

## Vynylon Prepared in Spring Water and Vynylon/Polyethylenedioxythiophene Composite



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### ABSTRACT

In this research, we prepared vynylon synthesized in spring water from Zao. Furthermore, vynylon/poly(3,4-ethylenedioxythiophene) (PEDOT) composite was prepared with vynylon synthesized in spring water. The compounds thus synthesized in this study were evaluated with infrared absorption (IR) spectroscopy measurements and scanning electron microscopy (SEM) observations.

**Keywords:** Composite, conductive polymer, spring water.

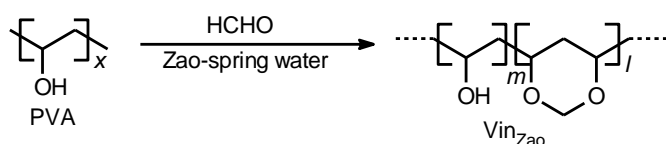
### Introduction

Hot spring water contains many kinds of minerals like Ca, Mg, Al, and ions such as  $\text{SO}_4^{2-}$ ,  $\text{Cl}^-$ , and  $\text{HSO}_4^-$  [1]. pH of the hot spring water depends on place and time [2].

In this research, vynylon was synthesized in the spring water obtained from Zao, Japan. Spring water was used as both acid catalyst and solvent during the reaction. Furthermore, vynylon/poly(3,4-ethylenedioxythiophene) (PEDOT, a conducting polymer) composite was synthesized with vynylon thus prepared in this study.

### Experimental

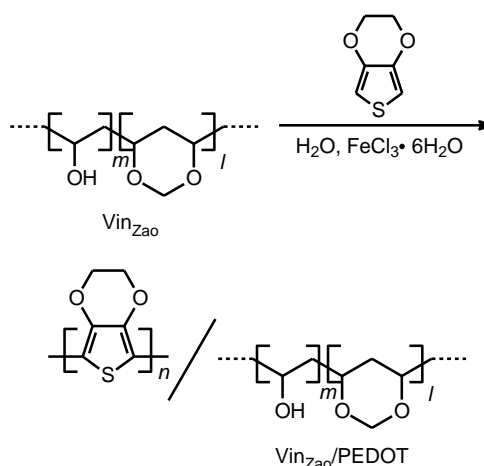
#### Synthesis



**Scheme 1.** Synthesis of  $\text{Vin}_{\text{Zao}}$  (vynylon) in the presence of Zao-spring water. PVA: polyvinylalcohol.

First, vynylon was synthesized in Zao-spring water (pH = 2.0, 60 mL) from polyvinylalcohol (PVA,

2.0 g) with a small volume of formaldehyde (Scheme 1). Vynylon was first synthesized by Sakurada in 1939. Polyvinylalcohol was formalized with  $\text{H}_2\text{SO}_4$ ,  $\text{Na}_2\text{SO}_4$ ,  $\text{HCHO}$  and  $\text{H}_2\text{O}$  in the study.  $\text{H}_2\text{SO}_4$  was used as an acid catalyst.  $\text{Na}_2\text{SO}_4$  was used to prevent softening of fiber by formalization.  $\text{H}_2\text{O}$  was employed as a solvent [3]. Therefore, the acid component of the Zao-spring water played a role of acid catalyst and a solvent.



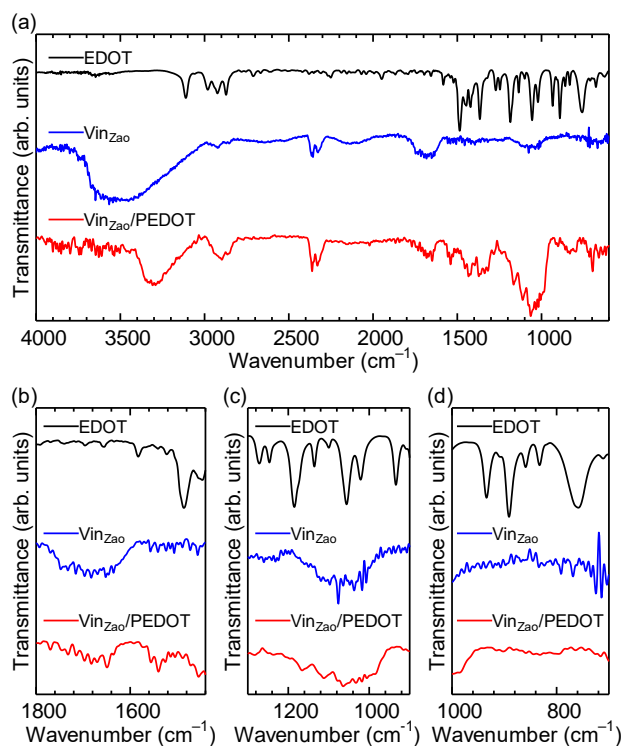
**Scheme 2.** Synthesis of  $\text{Vin}_{\text{Zao}}$ /PEDOT. PEDOT: poly(3,4-ethylenedioxythiophene).

