 氏 名 算 位 の 種 男 位 記 番 号 位 授 与 年 月 円 ((Jing-Jing Ye 博 士 (理学) 博 甲 第 10084 号 令和 3年 9月 24日 学位規則第4条第1項該当 数理物質科学研究科
Studies on the prepa	aration of organophosphines via sodium organophosphides
(ナトリウムオルガノ	ホスファイドを用いる有機ホスフィン類の合成に関する研究)
主查	筑波大学教授(連携大学院) 博士(工学) 韓 立彪
副查查	筑波大学教授 理学博士 市川淳士
副查查	筑波大学教授 博士(理学) 二瓶雅之
副查查	筑波大学准教授(連携大学院) 博士(工学) 中島裕美子

論文の要旨

Ms. Jing-Jing Ye's thesis described her achievements gained in studies on the development of new synthetic methods for organophosphines. Organophosphines are important reagents which widely used in organic synthesis as catalysts and ligands. For example, tertiary phosphines are widely used in many famous synthetic reactions as reagents in preparing vitamins, pharmaceuticals, and agrochemicals. They are also organocatalysts that can trig a variety of organic reactions. Moreover, they are excellent ligands for transition metal in catalysis reactions, especially, chiral phosphines that play a pivotal role in several asymmetric transition-metal-catalyzed reactions. Despite the importance, general and efficient methods for their synthesis are limited. In the industry, tertiary phosphines R₃P are generally produced either by the addition of PH₃ to olefins or by the nucleophilic substitution reactions of RM with phosphorus trichloride. In the laboratory, organophosphines are prepared by metal-catalyzed couplings of ArX with diarylphosphines Ar₂PH or nucleophilic substitution reactions of lithium phosphides Ar₂PLi with alkyl halides RX. However, these methods lack efficiency. Moreover, the starting materials used are toxic, bad-smelling, or difficult to prepare or handle. To precisely and efficiently generate organophosphines, the applicant has developed new methods for the generation of tertiary phosphines, a polymeric phosphine and BINAP derivatives by using sodium phosphides efficiently generated using SD.

Firstly, the applicant disclosed that sodium reacts more accurately and selectively than other alkali metals (Li and K) to generate R₂PNa via the cleavage of the P-Ph bond of R₂PPh. For example, by adding SD (sodium finely dispersed in mineral oil) to a THF solution of Ph₃P at room temperature, the solution turned to brown instantly, and after 0.5 h, Ph₂PNa was generated nearly quantitatively. This sodium phosphide generated was stable for several weeks under nitrogen at room temperature. A series of sodium phosphides

could be conveniently and accurately prepared similarly by using SD. Thus, chlorophosphines (Ph₂PCl, (-)-MenPhPCl and *n*-Bu₂PCl) and diphenylphosphinites (Ph₂POPh and Ph₂POEt) all readily reacted with SD to generate the corresponding sodium phosphides in quantitative yields. Triorganyl phosphines R₂PPh could also be employed as the substrates to efficiently produce R₂PNa via the selective cleavage of the Ph–P bonds by SD. The cleavage of Ar-P bonds by SD is substituent-depending, and that with an electron-withdrawing group was easier to be cleft. As to the selectivity of the cleavage of an aryl Ar-P bond vs. an alkyl R-P bond, it was found that the selectivity could be explained by considering the stability of the radical anion intermediates involved. The cleavage of the C-P bonds by an alkali metal was proposed to take place via a SET mechanism.

