Migita-Kosugi-Stille cross coupling type polycondensation at room temperature. The polymers thus synthesized in this study were characterized with infrared absorption spectroscopy (IR), UV-vis absorption spectroscopy (UV-vis) and fluorescence spectroscopy.

Keywords: Conjugated polymer, Migita-Kosugi-Stille cross coupling

Introduction

Ito and Shirakawa et al. discovered synthesis of polyacetylene film in 1975¹. After that, many types of conductive polymers have been synthesized. Conductive polymers are now applied for organic solar cells, transistors, organic light-emitting diodes²⁻⁴. Migita-Kosugi-Stille coupling is one of the effective reactions to synthesize conjugated polymers as precursor of conducting polymers⁵.

In this research, we attempted to synthesize conjugated polymers by Migita-Kosugi-Stille coupling at room temperature. This reaction has been usually conducted at high temperature.

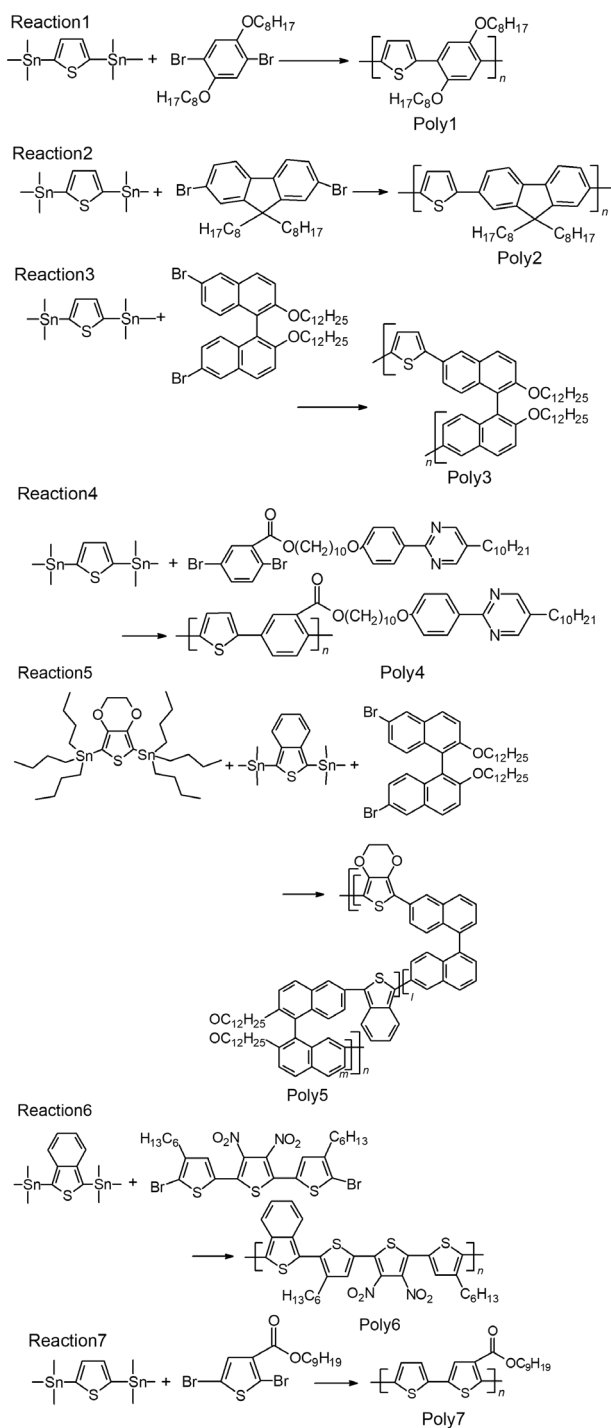
Experimental section

Migita-Kosugi-Stille type polycondensation reactions in the presence of a palladium catalyst were carried out (Scheme 1). The monomers were synthesized in previous research⁶. Dialkylstannylated monomers (monoA), dihalided monomers (monoB), bis(triphenylphosphine)palladium(II) dichloride (0.07 g) and iodine (0.02 g) were dissolved in tetrahydrofuran (0.5 mL). The mixture was stirred at room temperature (~25 °C) overnight. The mixtures

were poured into a large volume of methanol. The precipitates were centrifuged and the solvent was removed under reduced pressure to afford polymers. Molecular weights were < 1000 g/mol from the results of gel permeation chromatography in tetrahydrofuran against polystyrene calibration at room temperature.

Table 1 Synthesis.

Entry	monoA	monoB	Yield
Reaction1	0.05 g	0.05 g	8.4 mg, 19.9 %
Reaction2	0.05 g	0.05 g	2.7 mg, 6.2 %
Reaction3	0.05 g	0.05 g	4.3 mg, 9.5 %
Reaction4	0.028 g	0.05 g	17 mg, 38.7%
Reaction5	0.05 g, 0.05 g	0.05 g	8.3 mg
Reaction6	0.05 g	0.05 g	33 mg, 69.2 %
Reaction7	0.07 g	0.07 g	16 mg, 18.3 %



Scheme 1. Synthesis.