This polymer (vinyon) is abbreviated as  $\text{Vin}_{\text{Zao}}$ . Then,  $\text{Vin}_{\text{Zao}}$  thus prepared (0.02 g) and 3,4-ethylenedioxythiophene (EDOT, 0.2 g) were added into the distilled water (1 mL) and stirred. After two days,  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  (0.2 g) was added into the mixture over night (Scheme 2). The precipitate was collected by filtration and the solvent was removed under vacuum to afford a black solid.

## Results and discussion

### Molecular Structure

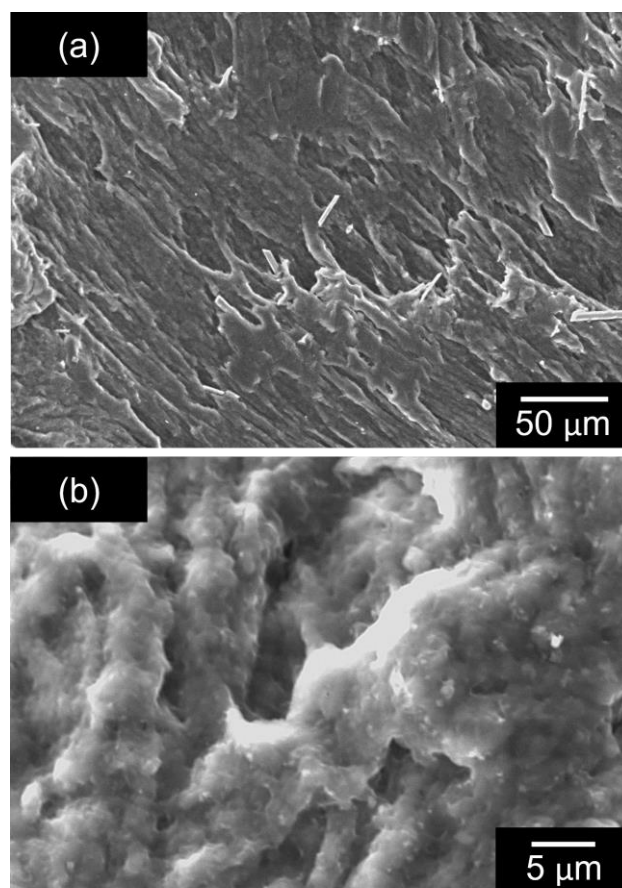
Figure 1 shows infrared absorption (IR) spectra for EDOT (monomer),  $\text{Vin}_{\text{Zao}}$  and  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  composite. EDOT exhibited bending vibration of C–H in aromatic ring with  $\alpha$  carbon at  $890\text{ cm}^{-1}$ . While,  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  showed no signal derived from the monomer at  $890\text{ cm}^{-1}$ .  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  shows C=O and C=C stretching vibration at  $1697$  and  $1539\text{ cm}^{-1}$  due to the vinyon structure, respectively. The signal at  $1150\text{ cm}^{-1}$  and  $1165\text{ cm}^{-1}$  is derived from the C–O–C stretching vibration of vinyon in  $\text{Vin}_{\text{Zao}}$  and  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$ , respectively. These results confirm that the synthesis of  $\text{Vin}_{\text{Zao}}$  and  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  was achieved.