Secondly, the applicant developed a new method to efficiently produce organophosphines by using aryl chlorides. First, the coupling reactions of phenyl halides PhX with Ph₂PNa generating Ph₃P were studied. It was found that both phenyl bromide and phenyl iodide gave low yields of the coupling product Ph₃P, while phenyl chloride gave high yield of Ph₃P. Therefore, the generally recognized inert PhCl, not the more reactive PhBr and PhI, was the chemical of choice for this reaction. The reaction was general and practicable. A series of aryl chlorides, with electron-donating groups or electron-withdrawing groups at the *para*-position of the chlorobenzene rings, all were readily phosphinated by Ph₂PNa, affording the corresponding organophosphines in high to excellent yields. In addition, chloronaphthalenes, heteroaryl chlorides, chloropyridines, chlorothiophenes and multi-chloroarenes were tolerable under similar conditions to give the expected phosphines in good to high yields. The couplings of ArCl with Ph₂PNa generating Ph₂PAr in high yields took place via a unique S_{RN}1 mechanism. Dichloroethene and aliphatic chlorides could be used directly in the coupling reactions with Ph₂PNa. Similarly, the chiral menthyl chlorides reacted with Ph₂PNa to give the corresponding chiral phosphines via S_N2 mechanism. To further elucidate the generality of the present method for the preparation of organophosphines, the reactions of other sodium phosphides (PhMePNa, Phn-BuPNa, etc.) with any chlorides were also studied. The reactivity of R₂PNa was strongly affected by the R substituent.

Thirdly, a novel poly(vinyldiphenylphosphine), that is soluble in common solvents, was prepared from the reactions of Ph_2PNa with the cheap poly(vinyl chloride) as an application of the method for the preparation of organophosphines using Ph_2PNa . By using the newly generated poly(vinyldiphenylphosphine) in Wittig reactions, the corresponding aromatic and aliphatic olefins were readily synthesized in good yields. The use of poly(vinyldiphenylphosphine) in Wittig reactions could overcome the shortcomings of the classical Wittig reactions using R_3P . Because the corresponding by-products tertiary phosphines oxides $R_3P(O)$ were soluble in organic solvents, they were difficult to remove from the products. On the other hand, because poly(vinyldiphenylphosphine oxide) was insoluble in common solvents, it could be easily removed by simple filtration. Moreover, comparing to the insoluble polymer-supported phosphines used in Wittig reactions, the newly prepared poly(vinyldiphenylphosphine) was more reactive and economic.

Finally, the applicant also studied the modification of BINAP using SD. By controlling the reaction conditions, both the corresponding cyclic binaphthylphospholes and binaphthyl-based phosphines BINAPs

could be generated, respectively. Early literatures reported that under similar conditions, lithium only generated cyclic binaphthylphospholes. Therefore, sodium was superior to lithium because sodium could not only cleave BINAP to generate the corresponding cyclic sodium phosphide, but also give binaphthyl-based disodium phosphides via the selective cleavage of P–C bonds.

審査の要旨

In the thesis, the applicant established new methods for the preparations of diaryl- or triaryl phosphines, poly(vinylphosphine), cyclic binaphthyl-based phosphole and BINAP derivatives via the reactions of sodium phosphides and organohalides. Firstly, a series of sodium phosphides R¹R²PNa were prepared from R¹R²PCl, R¹R²POR, R¹R²PAr via the selective cleavage of P-Cl bond, P-O bond, or P-C bond by using SD. Then, a series of unsymmetrical and asymmetrical phosphines were further synthesized by treating these prepared sodium phosphides with alkyl halides, aryl chlorides and polymer chloride. Besides, BINAP could also react with SD to general binaphthyl-based disodium phosphides or cyclic sodium phosphide by controlling the reaction conditions and delivering the corresponding BINAP derivatives and cyclic dinapthylphosphospholes, respectively. The applicant cited the backgrounds of the study correctly, conducted the research logically, and analyzed the results reliably, as evidenced by her two papers published in renowned international journals.

〔最終試験結果〕

[批評]

The final examination committee conducted a meeting as a final examination on July 14, 2021. The applicant provided an overview of dissertation, addressed questions and comments raised during Q&A session. All of the committee members reached a final decision that the applicant has passed the final examination.

[結論]

Based on the results of the above-mentioned dissertation defense and final examination, the final examination committee approved that the applicant is qualified to be awarded Doctor of Philosophy in Science.