

1 Stable isotopic and geochemical characteristics of groundwater in Kherlen River basin, a
2 semi-arid region in eastern Mongolia

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16

17 Abstract Inorganic solute ion concentrations and stable isotopes of oxygen and hydrogen in
18 groundwater, river water and precipitation were investigated to gain insight into the
19 groundwater recharge process in the Kherlen River basin, a semi-arid region in eastern
20 Mongolia. The solute constituents in the river water (main stream) were of Ca-HCO₃ type,
21 spatially invariant and low in concentration. Groundwater in the upstream region was also
22 characterized by a Ca-HCO₃ type, though all ion concentrations were higher than in the river
23 water. On the other hand, the chemical composition of the groundwater in the midstream
24 region (southern and eastern) was spatially variable and the Na⁺, Mg²⁺, Cl⁻ and HCO₃⁻
25 concentrations were considerably higher than in the river water and upstream groundwater.
26 The stable isotopic compositions showed an evaporation effect on the groundwater and river

1 water, as well as an altitude effect in the precipitation and river water. Preferential recharge
2 by relatively large rainfall events is thought to have caused the depleted isotopic ratio in the
3 groundwater in the dry regions. The stable isotope, chemical and hydrological data suggest
4 that the main stream water of the Kherlen River is recharged by precipitation that falls in a
5 headwater region at an altitude of more than 1650 m, and that the interaction between the
6 groundwater and river water is not dominant in the midstream and downstream regions of
7 the river basin.

8 Keywords: Groundwater; Stable isotope; Semi-arid; Recharge; Mongolia

9

10 1. Introduction

11 In arid and semi-arid regions, domestic noncommercial, industrial and agricultural
12 water uses depend largely on the amount of groundwater (Wang and Cheng, 1999).
13 However, due to unsuitable management of water resources, several problems related to
14 groundwater use have arisen such as extraordinary decline of the groundwater level in the
15 High Plains, United States, the North China Plain, and the Delhi area, India (Brown and
16 Halweil, 1998; Datta, et al., 1996; Kondo et al., 2001; Rodell and Famiglietti, 2002; Tase,
17 2000). Sustainable control of water resources with sufficient understanding of the
18 groundwater situation is therefore essential. In Mongolia, a semi-arid region of northeast
19 Asia, more than 90 % of the total population use groundwater for daily necessities (Sugita,
20 2003). Moreover, no management of the water resources has been carried out since
21 destatization was established in 1990. Accordingly, it is highly probable that potential
22 problems with groundwater resources will occur. In order to develop an effective water use
23 system, it is first necessary to scientifically understand the behavior of groundwater.
24 However, very few studies have previously documented groundwater resources in northeast
25 Asia (Chelmicki, 1984).

26 The multi tracer approach using the isotopes and solute concentrations in water has

1 been used for elucidating the origin of groundwater and the interaction between groundwater
2 and river water in semi-arid and arid regions. Onodera (1996) suggested that preferential
3 and partial infiltration under conditions of high rainfall intensity is a major mechanism of
4 groundwater recharge in a tropical semi-arid region in Tanzania. In addition, Taniguchi et
5 al. (1995), based on stable isotope and solute concentration data, clarified that in the Heife
6 river basin, northwestern China, the groundwater originates from a mountain region.
7 Moreover, Kabeya et al. (2002) compared the stable isotopic compositions of the
8 groundwater between sand dunes and grassland in the Nu Us desert, China, and concluded
9 that the stable isotopes of the groundwater in the grassland were concentrated as a result of
10 evapotranspiration. Understanding the effect of evaporation on stable isotopes is an
11 important factor in discussing the relationship between precipitation and groundwater
12 (Boronina et al., 2005), and geochemical weathering is also known to affect the chemical
13 compositions of groundwater and river water (Sami, 1992). Previous studies suggest the
14 importance of temporal and spatial heterogeneity in infiltration, evaporation and
15 groundwater recharge processes in arid and semi-arid regions (De Vries and Simmers, 2002;
16 Gee and Hillel, 1988; Vogel and Van Urk, 1975). In other words, groundwater recharge is
17 affected by site-specific conditions in these regions.

18 In the Kherlen River basin, eastern Mongolia, the spatial distribution of vegetation
19 shows a clear change along the river. Mountain forest is distributed upstream in the basin
20 while grassland, known as steppe, without any tall trees can be seen downstream. In
21 addition, a discontinuous permafrost zone is also observed in the upstream region (Sharkhuu,
22 2001). Thus, there are many factors to be considered when discussing the hydrological
23 processes in this basin. Hirabaru et al. (1999) described the geochemical composition of
24 the groundwater in a central area of Mongolia and warned against a worsening of
25 groundwater quality, while Davaa et al. (2002) investigated $\delta^{18}\text{O}$ and δD in the Tuul River,
26 which flows into Ulaanbaatar, the capital city of Mongolia, and reported seasonal changes in

1 the groundwater recharge system. Few studies, however, have investigated the isotopic and
2 chemical characteristics of groundwater and discussed the recharge and flow system of
3 groundwater on a catchment scale including multiple vegetation cover. The purpose of this
4 study is to clarify the groundwater recharge and flow system in the Kherlen River basin,
5 eastern Mongolia, using a multi-tracer approach.

6

7 2. Study area

8 The study area was located approximately 120 km east of Ulaanbaatar, the capital city
9 of Mongolia (Fig. 1). The altitude of the main study area ranges from 1484 m at
10 Mongenmorit (MNG) to 985 m at Underhaan (UDH), and the total length of the main stream
11 of the Kherlen River is approximately 300 km from MNG to UDH. Mesozoic and
12 Paleozoic granite and Carboniferous granite are dominant from the mountainous upstream
13 area to Kherlenbayan-Ulaan (KBU) (Mineral Resources Authority of Mongolia, 1999).
14 From KBU to UDH, the right bank of the Kherlen River mainly consists of Mesozoic
15 sandstone, while Cenozoic sandstone or siltstone is distributed around the Darhan (DH)
16 region and along the river. Sharkhuu (2001) showed that the southern boundary between
17 the discontinuous permafrost region and no-permafrost region is found around Baganuur
18 (BGN). Conifer trees such as larch are dominant in the mountainous region, upstream of
19 MNG, whereas the flat plain downstream of BGN is dominantly covered by grass with a
20 height of 5 to 10 cm during the growing season.

21 As summarized by Sugita et al. (2006) (Table 1) based on data obtained by the
22 Institute of Meteorology and Hydrology (IMH), Mongolia, from 1993 to 2003 and aridity
23 index data (AI; Budyko, 1974), it is relatively humid in the upstream regions of the Forest
24 Site (FOR), MNG, BGN, and KBU, but relatively dry in the midstream region of DH and
25 UDH. The discharge rate of the Kherlen River observed by the IMH at BGN, UDH and
26 CHB from 1990 to 2000 is shown in Fig. 2. All gauging stations observe a high flow in the

1 summer, with a slight decrease in the discharge rate as the river flows down from BGN to
2 CHB, though the rate of decrease is small.

3

4 3. Methods

5 The field surveys and sampling of river water, groundwater and spring water were
6 performed in the area from MNG to UDH in June 2002, and July and October 2003 as a part
7 of the intensive field observation campaign of the RAISE project (Sugita et al., 2006). In
8 total, 30 river water samples, 7 spring water samples and 80 groundwater samples were
9 collected to analyze the major ions and stable isotopes of deuterium (D) and oxygen 18 (^{18}O).
10 In addition, monthly and daily precipitation were sampled separately for stable isotope
11 analysis at MNG, KBU, and UDH from October 2002 to September 2003. Most of the
12 groundwater samples were taken from wells with a depth of less than 10 m and set in an
13 unconfined aquifer. Electrical conductivity (EC; HORIBA Ltd., Twin Cond B173), pH
14 (HORIBA Ltd., Twin PH Meter B121), water temperature (Nikkyo Technos Co., Ltd., Petten
15 Kocher), and groundwater level were measured on site. The locations of the wells were
16 determined using a portable GPS meter (GARMIN Ltd., GPSMAP 76S). Each water
17 sample was sealed in a polyethylene bottle with a volume of 100 mL then brought back to
18 the laboratory in Japan where the anions, cations, and stable isotopic ratios of D and ^{18}O
19 were analyzed. Major anions (F^- , Cl^- , NO_2^- , Br^- , NO_3^- , PO_4^{3-} and SO_4^{2-}) were analyzed
20 using an ion chromatograph analyzer (Shimadzu, Co. Ltd., HIC-SP/VP Super).
21 Bicarbonate (HCO_3^-) concentration was determined by the titration method with sulfuric acid.
22 Major cations (K^+ , Na^+ , Ca^{2+} and Mg^{2+}) were analyzed using an inductively coupled plasma
23 atomic emission spectrometer (Nippon Jarrell-Ash Co., Ltd. Model ICAP-757). The stable
24 isotopes of D and ^{18}O were measured with a mass spectrometer (Finnigan Inc., MAT 252).
25 As pretreatment for stable isotopes analysis, water samples were equilibrated with CO_2 gas
26 for ^{18}O and H_2 gas with a platinum catalyst for D. The isotopic ratios of D and ^{18}O are

