

1 **Steady State Characteristics of Acclimated Hydrogenotrophic**  
2 **Methanogens on Inorganic Substrate in Continuous Chemostat**  
3 **Reactors**

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1 **Abstract**

2 A Monod model has been used to describe the steady state characteristics of the  
3 acclimated mesophilic hydrogenotrophic methanogens in experimental chemostat  
4 reactors. The bacteria were fed with mineral salts and specific trace metals and a  
5 H<sub>2</sub>/CO<sub>2</sub> supply was used as a single limited substrate. Under steady state  
6 conditions, the growth yield ( $Y_{CH_4}$ ) reached 11.66 g cells per mmol of H<sub>2</sub>/CO<sub>2</sub>  
7 consumed. The daily cells generation average was  $5.67 \times 10^{11}$ ,  $5.25 \times 10^{11}$ ,  
8  $4.2 \times 10^{11}$  and  $2.1 \times 10^{11}$  cells/ L-culture for the dilutions 0.071/d, 0.083/d, 0.1/d  
9 and 0.125/d, respectively. The maximum specific growth rate ( $\mu_{max}$ ) and the  
10 Monod half-saturation coefficient ( $K_S$ ) were 0.15/d and 0.82 g/L, respectively.  
11 Using these results, the reactor performance was simulated. During the steady  
12 state, the simulation predicts the dependence of the H<sub>2</sub>/CO<sub>2</sub> concentration ( $S$ ) and  
13 the cell concentration ( $X$ ) on the dilution rate. The model fitted the experimental  
14 data well and was able to yield a maximum methanogenic activity of 0.24 L  
15 CH<sub>4</sub>/g VSS·d. The dilution rate was estimated to be 0.1/d. At the dilution rate  
16 of 0.14/d, the exponential cells washout was achieved.

17 **Keywords:** Hydrogenotrophic methanogens; chemostat; steady state; H<sub>2</sub>/CO<sub>2</sub> gas;  
18 simulation; trace metals

## 1 **1. Introduction**

2           Extensive research has demonstrated the capability of methanogenic  
3 bacteria to grow on molecular hydrogen (H<sub>2</sub>) and utilize carbon dioxide (CO<sub>2</sub>) as  
4 an energy source (Daniels et al., 1984; Jones et al., 1985). The methanogens are  
5 inhabitants of the anaerobic fermentor, which is currently regarded as an attractive  
6 process for the degradation of various types of organic matter into methane and  
7 other by-products.

8           The construction of a mathematical model is obviously a crucial step in  
9 the design, control and operation of a fermentor (Bastone, 2006). Based on  
10 engineering literature, several models have been proposed to simulate and  
11 improve the fermentor performance through the supply of organic substrates  
12 (Siegrist et al., 2002; Nopharatana et al., 2003). The Monod kinetics model,  
13 which is the simplest approach, relates the specific growth rate of the biomass ( $\mu$ )  
14 (1/d) to the residual substrate concentration ( $S$ ) (g/L).

$$15 \quad \mu = \frac{\mu_{\max} S}{K_S + S} \quad (1)$$

16 where  $\mu_{\max}$  is the maximum specific biomass growth rate and  $K_S$  is the  
17 half-saturation coefficient (g/L). This relationship has been used to model the  
18 single substrate H<sub>2</sub> consumption in the presence of another substrate (Robinson

1 and Tiedje, 1982). It has also been used to model anaerobic acidogenesis  
2 (Demirel and Yenigün, 2002). However, there is no report regarding the use of  
3 the Monod kinetics model for acclimated hydrogenotrophic methanogens under  
4 steady state continuous cultivation with recirculation of the single substrate  
5 H<sub>2</sub>/CO<sub>2</sub>. The chemostat proposed by Tang et al., 1997 was associated with  
6 steady state cultivation that provided a better conversion ratio of H<sub>2</sub>/CO<sub>2</sub> to  
7 methane and carbon dioxide (Valentine et al., 2004). Four chemostat reactors  
8 were used with varying rates of flow of the inorganic medium. The objectives  
9 were, firstly, to attain a steady state chemostat cultivation of H<sub>2</sub>/CO<sub>2</sub>-dependant  
10 methanogens and measure the limiting supply of substrate, then to apply the  
11 Monod relationship to standardize the characteristics of H<sub>2</sub>/CO<sub>2</sub> gas fermentation.  
12 The parameters obtained would provide data for the creation of a generic model  
13 for the dynamic simulation of biogas production from a methanogen culture using  
14 mixed H<sub>2</sub>/CO<sub>2</sub> gas as the substrate.