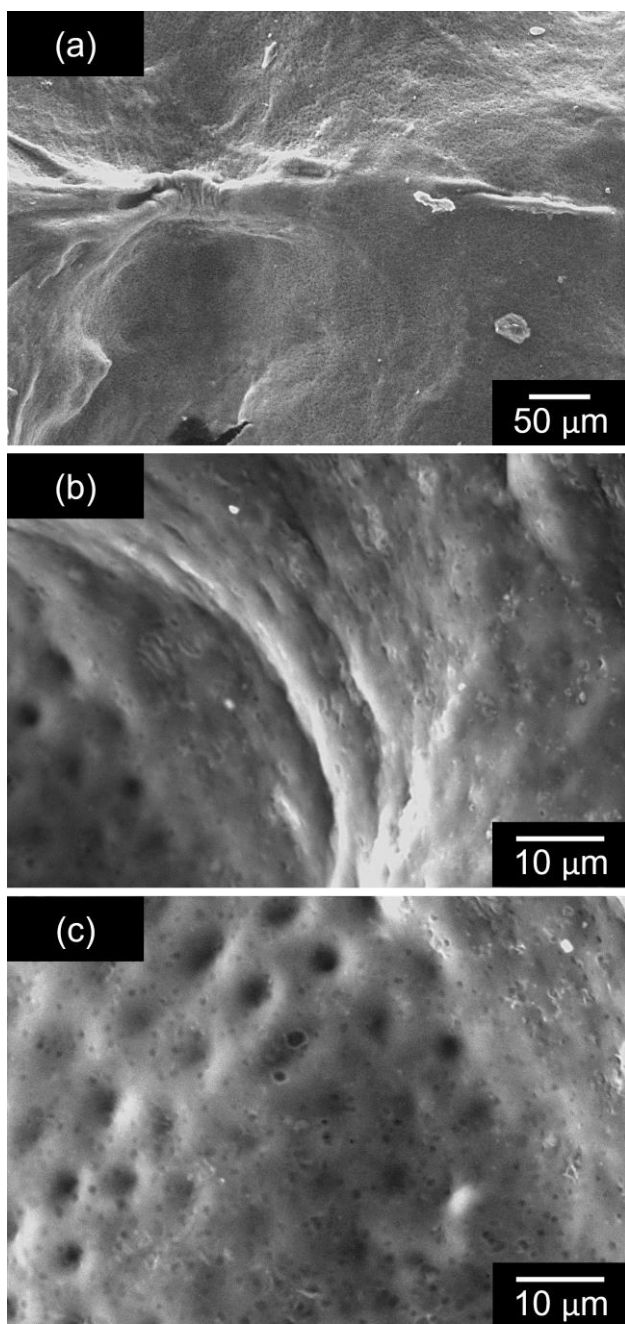
**Figure 1.** Infrared absorption spectra for 3,4-ethylenedioxythiophene (EDOT, monomer),  $\text{Vin}_{\text{Zao}}$  and  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$ . (a): full scale. (b):  $1800 - 1440\text{ cm}^{-1}$ . (c):  $1300 - 900\text{ cm}^{-1}$ . (d):  $1000 - 700\text{ cm}^{-1}$ .  $\text{Vin}_{\text{Zao}}$ : vinyon prepared in Zao-spring water.

### Surface images

Figures 2,3 show SEM images of non-Au coated  $\text{Vin}_{\text{Zao}}$  and  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$ , respectively.  $\text{Vin}_{\text{Zao}}$  has roughness structure (Figure 2a,b). While,  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  displays less roughness (Figure 3a). Magnification image of  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  evaluated that  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$  has porous structure on the surface (Figure 3b,c).



**Figure 2.** Scanning electron microscopy (SEM) images of  $\text{Vin}_{\text{Zao}}$ . (a): Low magnification. (b): High magnification.



**Figure 3.** SEM images of  $\text{Vin}_{\text{Zao}}/\text{PEDOT}$ . (a): Low magnification. (b,c): High magnification.

### Conclusions

Preparation of vinylon synthesized in Zao-spring water, and composite formation of the vinylon with PEDOT were successfully achieved. PEDOT as a conducting polymer was coated the entire surface of the vinylon.

### Acknowledgements

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### References

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