1 expressed by δD and $\delta^{18}O$, respectively, as follows:

2

$$3 \quad \delta_{sample} = \frac{R_{sample} - R_{SMOW}}{R_{SMOW}} \times 1000 \quad (\text{‰}) \quad (1)$$

4

5 where R is the ratio of D/H or $^{18}O/^{16}O$ in the sampled water (R_{sample}) or in Standard Mean
6 Ocean Water, SMOW (R_{SMOW}). The analytical errors were 0.1 ‰ for $\delta^{18}O$ and 1 ‰ for δD ,
7 respectively.

8

9 4. Results and Discussion

10 4-1. Stable isotopes

11 Fig. 3 presents the relationship between δD and $\delta^{18}O$ in precipitation sampled monthly
12 from October 2002 to September 2003 at MNG, KBU and UDH. The slope of the Local
13 Meteoric Water Line (LMWL) was similar to that of the Global Meteoric Water Line
14 (GMWL: $\delta D = 8\delta^{18}O + 10$):

15

$$16 \quad \delta D = 7.5\delta^{18}O + 2.1 \quad (2)$$

17

18 This shows that no evaporation occurred during precipitation. The monthly stable isotopic
19 ratio of the precipitation at the three locations ranged from -33.3 ‰ to -7.1 ‰ for $\delta^{18}O$ and
20 from -248.8 ‰ to -52.6 ‰ for δD during the observation period, and showed seasonal
21 variation with lower values in winter and higher values in summer (Yamanaka et al., 2006).
22 The variation range of δD and $\delta^{18}O$ in the groundwater was smaller than that of precipitation
23 (Fig. 4). The slope and interception (d-excess) of the regression line of the groundwater
24 data were lower than those of the LMWL:

25

1 $\delta D = 6.7\delta^{18}O - 11.2$ (3)

2

3 This shows that the groundwater is affected by evaporation. The stable isotopic
4 compositions of the groundwater and precipitation show that the shallow groundwater
5 mainly originates from precipitation that falls in the Kherlen River basin. The isotopic ratio
6 of the groundwater tended to be higher than that of the river water, and among the
7 groundwater samples, the δ values were the highest in the MNG and BGN (upstream)
8 regions. The δ values in the UDH region were lower than those in the MNG and BGN
9 regions, whereas the δ values in the DH and western region of DH (DH-W) showed wide
10 variation ranging from -14.3 ‰ to -10.7 ‰ for $\delta^{18}O$. The stable isotopes of the
11 groundwater in the MNG, BGN and UDH regions tended to be affected by evaporation more
12 than those in the other regions. However, this was thought to have been caused by
13 differences in the recharge process among regions; this will be discussed in more detail in
14 the following section.

15

16 4-2. Chemical composition

17 Fig. 5 shows the spatial distribution of the geochemical characteristics of the river water
18 and groundwater measured in July and October. No temporal change was found between
19 July and October in most of the sampling locations, although two wells at MNG and KBU
20 (W-MNG and W-KBU in Fig. 5) showed clear temporal changes. Because both wells were
21 newly bored in June and July 2003, the solute concentrations of the water in the boreholes
22 could have been affected by construction. Thus, the ion concentration data in these bore
23 holes were excluded from the following discussion.

24 The concentrations of major ions in the groundwater were notably higher than those in
25 the river water. The chemical composition of the main stream water was characterized as
26 Ca-HCO₃ type, as was the groundwater in the BGN and KBU regions, though the

1 concentrations of all ions in the groundwater were higher than in the river water. On the
 2 other hand, the groundwater in the surrounding region of DH (shown as well(W)36, W34,
 3 W70, W66 and W67 in Fig. 5) was characterized by a higher concentration of Na⁺ and
 4 HCO₃⁻, with some exceptions (W32 and W73). Stiff diagrams for the river and spring
 5 water are also shown in Fig. 6. The chemical patterns in the main stream water were
 6 uniform from MNG to UDH, though concentrations of the major ions increased slightly.
 7 The solute concentrations of the tributaries and springs were higher than those of the main
 8 stream water, though the chemical patterns were almost the same. A trilinear diagram is
 9 shown in Fig. 7 for the river water and in Fig. 8 for the groundwater. The solute
 10 constituents in the river (main and tributaries) and spring water were chemically unique and
 11 almost all the data fell into the category of Ca-HCO₃ type, whereas the chemical composition
 12 of the groundwater changed from place to place. The groundwater sampled in the BGN
 13 and the KBU regions was chemically similar to the river water from the main stream,
 14 whereas the chemical compositions of the groundwater in the region of DH, DH-W,
 15 Jargalthaan (JGH), and UDH are very different from those of the river water. Thus, the
 16 interaction between the groundwater and river water is possibly not dominant in the
 17 midstream and downstream regions of the Kherlen River basin.

18 The Na⁺ and Cl⁻ concentrations of the groundwater were considerably high in the
 19 DH-W region (W32, W36) as shown in Figs. 5 and 8. Fig. 9 shows the relationship
 20 between the Na⁺ and Cl⁻ concentrations of the groundwater. All plots distributed far from
 21 the sea water line and the line of 1:1 (Fig. 9), thus meteoric NaCl is not the source of Na⁺.
 22 Dissolution of sodium feldspars would increase Na⁺ relative to Cl⁻ by way of the following
 23 reaction (Wischusen, et al., 2004):

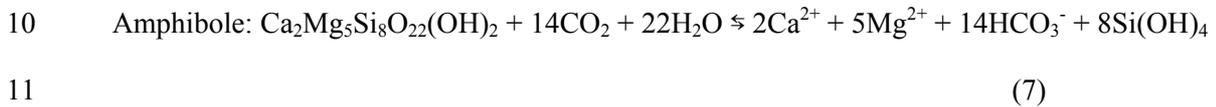
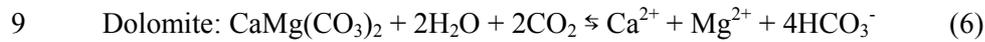
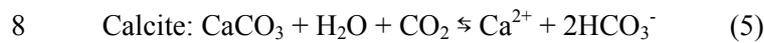
24



26

1 The ratio of Cl/Br varied throughout the Cl⁻ concentrations (Fig. 10), suggesting that
 2 evaporation alone does not cause the high Cl⁻ concentration of the groundwater in the DH-W
 3 region. In addition, Ca²⁺ was the dominant cation in the groundwater in the upstream region
 4 and the river water. The weathering of calcite, dolomite and amphibole generally accounts
 5 for the Ca²⁺ concentration in the river and groundwater and can be expressed by the
 6 following reactions:

7



12 The relationship between Ca²⁺ and HCO₃⁻ concentrations is shown in Fig. 11. The majority
 13 of plots distributed around the line of calcite, dolomite and amphibole weathering, thus
 14 suggesting that these weathering reactions mainly account for the Ca²⁺ concentration of the
 15 groundwater in the study area.