15

## 16 **2. Materials and Methods**

### 17 *2.1. Acclimation of hydrogenotrophic methanogens*

18 Anaerobic activated sludge (2.5 L) obtained from a domestic waste water

1 treatment plant (Ibaraki, Japan) was acclimated in duplicate 5 L reactors at  
2 mesophilic temperatures for 7 months; H<sub>2</sub>/CO<sub>2</sub> (80:20, v/v) gas was used as  
3 source of energy and carbon. Using a 2 L gas bag, the gas substrate was  
4 recirculated at 0.08 MPa. The archaea cultivation procedure (Sowers and Noll,  
5 1995) was implemented and continued until H<sub>2</sub>/CO<sub>2</sub>-dependant methanogens  
6 predominated the culture. The physical and chemical compositions of the  
7 acclimated sludge obtained were averaged from the 2 reactors and are shown in  
8 Table 1. According to these results, the acetic acid concentration was nil and  
9 56.6% methane was produced. The absence of acetic acid indicates the  
10 possibility that methane production from substrates other than the H<sub>2</sub>/CO<sub>2</sub> gas  
11 mixture can be excluded.

## 12 2.2. *Epifluorescent microscopy analysis*

13 The acclimated sludge samples were filtered using a polycarbonate black  
14 filter with 0.22 µm pore diameter. Using ethidine bromide (Sigma-Aldrich,  
15 Poland) as the stain, the sludge samples were observed under an ultra-microscope  
16 (OLYMPUS, BX 50, Japan). The apparently uniform blue autofluorescence  
17 verified the purity of the culture. A morphological examination of the bacteria  
18 revealed long rod-shaped cells and cocci.

1    2.3. *Experimental procedure for continuous chemostat reactors*

2           A modified version of the medium used by Yang et al. (2004) was  
3 applied; the final composition was as follows: mineral nutrients (in g/L of  
4 demineralized water) including  $K_2HPO_4$ , 3.4;  $KH_2PO_4$ , 3.4;  $Na_2CO_3$ , 2.54;  $NH_4Cl$ ,  
5 2.13; and trace metals (in mg/L of demineralized water) including  $MgSO_4 \cdot 7H_2O$ ,  
6 600;  $MnSO_4 \cdot 5H_2O$ , 5.3;  $NiCl_2 \cdot 6H_2O$ , 125;  $FeSO_4 \cdot 7H_2O$ , 28;  $CoCl_2 \cdot 6H_2O$ , 10;  
7  $ZnSO_4 \cdot 7H_2O$ , 30;  $CuSO_4 \cdot 5H_2O$ , 8;  $AlK(SO_4)_2 \cdot 12H_2O$ , 1;  $CaCl_2 \cdot 2H_2O$ , 40;  
8  $H_3BO_4$ , 1 and  $Na_2MoO_4 \cdot 2H_2O$ , 1.5. The mineral nutrients and trace metals were  
9 prepared anaerobically and the estimated pH values were 7.95 and 6.93,  
10 respectively. The media were stocked in two 5 L Duran vials. In order to  
11 maintain an anaerobic environment, the headspaces were flushed with  $H_2/CO_2$   
12 (80:20, v/v) for 15 min at 0.1 MPa.