Results and discussion

IR spectroscopy

Chemical structure of the polymers was confirmed with the infrared (IR) spectra (Figure 1). Table 2 shows assignments of IR spectra. The C–H vibrations were observed at ca. 2900 cm^{-1} for all the polymers. The C–H bending mode connected to the α carbon of the thiophene ring and $\delta_{\text{C-S-C}}$ was shown for the monomers at ca. 900 cm^{-1} . While, the signals

derived from this were not detected for the products, indicating the reactions were progressed.

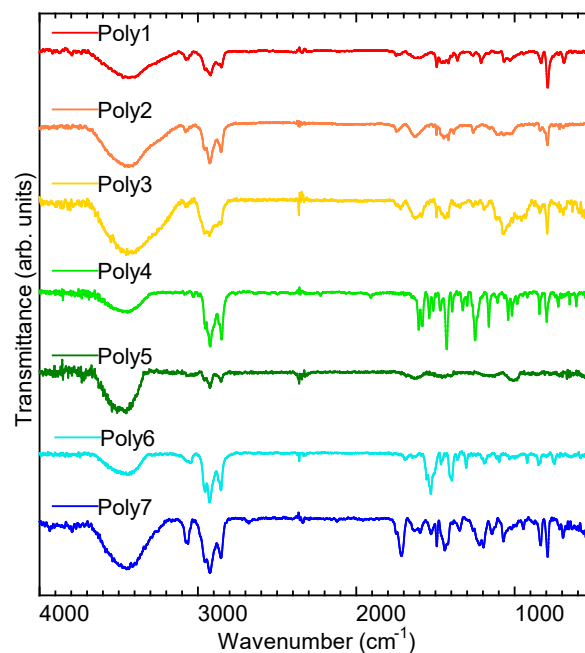


Figure 1. Infrared absorption spectra.

Table 2 Assignments of infrared (IR) spectra for the polymers.

Entry	$\nu_{\text{C=O}}$	$\nu_{\text{C=N}}$	ν_{NO_2}	$\nu_{\text{C=C}}$	$\delta_{\text{C-H}}$	$\nu_{\text{C-O}}$
Poly1	–	–	–	1492	1361	1211 1066
Poly2	–	–	–	1493	1384	–
Poly3	–	–	–	1493	–	1075
Poly4	1605	1583	–	1514 1470	1394	1249 1040
Poly5	–	–	–	–	–	1000
Poly6	–	–	1529 1306	1464	1396	–
Poly7	1606	–	–	1493	1382	–

ν : stretching vibration, δ : bending vibration

Optical properties

UV-vis and fluorescence spectra of the polymers were obtained in tetrahydrofuran solution (Figure 4).

All the polymers show absorption bands from 390 to 430 nm derived from π – π^* transition main chains.

Fluorescence spectroscopy measurements were carried out upon irradiation of an excitation light at 400 nm.

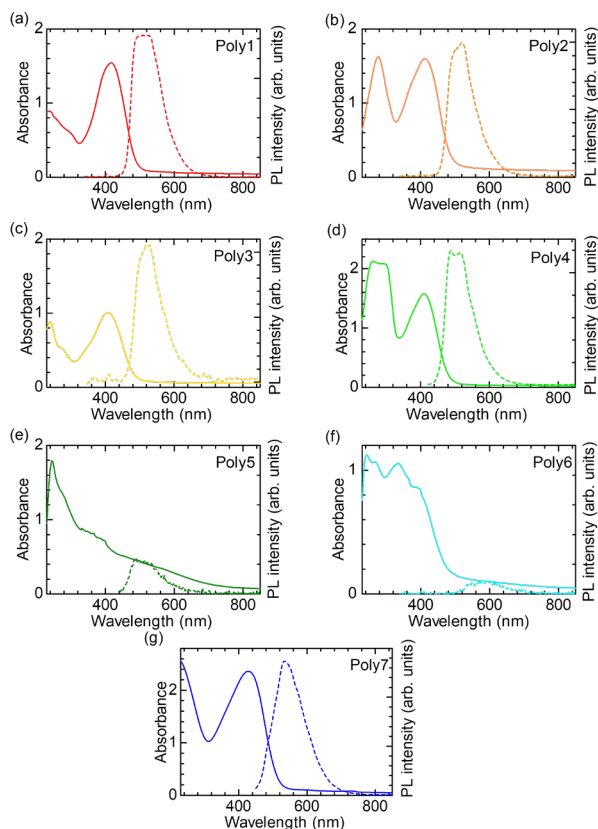


Figure 4. UV-vis spectra (solid line) and fluorescence spectra (PL, dashed line) of Poly1–Poly7 in tetrahydrofuran solution.

Table 3. UV-vis absorption and PL spectroscopy measurement results.

Entry	Absorption (nm)		PL (nm)
	Mru	Main chain	
Poly1	–	417	508
Poly2	277	412	521
Poly3	240	408	525
Poly4	261 300	412	489
Poly5	245	ca. 390	ca. 484
Poly6	242 333	397	ca. 479
Poly7	–	429	532

Mru: monomer repeat unit.

Conclusions

Migita-Kosugi-Stille type cross coupling polycondensation reactions were performed at room temperature. Although the molecular weights were

somewhat low, the resultants prepared at room temperature as a simple and convenient method, showing beautiful colors caused by conjugated skeleton. A series of the products have possibility to apply opt-functional polymeric dyes.

Acknowledgement

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