16

17 4-3. Origin of the Kherlen River water

18 Fig. 12 presents the relationship between δ¹⁸O and altitude in the river water, groundwater,
 19 spring water and volume weighted mean (annual and summer season) precipitation at MNG,
 20 KBU and UDH. The δ¹⁸O value of the river water and precipitation decreased with altitude
 21 based on the altitude effect (Siegenthaler and Oeschger, 1980; Ingraham, 1998; Pouge and
 22 Chamberlain, 2001), whereas the d-excess value increased with altitude (Fig. 13). Thus,
 23 the δ¹⁸O of the river water is determined by both the effect of distillation through evaporation
 24 from the river water surface and the altitude effect of precipitation. The majority of δ¹⁸O in
 25 the groundwater in the upstream and midstream regions was notably higher than that of the
 26 river water, though the quantity of groundwater samples was not sufficient in the upstream

1 region (Fig. 12). The chemistry of the river water was very different from that of the
2 groundwater especially in the midstream and lower stream regions (Fig. 5). Additionally,
3 the increase in discharge rate along the river (from BGN to CHB via UDH) was very small
4 as shown in Fig. 2. Thus the stable isotope, chemical and hydrological evidence all indicate
5 that groundwater inflow into the river is not dominant in the Kherlen River, and that the
6 main stream water originates from precipitation that falls in the headwater region, an area
7 with a higher altitude than MNG. The recharge altitude of the river water was evaluated by
8 extrapolating the regression line of the precipitation as shown in Fig. 12. The altitude effect
9 of the precipitation at MNG, KBU and UDH was estimated as 0.63 ‰/100m for $\delta^{18}\text{O}$, which
10 is a little higher than that reported previously (Siegenthaler and Oeschger, 1980; Pouge and
11 Chamberlain, 2001). Suppose the $\delta^{18}\text{O}$ value of -15.2 ‰ observed at the R14 site (Fig. 4)
12 represents the value of the river water in the headwater region, the recharge altitude of the
13 stream can be estimated as approximately 1650 m by extrapolating the regression line of the
14 precipitation. The highest altitude of the Kherlen River basin is approximately 2500 m.
15 Therefore, the origin of the main stream water is precipitation that falls in the headwater
16 region at an altitude of more than 1650 m, after which it flows downstream without major
17 interaction with the groundwater. The $\delta^{18}\text{O}$ of the groundwater, however, in MNG and the
18 spring water at an altitude of 1400 m was higher than that of the river water at the same
19 altitude. This does not support the estimated recharge altitude of the Kherlen River water.
20 There was no correlation between altitude and the $\delta^{18}\text{O}$ of the groundwater, and therefore we
21 could not estimate the $\delta^{18}\text{O}$ of the groundwater at an altitude higher than 1650 m. Thus,
22 how precipitation in regions higher than 1650 m recharges the river water cannot be
23 explained at present. Nevertheless, the presented data suggest that precipitation in the
24 headwater region is the origin of the Kherlen River water.

25
26 4-4. The recharge process of the groundwater

1 As shown in Fig. 4, the majority of stable isotopes in the groundwater in the upstream region
2 (MNG and BGN) were higher than those midstream (DH, DH-W), though the variation
3 range of the plots was large in the DH and DH-W regions. Fig. 14 shows the relationship
4 between $\delta^{18}\text{O}$ and the d-excess in the groundwater; $\delta^{18}\text{O}$ was negatively correlated with the
5 d-excess on a whole. The d-excess value is an index showing the evaporation effect on the
6 physicochemical characteristics of water: that is, if the water evaporates, the d-excess
7 decreases. The groundwater in BGN, MNG and UDH seemed to be affected by
8 evaporation more than that in DH and DH-W. The aridity index (AI) listed in Table 1
9 suggests that the groundwater in the DH and DH-W regions might be most affected by
10 evaporation. However, the question arises as to why the relatively dry DH region had a low
11 δ value and high d-excess. Onodera (1996) revealed that rainfall of less than 15 mm did not
12 contribute to groundwater recharge and that infiltrated soil water was evaporated soon after
13 the rainfall event in a semi-arid region in Tanzania, central Africa. Moreover, he also
14 showed that rainfall events with a volume of more than 20 mm and with a relatively lower
15 stable isotopic composition preferentially contributed to groundwater recharge. Thus, it is
16 probable that only large rainfall events with a lower stable isotope composition preferentially
17 recharge the groundwater in the dry DH and DH-W regions, but not in BGN, MNG and
18 UDH. Accordingly, as a result, the groundwater in DH and DH-W has a lower $\delta^{18}\text{O}$ value
19 and higher d-excess. One more interpretation is that heavy stable isotopes tend not to be
20 conserved in the soil if the infiltrated water is completely evaporated soon after the rainfall
21 event in the DH and DH-W regions. When the infiltrated subsurface water is completely
22 evaporated after a rainfall event isotopic evaporation signals do not remain in the soil
23 because deuterium and oxygen-18 constitute the water molecules themselves. Thus, under
24 highly dry conditions, the stable isotopes in the groundwater are depleted. In addition, the
25 temporal heterogeneity of the recharge is thought to have caused the large variation in the
26 stable isotopes in the groundwater in the DH and DH-W regions. On the other hand, in

1 MNG, BGN, and UDH, the subsurface water infiltrated during a rainfall event is partly
2 evaporated and condensed after the rainfall has ended, and therefore, the effect of
3 evaporation on the stable isotopic composition of the soil water remains. As a result, the
4 groundwater is probably recharged by soil water affected isotopically by evaporation, and
5 thus, the mean stable isotopic composition of the groundwater is higher than the annual
6 volume weighted mean precipitation seen in BGN. This process is also observed in warm
7 humid regions (Tsujimura and Tanaka, 1998). Consequently, because only relatively large
8 rainfall events can recharge the groundwater and the effect of the large amount of
9 evaporation does not remain in the subsurface water in the dry DH and DH-W regions, the
10 groundwater has a relatively low stable isotopic composition that varies largely and a high
11 d-excess value. In the relatively humid MNG and BGN regions, on the other hand, the
12 groundwater has a relatively higher isotopic value and lower d-excess.

13 Under warm humid conditions, the stable isotopic composition of the groundwater is
14 generally very constant both temporally and spatially compared to that of precipitation.
15 Moreover, the stable isotopic ratio of the groundwater corresponds to a value a little larger
16 than the annual volume weighted mean value of precipitation in that region (Tsujimura and
17 Tanaka, 1998). On the other hand, the heterogeneity of the groundwater recharge process
18 in arid and semi-arid regions was previously reported (Gee and Hillel, 1988). We adopted
19 stable isotopic and chemical tracers to investigate the groundwater behavior in a semi-arid
20 river basin in Mongolia where no such information was previously available, revealing
21 important isotopic and chemical characteristics of the groundwater recharge process.
22 Quantitative analysis is now necessary to determine the groundwater and river water
23 recharge rate in the study area.

24

25

26 5. Conclusions

1 The present study examined the stable isotopic composition and chemical
2 characteristics of groundwater, river water and precipitation in the Kherlen River basin, a
3 semi-arid region in eastern Mongolia. The chemical composition of the groundwater was
4 shown to differ from that of the main stream water. In addition, the solute constituents in
5 the river water were of a Ca-HCO₃ type, spatially invariant and low in concentration. The
6 groundwater in the upstream region was also characterized by a Ca-HCO₃ type, though all
7 ion concentrations were higher than in the river water. On the other hand, the chemical
8 composition of the groundwater in the southern and eastern regions was spatially variable
9 and Na⁺, Mg²⁺, Cl⁻ and HCO₃⁻ concentrations were considerably higher than in the river
10 water and upstream groundwater. The effect of evaporation and geochemical weathering
11 were thought to have caused the high concentration of these ions considering the ion ratios of
12 Na⁺/Cl⁻, Ca²⁺/HCO₃⁻.

13 The stable isotopic compositions showed an evaporation effect on the groundwater and
14 river water, as well as an altitude effect in the precipitation and river water. The δ¹⁸O and
15 δD values of the groundwater were relatively high and invariant in the MNG and BGN
16 regions, whereas in the DH and DH-W regions they were spatially variable and relatively
17 low. In the relatively humid MNG and BGN regions, the evaporation effect during the
18 recharge process caused groundwater with a high stable isotopic ratio and low d-excess. In
19 contrast, preferential recharge by relatively large rainfall events with a relatively low stable
20 isotopic composition likely resulted in the depleted isotopic ratio in the groundwater in the
21 relatively dry DH and DH-W regions. The high d-excess and low δ values in the
22 groundwater seem to be explained by the fact that isotopic evaporation signals do not remain
23 in the soil when the infiltrated subsurface water is completely evaporated after a rainfall
24 event. The stable isotope, chemical and hydrological data suggest that the main stream
25 water of the Kherlen River is recharged by precipitation that falls in a headwater region
26 located at an altitude of more than 1650 m, and that the interaction between the groundwater

1 and river water is not dominant in the midstream and downstream regions of the river basin.