13    *Chemostat cultivation:* Using different dilution rates (D), i.e.  $D_1 = 0.071/d$ ,  $D_2 =$   
14  $0.083/d$ ,  $D_3 = 0.1/d$  and  $D_4 = 0.125/d$  of the nutrients and fixed amounts of the  
15 trace metals (5 mL), the acclimated sludge was grown in 4 chemostats comprising  
16 semi-continuously stirred tank reactors of 1 L, each with a 500 mL working  
17 volume incubated at  $37 \pm 2^\circ C$ . The chemostat reactors were connected to four  
18 airtight pumps for the recirculation of  $H_2/CO_2$  (80:20, v/v). Each pump was

1 connected to a timer working at 1 min intervals; the flow rate of H<sub>2</sub>/CO<sub>2</sub> (80:20,  
2 v/v) gas was maintained at 0.83 mL/min at 0.08 MPa to maximize the mass  
3 transfer between the gas and the liquid (Coates et al., 1996). The liquid growth  
4 culture was taken from the reactors and replaced anaerobically with inorganic  
5 nutrients and trace metals at the dilution rates stated above. A new aluminum  
6 GL Sciences gas sampling bag (CCK) consisting of two mini valves containing  
7 pure H<sub>2</sub>/CO<sub>2</sub> gas was connected to the recirculation gas system at 24 h intervals to  
8 obtain a daily estimate of the CH<sub>4</sub> gas produced during the experiment until a the  
9 steady state fermentation was achieved.

#### 10 *2.4. Analytical methods*

11 The composition of the gas produced was determined using a GC-14B  
12 (Shimadzu) gas chromatograph equipped with a thermal conductivity detector  
13 connected to a C-R8A data analyzer. A high performance column packed with  
14 50/80 mesh Porapak Q was used. The temperatures of the injection, column and  
15 detector were set to 100, 50, 100°C, respectively. Argon was used as the carrier  
16 gas at a pressure of 0.5 MPa. The total solids (TS), volatile suspended solids  
17 (VSS) and volatile solids (VS) were determined with 5 duplicate analyses to  
18 ensure accuracy of the results obtained, according to the Standard Methods for

1 Examination of Water and Wastewater (APHA, 2005). The data were  
2 subsequently averaged; the deviations obtained were found to be less than 4%.  
3 The pH was monitored *in situ*. Cell number was estimated by a direct count using  
4 fluorescence microscopy (OLYMPUS, BX 50).

5

### 6 **3. Results and Discussion**

#### 7 *3.1. Steady state characteristics of chemostat H<sub>2</sub>/CO<sub>2</sub>-dependant methanogens*

8 Table 2 shows the measurements of the decrease in the gas concentration from the  
9 replaced aluminum gas sampling CCK bag. In the steady state, the percentage  
10 consumption rates of H<sub>2</sub> and CO<sub>2</sub> estimated individually, are approximately equal,  
11 and the equilibrium ratio did not change with variations in the dilution rates.  
12 The results suggest that the H<sub>2</sub>/CO<sub>2</sub> complex gas mixture can be considered as a  
13 single substrate during the determination of the kinetics parameters of these  
14 bacteria. The pH was not found to be a significant factor in the inhibition of  
15 bacterial activity within the continuously flowing reactor (Keshtkar et al., 2003)  
16 for any dilutions; this is due to the buffering effect of the media and the dissolved  
17 H<sub>2</sub>/CO<sub>2</sub> in the liquid. However, the inhibitory effect of pH on the physiological  
18 activities of hydrogenotrophic methanogens varies among different species



1 (Savant et al., 2002; Kotsyurbenko et al., 2004; Wolfe and Higgins, 1979).  
2 Costello et al. (1991) presumed that the Monod kinetics can be regarded as a  
3 mechanism for correction of the pH inhibition during the determination of the  
4 kinetic parameters. As presented in Fig.1, cell production under these conditions  
5 illustrates that the biomass concentration decreases with an increase of dilution,  
6 data similar to the results obtained by Haydock et al. (2004). The daily cells  
7 generated measurements were  $5.67 \times 10^{11}$ ,  $5.25 \times 10^{11}$ ,  $4.2 \times 10^{11}$  and  $2.1 \times 10^{11}$   
8 cells/L-culture for 0.073/d, 0.083/d, 0.1/d and 0.125/d, respectively. No bacteria  
9 carrier was used in the different reactors; thus, the dilution effect explains the data.  
10 The specific methanogenic activity (SMA) at different dilutions was determined  
11 according to the Ahring and Sorensen (1993) and the results are presented in Fig.  
12 2. The specific methane production rate was varied from 0.6 to 0.7 mmol/g cell  
13 hr, values lower than the 7.5 to 11.3 mmol/g cell hr determined by Wise et al.  
14 (1978) under the same conditions for a cell recycling system. In the steady state,  
15 the highest hydrogenotrophic methanogen activity was observed at a dilution of  
16 0.1/d, 0.24 L CH<sub>4</sub>/g VSS·d, due to the low quantities of trace metals available at  
17 low dilution rates. The methanogenic activity plot suggests a decoupling  
18 mechanism of anabolism and catabolism with regard to the metabolic activity of