2

3 Acknowledgements

4 The authors would like to thank Dr. D. Azzaya, Director of the Institute of Meteorology and
5 Hydrology, the National Agency for Meteorology, Hydrology and Environment Monitoring
6 of Mongolia for support in the field survey. Thanks are also due to Dr. N. Jadambaa,
7 Institute of Geocology, Mongolian Academy of Sciences, and Dr. N. Sharkhuu, Institute of
8 Geography, Mongolian Academy of Sciences, for providing important information on the
9 groundwater in Mongolia. This work was supported by a CREST project (The Rangelands
10 Atmosphere- Hydrosphere-Biosphere Interaction Study Experiment in Northeastern Asia) of
11 the Japan Science and Technology Agency. Partial support also came from the Global
12 Environment Research Fund of the Ministry of Environment, Japan, and the University of
13 Tsukuba Research Projects A.

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1 Figure legends

2 Fig. 1 Location of the study area. FOR: Forest site, MNG: Mongenmorit, BGN: Baganuur,
3 KBU: Kherlenbayan-Ulaan, DGH: Delgerhaan, JGH: Jargalthaan, DH: Darhan, UDH:
4 Underhaan, CHB: Choibalsan.

5 Fig. 2 The monthly discharge rate of the Kherlen River at BGN, UDH and CHB gauging
6 stations from 1990 to 2000 according to the IMH.

7 Fig. 3 The relationship between the $\delta^{18}\text{O}$ and δD of monthly precipitation and the annual
8 volume weighted mean precipitation (October 2002 to September 2003; the KBU data
9 was missing for April. The summer season denotes May to October and the winter
10 season November to April).

11 Fig. 4 $\delta^{18}\text{O}$ versus δD in the groundwater and annual volume weighted mean precipitation.
12 DH-W denotes the western region of DH. GW-others denotes the groundwater at KBU,
13 DGH, and JGH. River water denotes the data of the main stream of the Kherlen River.

14 Fig. 5 Spatial distribution of the chemical composition of the river water and groundwater.

15 Fig. 6 Spatial distribution of the chemical composition of the river water.

16 Fig. 7 Trilinear diagram of the river water samples obtained in July.

17 Fig. 8 Trilinear diagram of the groundwater samples.

18 Fig. 9 The Cl^- versus Na^+ concentration of the groundwater. The dotted line shows the
19 ocean water line. 'Others' denotes all data except for DH-W, DH and UDH.

20 Fig. 10 The Cl^- concentration versus Cl/Br ratio of the groundwater.

21 Fig. 11 The Ca^{2+} versus HCO_3^- concentration of the groundwater. The dotted line and
22 'others' are as in Fig. 9.

23 Fig. 12 The altitude versus $\delta^{18}\text{O}$ of the river water, groundwater, spring water and annual /
24 summer volume weighted mean precipitation. GW denotes the groundwater, and
25 'GW-DH' denotes the DH and DH-W regions combined. RRW: regression line of the
26 river water, RAP: regression line of the annual volume weighted mean precipitation,

1 RSP: regression line of the volume weighted mean precipitation in the summer season
2 (April – October).

3 Fig. 13 The spatial variation in $\delta^{18}\text{O}$ and the d-excess value along the Kherlen River starting
4 at the R14 site (see Fig. 6).

5 Fig. 14 The $\delta^{18}\text{O}$ versus d-excess value of the groundwater. ‘Others’ denotes the samples
6 obtained at KBU, DGH and JGH.

7

Table 1 Mean annual precipitation and aridity index (AI) (by Budyko, 197 in the Kherlen River basin. The abbreviations are listed in Fig. 1.

Site ID	Annual precipitation (mm/y)	Aridity Index
MNG	282	0.52
BGN	213	0.47
KBU	181	0.41
JGH	187	0.39
UDH	226	0.35
DH	216	0.35

The mean annual precipitation was obtained from 1993-2003 by IMH.
The potential evaporation used for estimation of AI was calculated with Penman method for 1988 (Sugita, 2003).

1

2

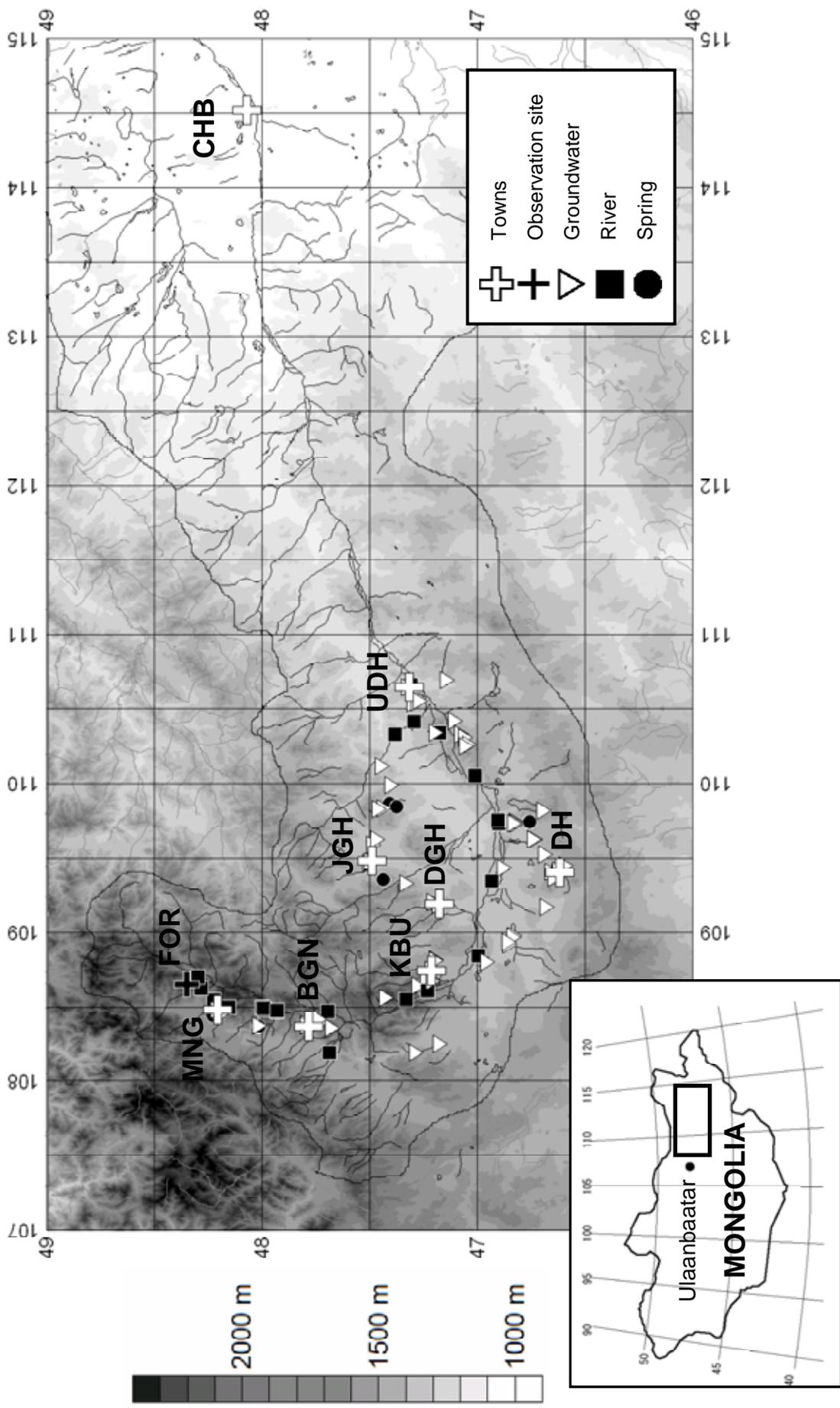


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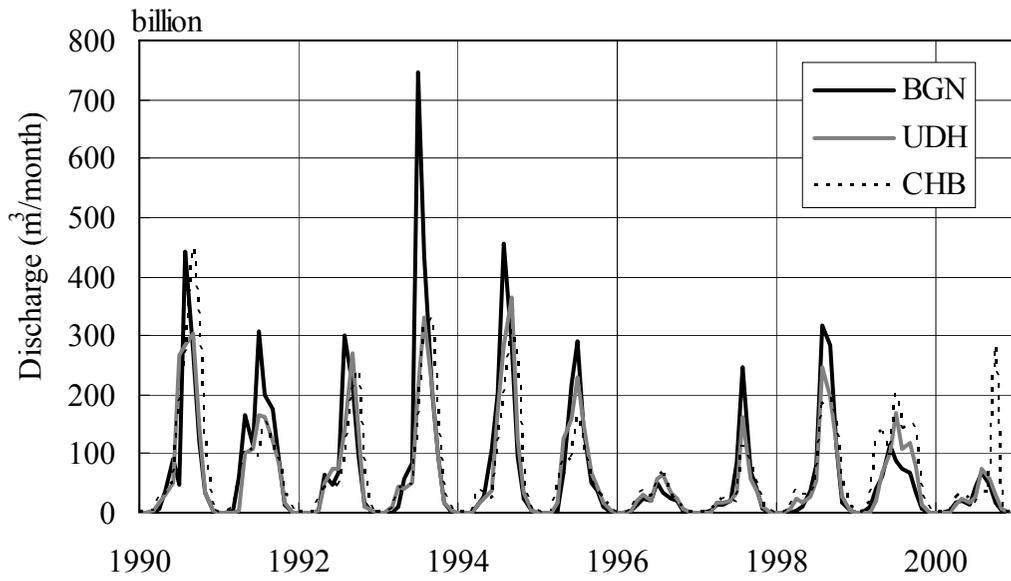


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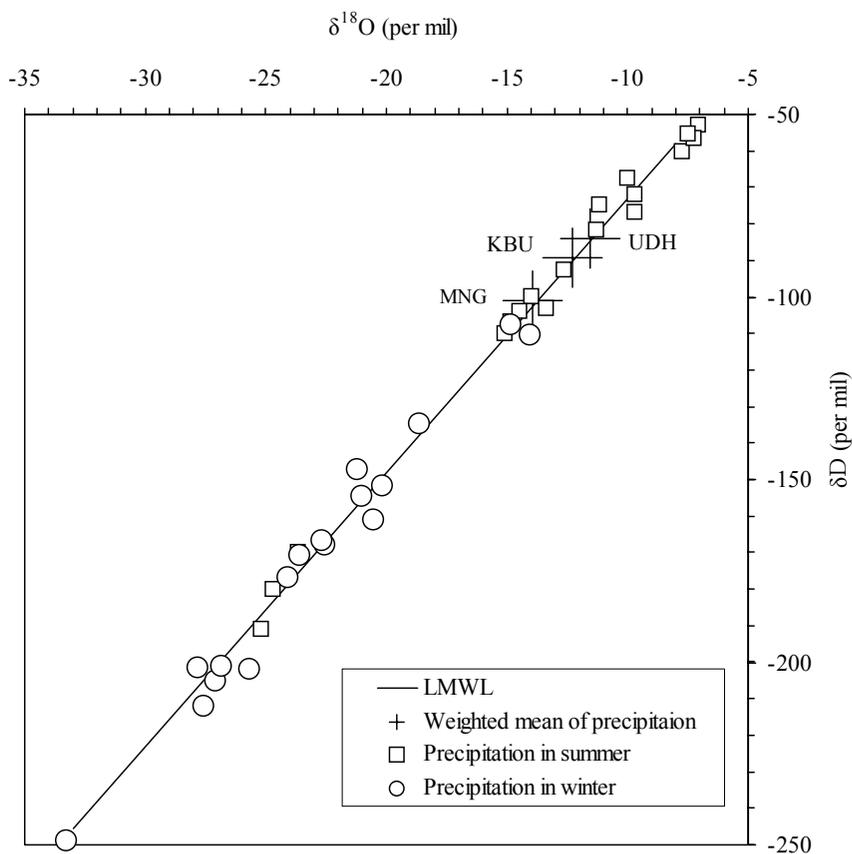


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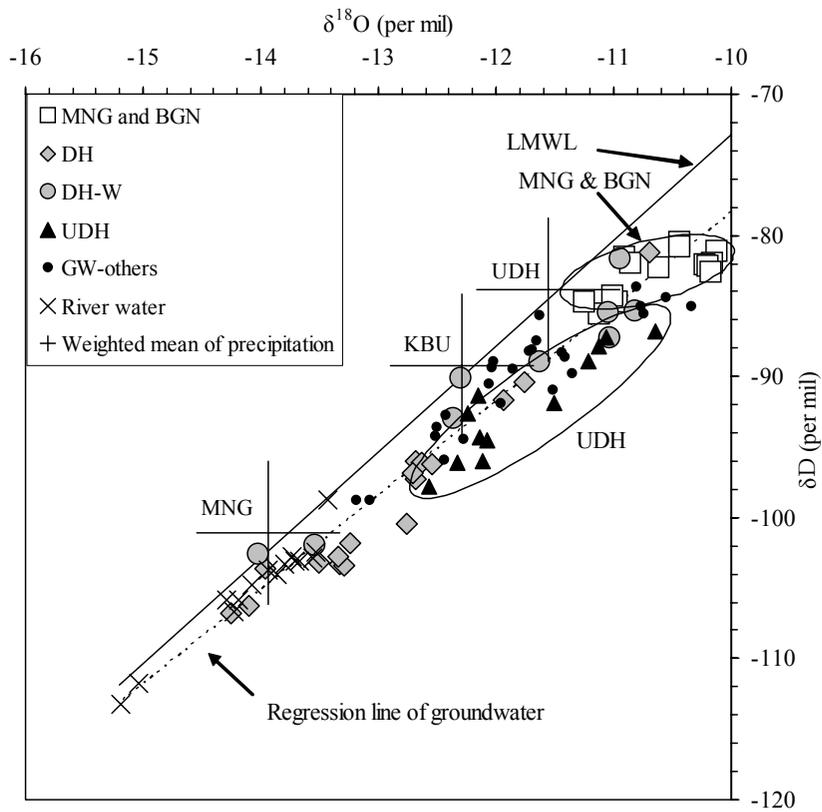


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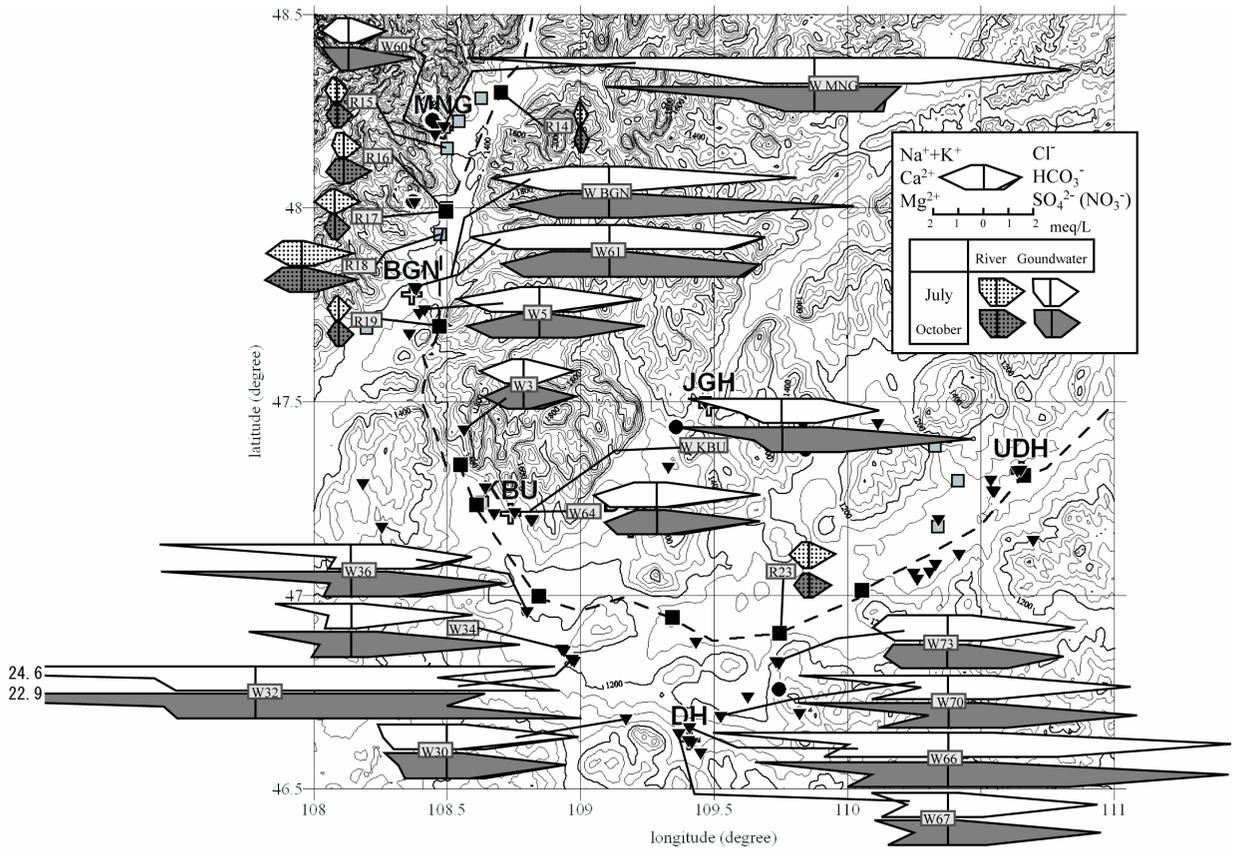


Fig. 5 Spatial distribution of the chemical composition of the river water and groundwater.