1 hydrogenotrophic methanogens (Schönheit et al., 1980; Fardeau et al., 1987).

### 2 3.2. Kinetic evaluation in chemostat reactor using H<sub>2</sub>/CO<sub>2</sub> as substrate

3 In chemostat cultivation, the Monod equation is expressed as follows:

$$4 \quad \frac{1}{D} = \frac{K_s}{\mu_{\max}} \frac{1}{S} + \frac{1}{\mu_{\max}} \quad (2)$$

5 where  $D$  is the dilution rate (1/d).

6 Under steady state conditions, the bacterial growth yield ( $Y_{X/S}$ ) is expressed as

7 follows:

$$8 \quad Y_{X/S} = \frac{X}{S_0 - S}$$

9 (3)

10 where  $S_0$  and  $S$  represent the concentration of substrate (g/L) at the start of the

11 experiment and at time  $t$ , respectively.  $X$  represents the biomass concentration

12 (g/L).

13 Considering the complex gas mixture of H<sub>2</sub>/CO<sub>2</sub> as a single substrate,  $K_s$

14 and  $\mu_{\max}$  were determined from Eq. (2) (Lee, 1992). The experimental results

15 from the chemostat cultivation showed a maximum growth yield of 11.66 g cells

1 for each mmol of H<sub>2</sub>/CO<sub>2</sub> substrate consumption. This growth rate was found to  
2 be higher than the ranges of 0.0053~0.0095 g/mmol and 0.00279~0.0079 g/mmol  
3 that were obtained using a pure mixture of *Methanosarcina* sp. strain 227 and  
4 *Methanosarcina mazei* under H<sub>2</sub>/CO<sub>2</sub> in media containing minerals (Ferguson and  
5 Mah, 1983). The operational design of our experiment reveals a good productive  
6 fermentation system. The saturation constant  $K_S$  was 0.82 g/L, greater than the  
7 0.116 g/L that is necessary for methanogenic bacteria when they compete with  
8 sulphate-reducing bacteria for acetate (Bhattacharya et al., 1996) or other  
9 substrates (Gilardo-Gomez et al., 1992). With inorganic materials as substrate, it is  
10 suggested that a high concentration of H<sub>2</sub>/CO<sub>2</sub> is necessary to achieve the cells  
11 production. However, the maximum specific growth obtained with H<sub>2</sub>/CO<sub>2</sub> as a  
12 unique substrate was 0.15/d, and it is 10 times lower than the concentration 1.5/d  
13 that was reported for H<sub>2</sub> and CO<sub>2</sub> when acclimated mesophilic methanogen  
14 culture was grown using separate organic substrates for H<sub>2</sub> and CO<sub>2</sub> (Zhang and  
15 Maekawa, 1994). The flow rate of the gas mixture applied in the present study  
16 was 8.3 mL/min, a value substantially lower than that applied by Zhang &  
17 Maekawa (1994). The results presented here correspond well with those of

1 previous literature regarding the maximum specific growth rate as a function of  
2 gas flow rate (Schill et al., 1996).