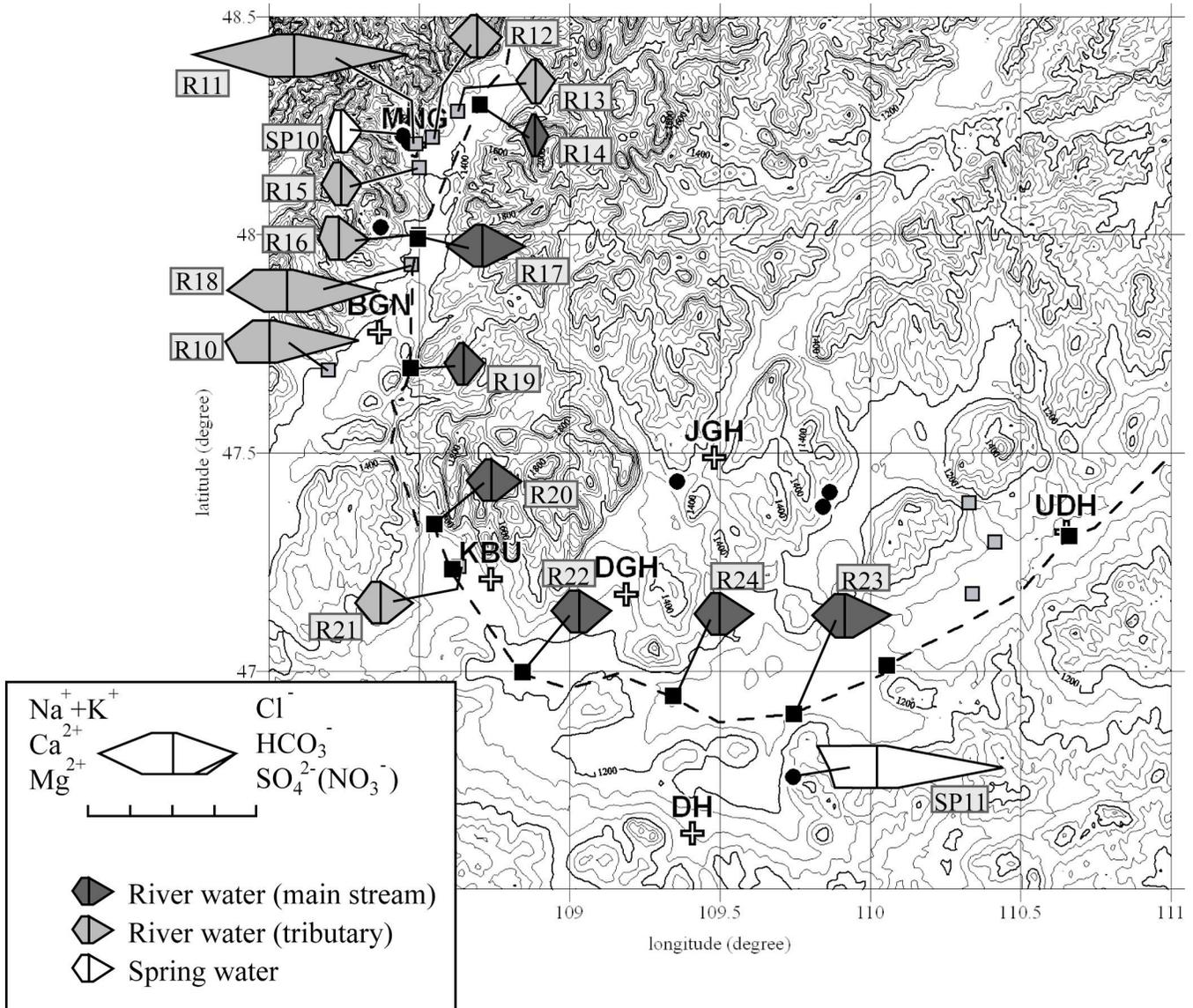


Fig. 6 Spatial distribution of the chemical composition of the river water.

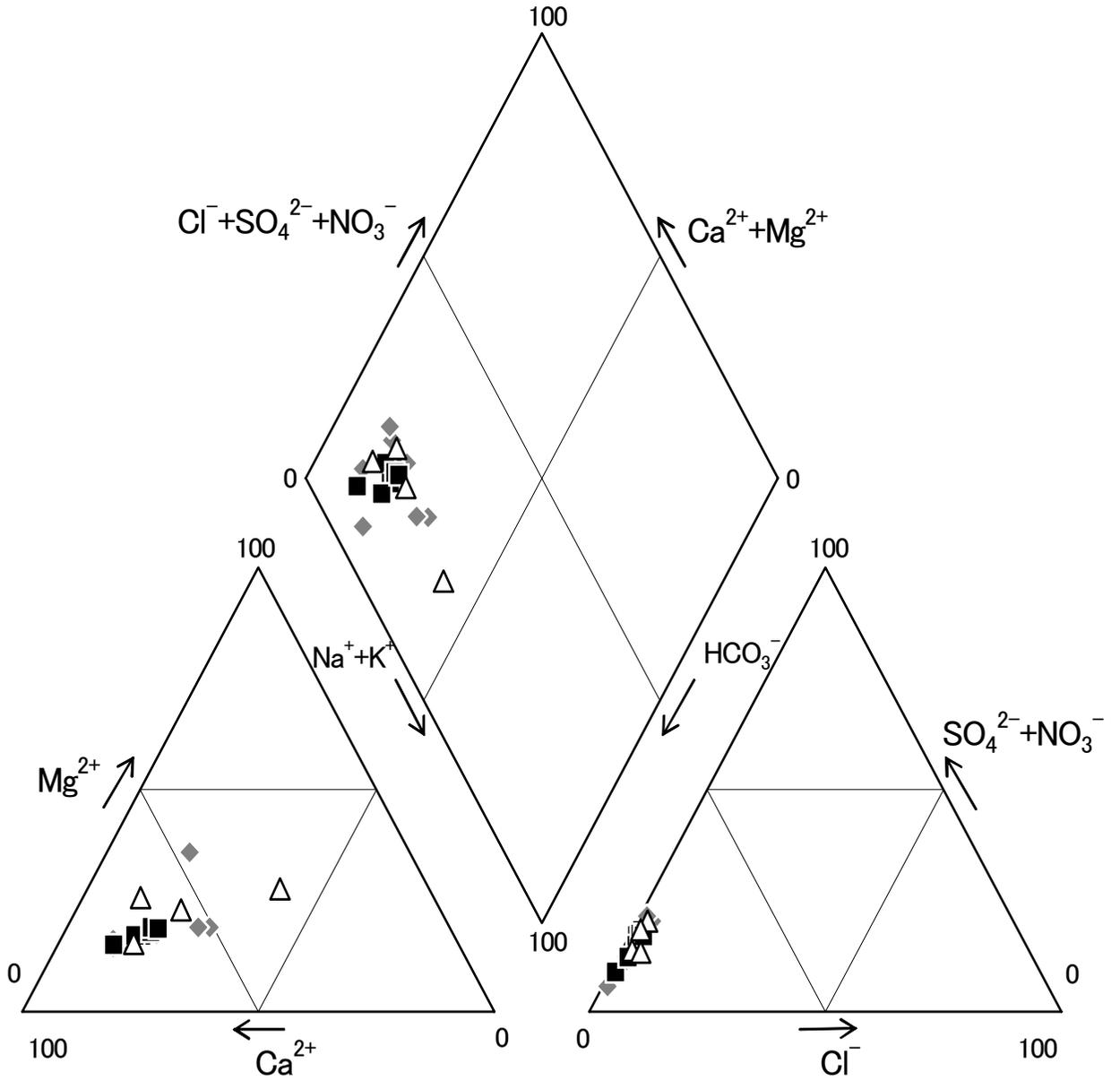
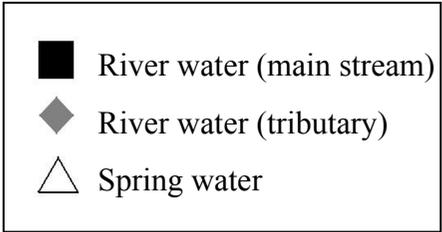


Fig. 7 Trilinear diagram of the river water samples obtained in July.

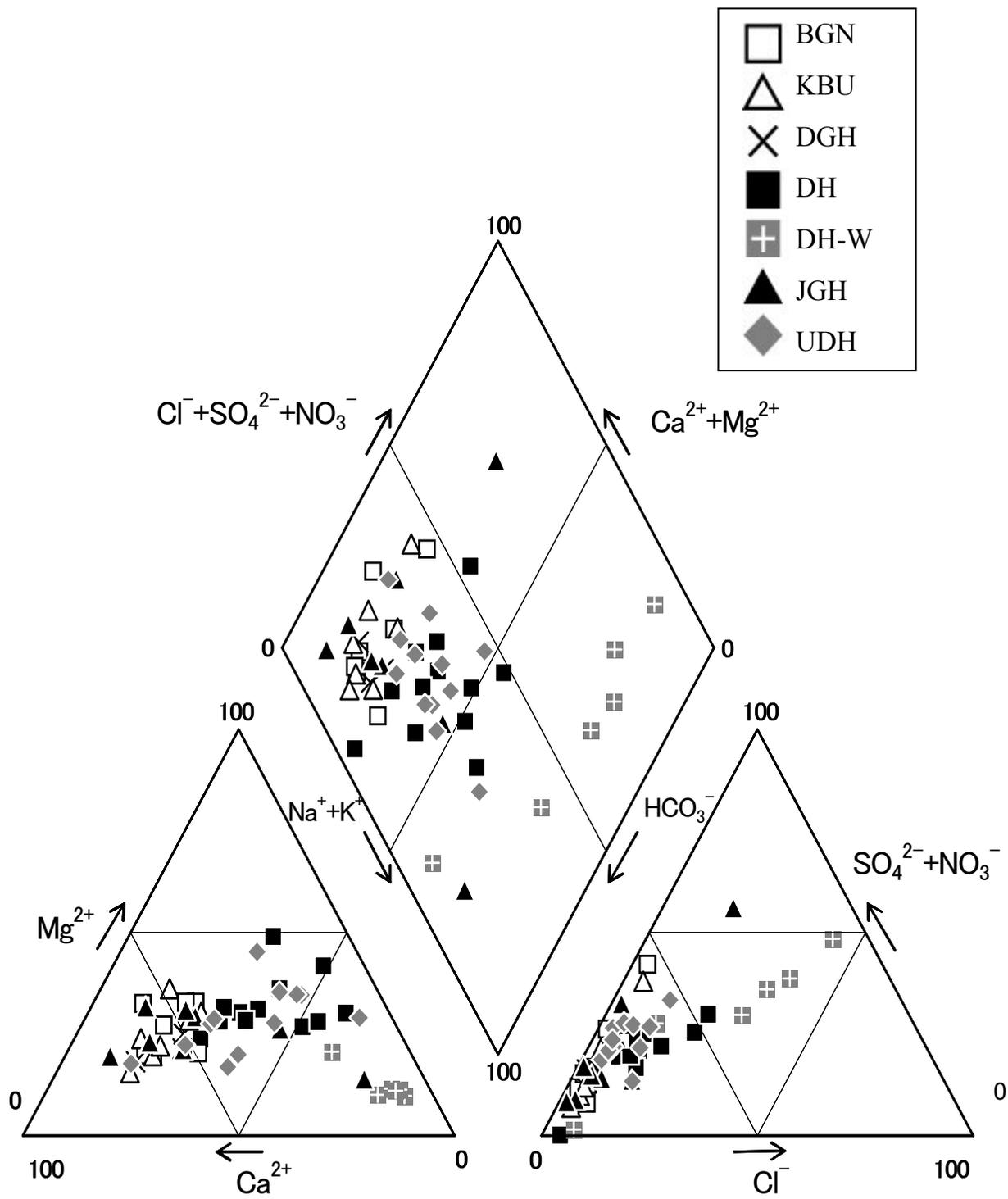


Fig. 8 Trilinear diagram of the groundwater samples.

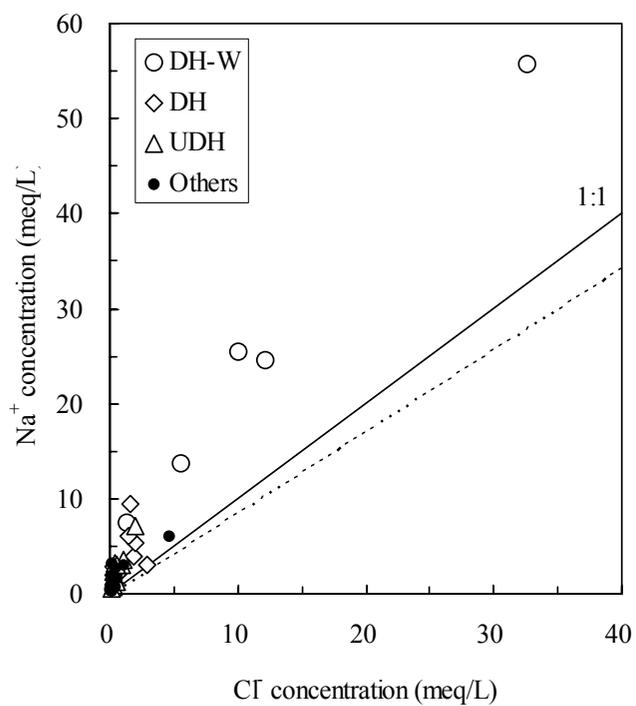


Fig. 9 The Cl⁻ versus Na⁺ concentration of the groundwater. The dotted line shows the ocean water line. 'Others' denotes all data except for DH-W, DH and UDH.

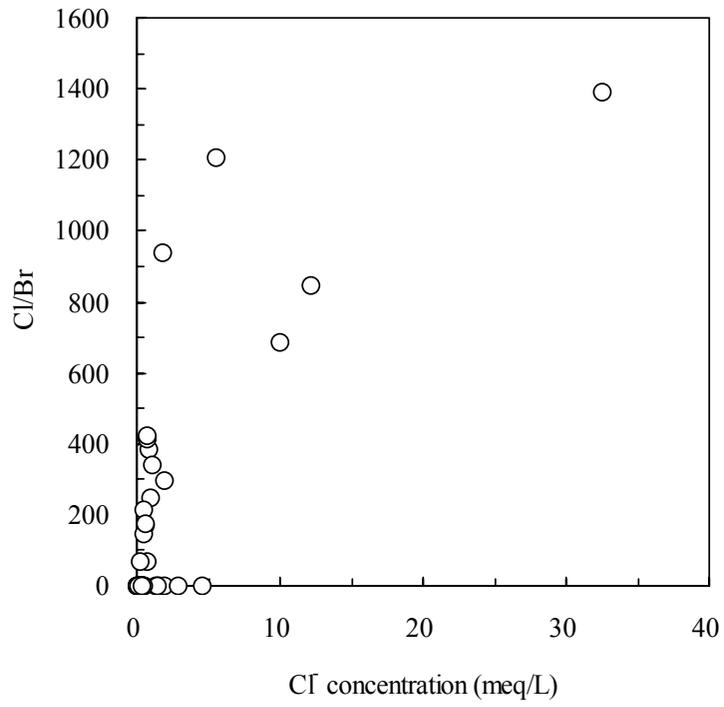


Fig. 10 Cl⁻ concentration versus Cl/Br ratio of the groundwater.