### 3 3.3. Simulation

4 To simulate the behavior of the substrate ( $S$ ), the cell production rate  
5 ( $DX$ ) and the cell concentration ( $X$ ) in the chemostat reactor under steady state  
6 conditions, the following relations are used:

$$7 \quad S = \frac{K_S D}{\mu_{\max} - D} \quad (5)$$

$$8 \quad X = Y_{X/S} \left( S_0 - \frac{K_S D}{\mu_{\max} - D} \right) \quad (6)$$

$$9 \quad DX = Y_{X/S} D \left( S_0 - \frac{K_S D}{\mu_{\max} - D} \right) \quad (7)$$

10 The washout occurs when the cell concentration is equal to zero

$$11 \quad D_{crit} = \mu_{\max} \frac{S_0}{K_S + S_0} \quad (10)$$

12 The slow growth of the CO<sub>2</sub>-reducing methanogens normally involve a  
13 retention time greater than 10 d for efficient and stable operation (Zinder, 1993).

14 Fig. 3 summarizes the predictive plots for the hydrogenotrophic methanogen

1 growth and substrate supply at different dilution rates. The graphs indicate the  
2 maximum bacterial rate of growth at low dilution. This amount decreases as the  
3 dilution increases until it reaches  $D_m$ , the maximum dilution where the amount of  
4 bacteria in the reactor due to the dilution rate is in equilibrium with the limited  
5 supply of nutrients and trace metals. Subsequently, the cell washout begins to  
6 reach  $D_{crit}$  at the high dilution rate of 0.14/d; the predicted  $D_m$  was estimated as  
7 0.11/d. However, the cell production rate ( $DX$ ) increases with increasing dilution  
8 rate. The experimental data were in agreement with the predictive plots. The  
9 proposed operation system shows that the rates of dilution must be around 0.11/d  
10 to attempt a short retention time and a high substrate conversion rate at relatively  
11 low cell concentration in order to obviate the problems due to long retention time  
12 and toxicity effects of the trace elements. In the present study, the methanogenic  
13 bacteria attain an efficient activity level after a short retention time. This  
14 prediction was verified by the SMA previously determined (Fig. 2) where the  
15 highest methanogenic bacterial activity was 0.24 L CH<sub>4</sub> /g VSS·d, detected at  
16 0.1/d; this value was found to be close to the predicted value of  $D_m$ , which was  
17 0.11/d. The predicted critical dilution ( $D_{crit}$ ) value was 0.14/d. The CH<sub>4</sub>  
18 production effectively dropped when the dilution increased beyond  $D_{crit}$ . Table 3

1 presents a suitable range of parameters for chemostat cultivation with a mixture of  
2 H<sub>2</sub>/CO<sub>2</sub> (80:20, v/v) as substrate which can serve to design a reactor plant.

3

#### 4 **4. Conclusion**

5 Four anaerobic continuously stirred tank reactors containing cultures of  
6 acclimated hydrogenotrophic methanogen, obtained from a domestic anaerobic  
7 wastewater treatment plant, were run at different dilution rates in order to obtain  
8 engineering data necessary to simulate steady state chemostat cultivation with a  
9 supply of inorganic substrate. Under the conditions of steady state cultivation,  
10 the Monod model was applied to determine the kinetic parameters,  $\mu_{max}$  and  $K_S$ .  
11 The values found were 0.15/d and 0.82 g/L, respectively. From these results, a  
12 simulation was performed to predict the dilution rates at which bacteria washout  
13 began and at which maximum bacterial activity could be achieved for a  
14 concentration of 11.14 g/L of H<sub>2</sub>/CO<sub>2</sub>. The critical dilution rate for the  
15 chemostat was 0.14/d, and the maximum bacterial activity occurred at 0.11/d.  
16 The predictions agreed with the data obtained; in addition, the maximum methane  
17 gas production rate of 1.21 L/d was achieved at a dilution of 0.1/d. Thus, the  
18 engineering results can be summarized and applied to the large scale use of

1 H<sub>2</sub>/CO<sub>2</sub>-dependant methanogens for biogas production using only inorganic  
2 substrates. The results demonstrated that, at steady state, the concentration of  
3 cells was related more closely to the dilution rate than the H<sub>2</sub>/CO<sub>2</sub> concentration.

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