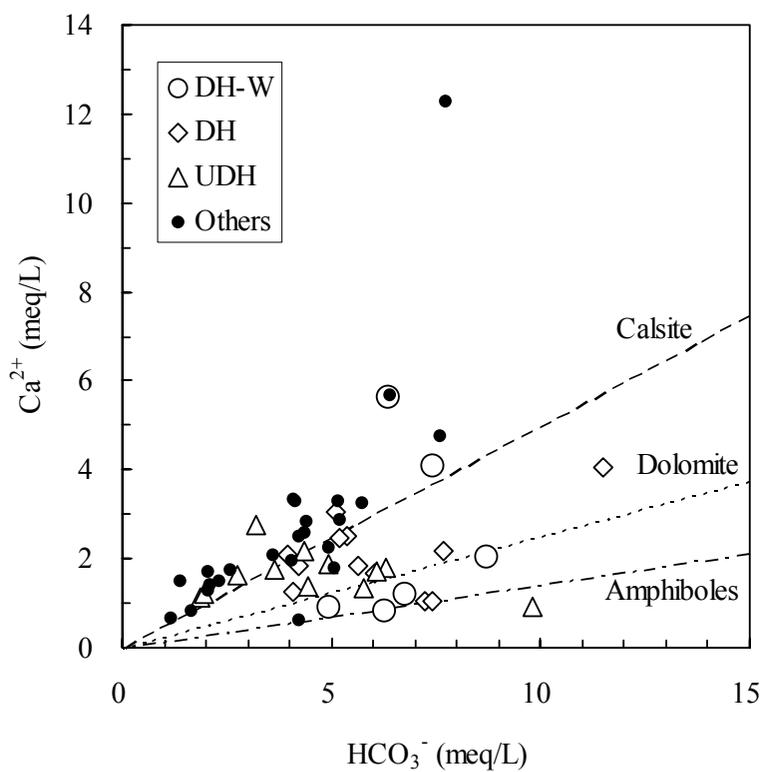


Fig. 11 Ca^{2+} versus HCO_3^- concentrations of the groundwater. The dotted line and 'others' are as in Fig. 9.

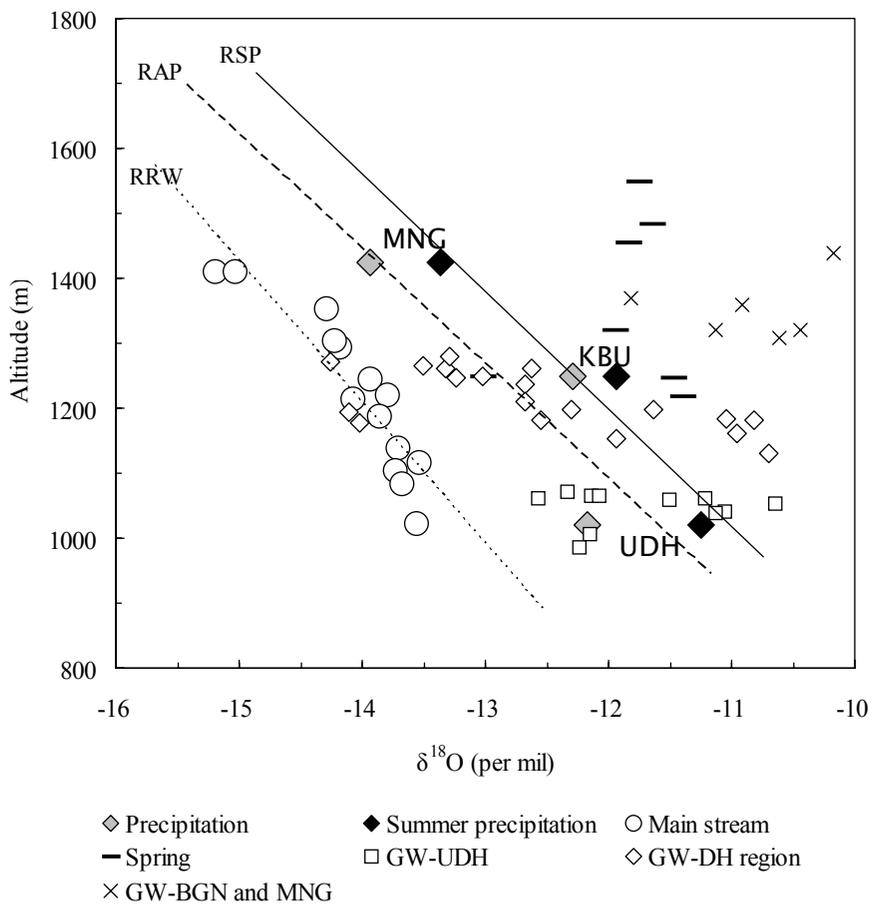


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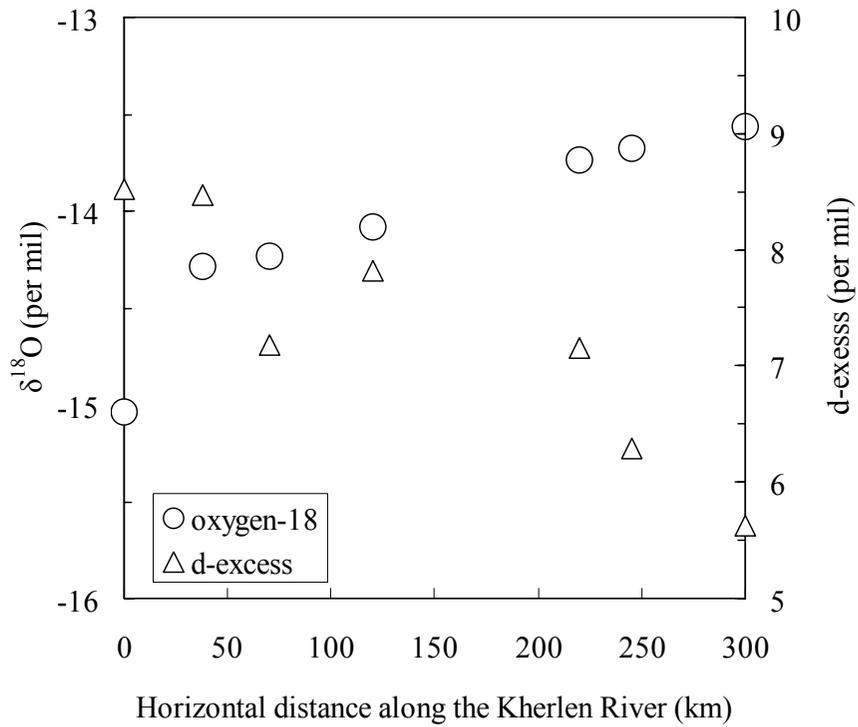


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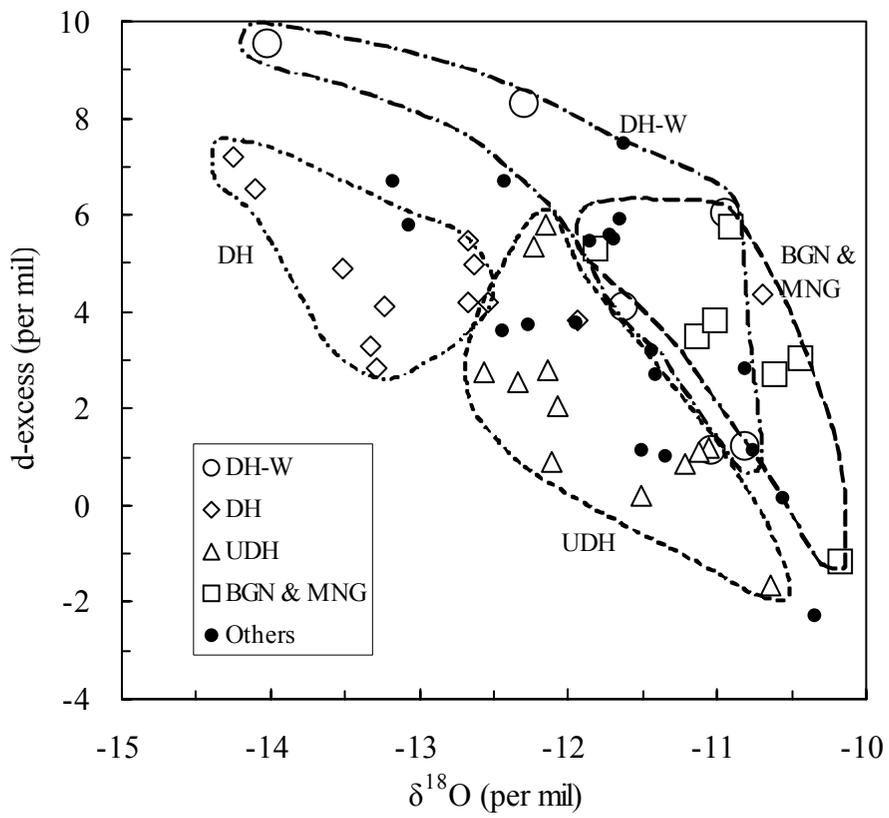


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